
 Cite this: *RSC Adv.*, 2018, 8, 7744

Electronic properties of atomically thin MoS₂ layers grown by physical vapour deposition: band structure and energy level alignment at layer/substrate interfaces†

 Fabio Bussolotti,^{ID}*^a Jainwei Chai,^a Ming Yang,^{ID}^a Hiroyo Kawai,^a Zheng Zhang,^{ID}^a Shijie Wang,^a Swee Liang Wong,^a Carlos Manzano,^a Yuli Huang,^a Dongzhi Chi^a and Kuan Eng Johnson Goh^{*ab}

We present an analysis of the electronic properties of an MoS₂ monolayer (ML) and bilayer (BL) as-grown on a highly ordered pyrolytic graphite (HOPG) substrate by physical vapour deposition (PVD), using lab-based angle-resolved photoemission spectroscopy (ARPES) supported by scanning tunnelling microscopy (STM) and X-ray photoelectron spectroscopy (XPS) for morphology and elemental assessments, respectively. Despite the presence of multiple domains (causing in-plane rotational disorder) and structural defects, electronic band dispersions were clearly observed, reflecting the high density of electronic states along the high symmetry directions of MoS₂ single crystal domains. In particular, the thickness dependent direct-to-indirect band gap transition previously reported only for MoS₂ layers obtained by exfoliation or *via* epitaxial growth processes, was found to be also accessible in our PVD grown MoS₂ samples. At the same time, electronic gap states were detected, and attributed mainly to structural defects in the 2D layers. Finally, we discuss and clarify the role of the electronic gap states and the interlayer coupling in controlling the energy level alignment at the MoS₂/substrate interface.

 Received 22nd January 2018
Accepted 12th February 2018

DOI: 10.1039/c8ra00635k

rsc.li/rsc-advances

1. Introduction

Transition-metal dichalcogenides (TMDCs; MX₂ where M = Mo or W and X = S, Se, or Te), are a wide class of layered semi-conducting materials with promising functionalities for optoelectronic applications.¹ The lattices of bulk TMDCs are formed by covalently bonded X–M–X hexagonal layers, which weakly bond with adjacent layers *via* van der Waals forces. At the monolayer limit (i) an indirect-to-direct band gap transition occurs due to the missing interlayer interaction^{2–6} which enhances the light absorption and the emission efficiency and (ii) a strong spin–orbit coupling combined with the broken inversion symmetry results in spin polarized bands,^{7,8} which makes TMDC monolayers suitable for spintronic applications.

High quality TMDCs layers with micrometer lateral size were first obtained from single crystals by a mechanical exfoliation technique, which is commonly used to isolate graphene layers.⁹

To meet the low cost wafer-scale fabrication requirements for industry adoption, large-scale deposition methods based on physical vapour deposition (PVD) and chemical vapour deposition (CVD) techniques were alternatively developed.^{10,11} A detailed characterization of the electronic properties of the as-grown TMDC layers at variance of the deposition conditions is critical for the device optimization, as they control the charge transport in the layer and at its interface with conductive electrodes. In this context, angle resolved photoemission spectroscopy (ARPES) studies have already proven to be useful for elucidating the electronic band structures of both exfoliated¹² and large scale growth TMDC single crystal layers.^{13–16} Synchrotron-based ARPES facilities can achieve excellent energy and spatial resolutions due high intensity and collimated photon sources ($\leq 100 \mu\text{m}$ of spot size^{17,18}) but *in situ* 2D TMDCs growth cannot be generally provided which is detrimental for a proper optimization of the deposition process. In contrast, a lab-based ARPES may be installed in proximity to growth systems, but they are commonly affected by lower energy/spatial resolution due to their limited light source intensity and relatively large photon spot size ($\sim 1 \text{ mm}$). In addition, structural defects (atomic vacancies within the lattice, grain boundaries, *etc.*) are generally introduced in the large scale growth process^{19,20} and can critically affect the electronic and optical properties of the layer. Despite their importance for

^aInstitute of Materials Research and Engineering, A*STAR (Agency for Science, Technology and Research), #08-03, 2 Fusionopolis Way, Innovis, Singapore 138634, Singapore. E-mail: b.fabio@imre.a-star.edu.sg

^bDepartment of Physics, National University of Singapore, 2 Science Drive 3, Singapore 117542, Singapore. E-mail: kejgoh@yahoo.com

† Electronic supplementary information (ESI) available: Contains additional ARPES data, STM data, Raman spectroscopy data and description of the layer structure. See DOI: 10.1039/c8ra00635k



applications, a detailed study of how the structural defects can impact on (i) the electronic band structures of as-grown TMDC layers and (ii) the charge injection at conductive interface is currently lacking.

In this paper, we report on the electronic properties of multi-domain MoS₂ monolayer (ML) and bilayer (BL) grown by PVD on highly ordered pyrolytic graphite (HOPG) substrate using a lab-based ARPES system. The coexistence between in-plane rotational disorder and measurable band dispersion were demonstrated. The complex interplay between the defect related electronic gap states, and the interlayer coupling in determining the energy level alignment at MoS₂/substrate interface were discussed.

2. Experimental

2.1 Sample preparation

HOPG substrates (HOPG, ZYA grade, SPI Supplies) were cleaved in air immediately before the introduction into the PVD growth chamber (pressure $\sim 10^{-6}$ mbar) and the surface was cleaned by *in situ* annealing at ~ 1000 K for 10 min. The MoS₂ layers (see schematic in Fig. S1 of ESI†) were deposited on the HOPG substrate by DC magnetron sputtering deposition technique from molybdenum (Mo) and sulphur (S) sources. The layer thickness was controlled by properly adjusting the total deposition time. Further details on the deposition technique were reported before.¹⁰ After deposition, the MoS₂ layers were annealed *in situ* at ~ 1000 K for 5 min to improve the layer crystallinity.

A high quality MoS₂ natural crystal was also used as a reference for ARPES measurements. The crystal was cleaved in air before introduction into the preparation chamber of the ARPES system (pressure $< 10^{-9}$ mbar).

2.2 Scanning tunnel microscopy (STM)

STM measurements were carried out in a custom-built multi-chamber ultra-high vacuum system housing an Omicron LT-STM, with a base pressure in the 10^{-10} mbar range. A chemically etched tungsten tip was used. The sample was degassed at 570 K for ~ 12 h before STM analysis. The sample was kept at 77 K during all the measurements to obtain better resolution and all STM images were recorded in constant current mode.

2.3 X-ray photoelectron spectroscopy (XPS)

XPS measurements were performed at room temperature (298 K) in VG-ESCA lab 220i-XL using a monochromatic Al-K _{α} source (photon energy $h\nu = 1486.7$ eV). The samples were introduced from air in the UHV system (pressure $< 10^{-9}$ mbar) and then measured as received. In the recorded kinetic energy range of Mo 3d and S 2p core levels (1100–1300 eV) a photoelectron mean free path of ~ 5 nm (ref. 21) can be estimated, which is much larger than the typical thickness of surface adsorbate layers resulting from the air exposure (~ 0.5 nm). In this context, the air exposure of the MoS₂ samples had a limited impact on the measured XPS data, only resulting in a small attenuation of the XPS signal from the MoS₂ layers. The binding energy scale was

referred to the lowest binding energy component (which is attributed to sp^2 C=C) in the C 1s spectra fixed at 284.5 eV, as measured in HOPG substrate, the energy resolution being set to 0.2 eV. XPS data were analysed by least square peak fitting procedure. Core level peaks were simulated by mixed Gaussian–Lorentzian functions. For the spin–orbit doublets, peak functions with same full width at half maximum (FWHM) were used, the spin orbit energy separation being set to 3.1 eV and 1.8 eV for Mo 3d and S 2p doublet, respectively.²² For each doublet, the relative intensity of the components (spin–orbit ratio) was fixed to 3 : 2 for Mo 3d and 2 : 1 for S 2p doublet.²²

2.4 ARPES

The ARPES measurements were conducted at room temperature (298 K) in a custom-designed ARPES system, with a hemispherical electron analyser (SCIENTA DA30L) and monochromatized HeI _{α} ($h\nu = 21.218$ eV) radiation source (SCIENTA VUV5k). More details on the experimental setup can be found in ref. 23. A schematic description of the experimental geometry is shown in Fig. S2 of ESI.† Due to the special design of the analyser lens, data acquisition is possible in (i) “normal” ARPES mode, where the emission angle (θ_x) is defined in the photoemission incidence plane and (ii) “deflection” ARPES mode, where full photoemission cone is accessible [*i.e.* both θ_x and θ_y are simultaneously measured (see Fig. S2 of ESI.†)] within a range of $\pm 15^\circ$ with respect to the surface normal direction. Higher angular limits, to reach the boundaries of the SBZ, were obtained by proper adjustment of sample surface orientation, defined by Θ_x and Θ_y angles, with respect to the analyser lens entrance axis z (Fig. S2 of ESI.†). More details on the experimental setup can be found in ref. 23. In both the ARPES acquisition modes the total energy resolution was set to 20 meV, the angular resolution being better than 0.2° . The binding energy scale was referred to the Fermi level (E_F) as measured for a clean gold substrate.

All the ARPES data were acquired at 297 K. Before ARPES measurements the samples, as introduced from air in the UHV system, were annealed *in situ* (pressure $< 10^{-9}$ mbar) at about 470 K for 12 h to remove surface adsorbates resulting from the air exposure.

2.5 Band structure calculations

First-principles band structure along the Γ KM and Γ M Γ direction of a MoS₂ ML and BL were calculated by using the HSE06 hybrid functional with spin–orbit coupling as implemented in the Vienna *ab initio* simulation package (VASP).^{24,25} In calculating the band structure of ML and BL the experimental in plane lattice constant of 3.16 Å (ref. 26) was used. For BL the interlayer distance was set to 6.15 Å.²⁶

3. Results and discussions

Fig. 1(a) and (b) show representative STM images of MoS₂ layers deposited on HOPG, as obtained for ML [panel (a)] and BL [panel (b)] nominal layer thickness (see Methods). Large scale STM images are reported in Fig. S3 of the ESI.† The MoS₂ ML



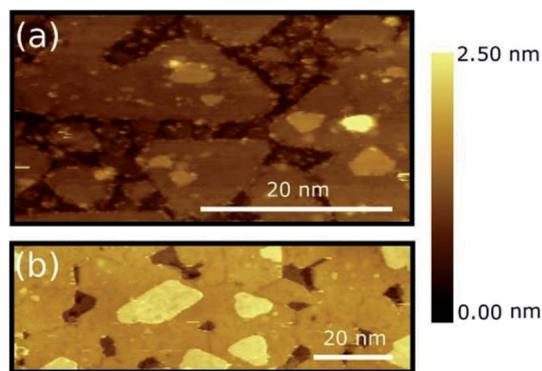


Fig. 1 (a) STM topographic images of the MoS₂ ML (V_{tip} : 1.0 V, I_{Tunnel} : 100 pA) (b) STM topographic images of the MoS₂ BL (V_{tip} : 1.0 V, I_{Tunnel} : 50 pA). STM images were acquired at 77 K of sample temperature. The colour scale indicates the relative height variation from HOPG (darker area) to top MoS₂ layers (brighter areas).

deposition results in an almost complete coverage of the HOPG substrate [darker area in panel (a) and (b)], the layer mainly originating by the coalescence of multiple MoS₂ grains and triangular islands with a typical lateral size < 50 nm. Smaller grains are also formed on top of some of the first layer islands clearly indicating that PVD deposited MoS₂ grows in an island growth mode rather than in a layer-by-layer mode on HOPG substrate.²⁷ A MoS₂ interlayer separation of 6 ± 1 Å was estimated by extracting line profile across various step regions (see Fig. S3(c) of ESI†), with an MoS₂–HOPG separation of 6 ± 1 Å, in good consistence with previous experimental reports.^{20,26} Small clusters of several nm in size in the flat areas between the grains are also clearly observed. Similar clusters were also found in other samples grown with comparable nominal thickness and can be tentatively identified with MoS₂ particles by-product of the growth process that could not coalesce or bind to existing islands and that likely act as the MoS₂ island precursors.

The sample with a MoS₂ BL seems to undergo a similar growth process [see Fig. 1(b)]. Grain boundaries at the grain joints can also be clearly distinguished. Similar to the case of the MoS₂ ML, the bilayer also shows triangular islands on top of the second layer indicating the early formation stages of a subsequent third layer. Additional Raman spectroscopy characterization of the MoS₂ ML and BL is provided in Fig. S3(d) of ESI.†

A similar multi-domain 2D structures were reported for MoS₂ layers deposited on various insulating amorphous substrates as SiO₂ (ref. 10 and 28) where, however, a larger grain size was generally observed (~20 μm of lateral size). The difference in domain sizes can be tentatively ascribed to (i) change in the growth conditions (substrate temperature, precursor, deposition techniques, etc.) and/or (ii) higher defect density on the HOPG substrate which can increase the number of possible of nucleation centres. Despite the difference in the grain size, the growth morphologies of PVD-grown MoS₂ layers on HOPG substrate resembles those of TMDC layers deposited on insulating amorphous substrates as commonly required for applications.

Fig. 2(a) shows the XPS survey spectra as acquired for MoS₂ ML and BL, with Mo 3d and S 2p core level binding energy regions highlighted in Fig. 2(b). The results of the corresponding peak fitting analysis (see Section 2.3) are also included.

No significant changes in the core level peak positions and widths were observed with the increase of the MoS₂ layer thickness. The peaks at 229.6 (163.6) and 237.7 (162.4) eV can be attributed to the Mo 3d_{5/2} (S 2p_{3/2}) and Mo 3d_{3/2} (S 2p_{1/2}) orbitals, respectively, in good agreement with previously reported binding energy values for the MoS₂ layers.^{13,29} The S 2s component at 226.7 eV is also clearly visible.^{10,29} Finally, in both MoS₂ ML and BL XPS data, two weak shoulders are visible at the low binding energy sides of the Mo3d main components, which can be reproduced by an additional Mo 3d spin-orbit doublet (Fig. 2). The binding energy positions of the doublet (228.4 for Mo 3d_{5/2}, 225.3 eV for Mo 3d_{3/2}) are consistent with the presence of under-coordinated Mo atoms of sub-stoichiometric MoS_{2-x}, due to S-vacancies in MoS₂ layers.³⁰ The relative amount of

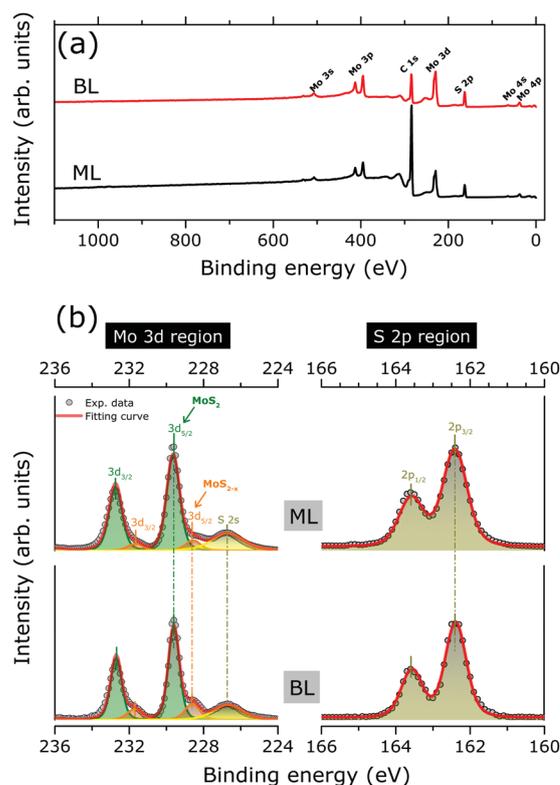


Fig. 2 (a) XPS survey spectra of the as grown MoS₂ ML and BL on HOPG (b) XPS data (grey full circles) of the as grown MoS₂ ML and BL as acquired in the Mo3d (left panels) and S 2p (right panels) binding energy regions. Clear peak doublet structures resulting from the spin orbit-splitting of the Mo3d (3d_{5/2} and 3d_{3/2}) and S 2p (2p_{3/2} and 2p_{1/2}) core levels are observed. The S 2s core level is also observed in the Mo3d binding energy region. Two weak shoulders at the high binding energy sides of the main Mo3d peaks are visible, as related to the presence of MoS_{2-x} phase. The peak positions, as evaluated by least square fitting of the XPS data (see text for more details), are marked by vertical bars. The Mo3d doublet of MoS₂ (MoS_{2-x}) phase is indicated by green (orange) curves/areas. The S 2s (S 2p) components are indicated by bright (dark) yellow curves/area. For each region, the cumulative fitting curve is also indicated (red curves).



under-coordinated Mo atoms with respect to the Mo atoms of the fully stoichiometric MoS₂, were extracted from the ratio of the corresponding 3d doublets' area. A defect concentration of ~10% and ~20% was found for the MoS₂ ML and MoS₂ BL, respectively.

Structural defects resulting from atomic vacancies were reported to introduce electronic states in the energy gap between the valence and conduction bands of semiconducting materials.³¹ By acting as electron (hole) donor/acceptor centres, gap states may affect the E_F position in the energy gap and result in a p-type (*i.e.* E_F closer to the valence band edge) or n-type doping (*i.e.* E_F closer to the conduction band edge) of the semiconducting materials. In S-defective MoS₂ layers an n-type doping was theoretically predicted³² with the S-vacancies introducing a high density of localized states close to conduction band edge.³³ Moreover, the S-vacancies formation can be also accompanied by lattice distortion/reconstruction around each defect sites.³⁴ Once laterally distributed in the 2D system, the S-related defects may alter, in principle, the lattice periodicity of the MoS₂ layers with a consequent impact on the electronic band dispersion. With this in mind, the band structures on the PVD grown MoS₂ ML and BL were carefully investigated by ARPES.

Representative ARPES constant energy maps (binding energy = 2.3 eV) in the 2D reciprocal space (k_x , k_y) are plotted in Fig. 3(a) (MoS₂ ML) and (b) (MoS₂ BL). The k_x and k_y represents the electron momentum component parallel to the substrate surface and measured in the xz (k_x) and yz (k_y) detector plane (see Fig. S2 of ESI†). The corresponding ARPES data of a MoS₂ single crystal are also included in Fig. 3(c) for comparison, the

ΓK high symmetry direction of the MoS₂ SBZ lying in the xz detector plane [see inset in panel (c)].

The ARPES constant energy maps of MoS₂ ML and BL [Fig. 3(a) and (b)] show circular intensity patterns around the Γ point of the SBZ ($k_x = 0.00 \text{ \AA}^{-1}$, $k_y = 0.00 \text{ \AA}^{-1}$). Similar circular-like patterns were obtained at different binding energy values and momentum ranges (data not shown). In MoS₂ single crystal, a clear hexagonal-like ARPES map was observed [Fig. 3(c)]. In the ARPES measurements of single crystal materials the momentum distribution of the photoemission intensity was reported to reflect the symmetry of SBZ,^{35,36} as it was consistently found for the present MoS₂ bulk sample [Fig. 3(c) and inset].

The circular-like ARPES maps observed for the PVD grown MoS₂ ML and BL samples [Fig. 3(a) and (b)] originate from their multi-domain structure, evidenced by the STM analysis (see Fig. 1 and S3 of ESI†). In particular, the circular-like intensity patterns may be viewed as the result the incoherent superposition of many hexagonal patterns from each single crystal domain, reflecting the symmetry of the corresponding SBZs and their relative random orientation in the reciprocal space [see inset in Fig. 3(a) and (b)]. Our data suggest that the MoS₂ ML and BL layer samples both consist of finite-size single crystalline domains which are much smaller than the analysis area (beam spot size ~ 800 μm) with a complete in-plane rotational disorder.

The random orientation of the MoS₂ ML and BL domains may partially originate from the rotational disorder of the HOPG substrate (lateral size of single crystal domain <100 μm (ref. 37)) which is transferred onto the 2D structure of the MoS₂ layers due to the epitaxial growth conditions.³⁸

Fig. 3(d) and (e) shows the ARPES intensity map of MoS₂ ML (d) and BL (e) as a function of the binding energy and momentum component k_x . For 2D layers with multi-domain structure, the band dispersion measured by ARPES along a given direction of the reciprocal space results from the superposition of the band dispersions of the various single crystal domains. In case of full in-plane rotational disorder, the band dispersion measured by ARPES is averaged over the entire SBZ of each single crystal domains which, in general, is expected to lead to no dispersion.

Despite the significant in-plane rotational disorder, giving the circular intensity map of Fig. 3(a) and (b), a clear band dispersion behaviour was observed in a wide energy/momentum range for both the ML and BL ARPES data [Fig. 3(d) and (e)]. The results qualitatively resemble the ARPES intensity map of MoS₂ bulk single crystal as acquired along the ΓK high symmetry direction [Fig. 3(f)].

A similar coexistence of in-plane rotational disorder with measured band dispersion by ARPES were reported for HOPG.³⁷ In particular, for any selected radial direction in the reciprocal space, ARPES mapping resulted from the superposition of the band dispersion of a single crystal graphite as measured along the high symmetry directions of the SBZ.³⁷ The ARPES results were explained by the high density of electronic states in the various single crystal grains of HOPG, as mainly localized along

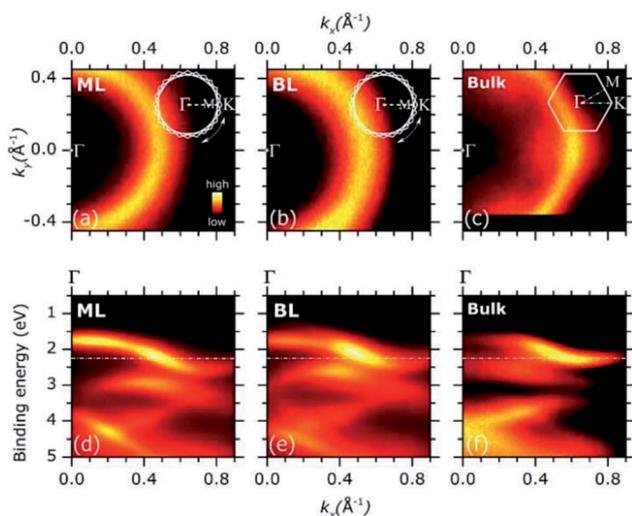


Fig. 3 (a)–(c): Representative ARPES intensity constant energy map [binding energy = 2.30 eV, dash dotted line in panel (d)–(f)] of MoS₂ ML (a), BL (b) and MoS₂ bulk single crystal (c) as in the 2D (k_x , k_y) reciprocal space. The impact of the in-plane rotational disorder on the intensity pattern of the constant energy maps of the MoS₂ ML and BL is schematically described in the insets with respect to the case of MoS₂ bulk single crystal (see main text for details). (d)–(f): ARPES intensity of MoS₂ ML (d), BL (e) as grown on HOPG substrates and MoS₂ bulk single crystal (f) as a function of the binding energy and momentum component k_x .



the high symmetry directions of the corresponding SBZ *i.e.* van Hove singularities in the electronic density of states (DOS).³⁷

Adopting a similar treatment in our analysis, the experimental ARPES band dispersion of MoS₂ ML and BL were compared with the calculated valence band structures along the ΓK and ΓM directions of the hexagonal SBZ. The results are shown in Fig. 4. The second derivative of the corresponding raw ARPES data intensity with respect to the energy ($-d^2I/d^2E$) is reported, in order to enhance the visibility of the experimental band dispersions.

In MoS₂ ML [left panel in Fig. 4(a)], a single band (VB_{ML}) is observed near the Γ point at ~ 1.75 eV. As momentum increases the band dispersion gradually separates into two broad bands, their turning points being located at 1.14 \AA^{-1} and 1.32 \AA^{-1} . By comparison with theoretical calculations [right panel of Fig. 4(a)], two (superimposed) band dispersions along the ΓM and ΓK directions can be clearly identified. The positions of the M (1.14 \AA^{-1}) and K (1.32 \AA^{-1}) points reflect the periodicity of the hexagonal SBZ in the reciprocal lattice [see Fig. S1 of the ESI†]. For the MoS₂ BL [left panel in Fig. 4(b)] the experimental data can be similarly described, with a comparable level of accuracy, in terms of superimposed ΓM and ΓK band dispersion [right panel in Fig. 4(b)].

As for the HOPG case, the comparison between the experimental and theoretical band dispersions suggests a strong localization of the electronic density of states along the high symmetry directions of the SBZ of the MoS₂ layers. This allows

the ARPES measurements of the electronic band dispersions even in presence of in-plane rotational disorder.³⁷ The results are further confirmed by the analysis ARPES spectra as acquired in a wider binding energy range, as shown in Fig. S4 of ESI.† More recently, the presence of van Hove singularities in the electronic density of states was theoretically suggested in order to explain the enhanced photo-absorption and hole–electron generation in 2D TMDCs layer.³⁹

The energy broadening observed close to the high symmetry points of the experimental valence band dispersions in Fig. 4 can be ascribed to an “averaging” effect in introduced by the in-plane rotational disorder (see Fig. S5 of ESI for more discussion†) as similarly reported for HOPG samples.³⁷ Because of the energy broadening, the small spin orbit splitting (~ 150 meV) at K point cannot be resolved in our experimental data. Despite the broadening, however, it is rather apparent that the position of the valence band edge on MoS₂ ML (*i.e.* at lowest binding energy) is located at K point (~ 1.60 eV) instead of Γ (~ 1.75 eV). This is highlighted in Fig. S6 of ESI† where a more detailed comparison of the MoS₂ ML experimental and theoretical band dispersion near Γ and K point is presented.

As for the MoS₂ BL, a remarkable result is that the valence band at K is located at higher binding energy with respect to the Γ point, where two band [VB_{BL}^1 , VB_{BL}^2 in left panel of Fig. 4(b)] are observed. The different intensities of VB_{BL}^1 and VB_{BL}^2 band in the second derivative plot in Fig. 4(b) reflect their relative intensities in the MoS₂ BL ARPES data [see Fig. S7 of ESI†]. Moreover, a decreasing of the overall ARPES signal at Γ point is observed in passing from ML to BL thickness [see comparison between Fig. 3(a) and (b) and Fig. S7 of ESI†], for almost completely vanishing in the bulk MoS₂ single crystal [Fig. 3(c)]. This behaviour was previously observed in ARPES measurements on exfoliated MoS₂ multilayer and related to matrix effect and multiple electron scattering during the photoemission process.¹² Since the electronic states at Γ point are mainly derived from the Mo d_{z^2} orbital in few-layer and bulk MoS₂,⁴⁰ the weak spectral intensity with respect to the ML case was explained as due to the slightly smaller in-plane lattice parameter in MoS₂ multilayer and bulk,⁷ which allows for greater shielding by the S 2p orbitals.⁴¹

As shown in Fig. 4, the valence band dispersions of MoS₂ ML and BL on HOPG along the ΓK direction are very well reproduced by the first principle calculation on MoS₂ isolated layers. In view of our results and quite consistently with previous observations reported on similar systems,¹⁵ no significant impact of the HOPG dielectric screening on the valence band structure on deposited MoS₂ layers appear to be present for our samples. More generally, any relevant substrate related effects are expected to affect differently the binding energy of the valence band states at Γ and K point, thus causing a valence band distortion with respect to the isolated layer case.¹⁶ The different impact of the substrate on the binding energy of Γ and K point reflects the difference in the spatial extension of the corresponding wave functions *e.g.* “ \sim out-of-plane” main orbital character (Mo3d_{*z*²}) at Γ vs. main “ \sim in-plane” character at K point (Mo3d_{*x*²-*y*²}, Mo3d_{*xy*}).⁴⁰ For the above reasons, the difference between theoretical and experimental bandwidth along

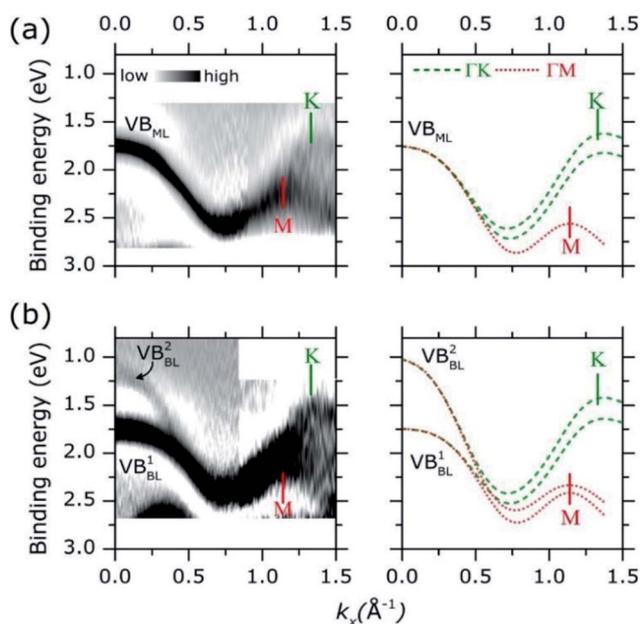


Fig. 4 (a) Experimental (left panel) and theoretical (right panel) band dispersion along ΓM and ΓK direction of MoS₂ ML on HOPG. Experimental band dispersion (darker area) were obtained by second derivative filter of corresponding ARPES data. Color intensity scale is indicated. Theoretical band dispersions along ΓK and ΓM direction are plotted as dashed green and dotted lines, respectively. Theoretical data were aligned at the binding energy position of the highest intense band at Γ point ($k_x = 0.00 \text{ \AA}^{-1}$). The position of K and M high symmetry point in the reciprocal space is indicated by vertical bar. (b) Same as in panel (a) for the BL.



the ΓM direction, where wave functions with out-of-plane orbital character also exist ($\text{Mo}3d_{z^2}$ at M) are unlikely due to substrate related effect, but may be attributed to other subtleties *e.g.* the details of DFT calculations. A significant distortion of the valence band along both the ΓK and ΓM direction would be observed otherwise. Further experimental and theoretical studies on MoS_2 thin film with larger lateral size are currently in progress with the aim to clarify this issue.

The clear difference between the experimental band dispersion of MoS_2 ML and BL (Fig. 4) supports the occurrence of a direct-to-indirect band transition with layer thickness, as seen in previous photoluminescence studies.^{2,3} This change in the electronic structure was ascribed to the interlayer interaction,^{2,3} which is responsible for the valence band splitting observed at the Γ point of the MoS_2 BL [Fig. 4(b)] and the consequent transition of the valence band edge from the K point [ML, direct gap, Fig. 4(a)] to Γ point (BL, indirect gap). Similar evolution of the band structure as a function of number of layers was only reported, with comparable energy resolution and data quality, in synchrotron based ARPES studies of MoS_2 single crystal layers as obtained by direct exfoliation¹² or by direct epitaxial growth on conductive single crystal.¹⁴ Moreover, the relatively high concentration of S vacancies in ML ($\sim 10\%$) and BL sample ($\sim 20\%$) does not seem to have a significant impact on the measured band dispersion, which remain very similar to those of a free layers and BL. In the defect-related structural disorder is expected to simply cause an energy broadening of the ARPES spectra as mediated by electron scattering effect during the photoemission process.³⁵ At the same time, a greater defect concentration near the domain boundaries can be suggested,²⁰ thus leaving relatively unaffected the electronic band structure of the single crystal domain "inner" regions. In this context, our investigation demonstrate the possibility of fundamental studies on large scale growth TMDCs layers, as generally produced for applications, and whose growth morphology and structural properties (in plane rotational disorder, defects) were commonly assumed to hinder important details of the electronic band structures so limiting the ARPES studies to high quality single crystal TMDCs layers.

Finally, the energy level alignment of ML and BL at the interface with HOPG is discussed, with the support of the results of the previous ARPES and XPS results. Fig. 5(a) shows the energy diagram in the MoS_2 ML and BL. The theoretical band dispersions along the ΓK directions were included, as they well reproduce (difference < 0.1 eV) the valence band edge positions at Γ and K point. The $\text{Mo} 3d_{5/2}$ and $\text{S} 2p_{3/2}$ core level position (as in Fig. 2) are also indicated. The energy position of the calculated conduction bands along the ΓK direction of the MoS_2 ML and BL were rigidly shifted to reproduce the experimental gap values, as extracted from scanning tunnel spectroscopy measurements on MoS_2 layers deposited on HOPG substrate.²⁰

In MoS_2 ML the E_F position in the (direct) energy gap lies at ~ 0.7 eV from the conduction band indicating that the MoS_2 layer is strongly n-doped. With respect to ML case the E_F position in the BL is located closer to the center of the (indirect)

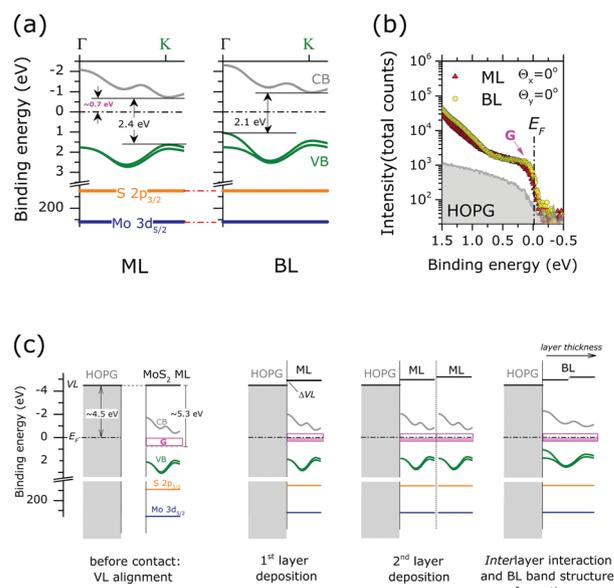


Fig. 5 (a) Energy level diagram for the MoS_2 ML (left) and BL (right). The calculated valence band (VB) and conduction band (CB) dispersion along ΓK direction are reported (see the figure). $\text{Mo} 3d_{5/2}$ and $\text{S} 2p_{3/2}$ binding energy positions of stoichiometric MoS_2 are also indicated. The energy positions of the calculated CBs of the MoS_2 ML and BL were rigidly shifted to reproduce the experimental gap value, as extracted from STS data of ref. 20 (b) angle integrated photoemission spectra ($-15^\circ \leq \theta_x \leq 15^\circ$, $-1^\circ \leq \theta_y \leq 1^\circ$) of MoS_2 ML, MoS_2 BL and HOPG bare substrate in the Fermi level (E_F) binding energy range as acquired at normal emission condition ($\theta_x = 0^\circ$, $\theta_y = 0^\circ$). The intensity is reported in log scale to highlight minor variation of the photoemission signal. A clear increase in the density of gap state at E_F (G) is observed upon MoS_2 layers growth. Note the tailing of the ML and BL peak contributing to the intensity in the energy gap (>0.75 eV from E_F) (c) schematic illustration of the energy level alignment mechanism at ML and BL interface with HOPG substrate. Shaded pink areas represent occupied gap states.

energy gap of the MoS_2 BL, while no change in the core level binding energy position and FWHM is observed.

n-type doping of MoS_2 ML was previously reported in ARPES studies of exfoliated¹² and CVD-grown ML.⁴² The ML doping is likely to be ascribed to the high density of S vacancies, as evidenced in our XPS analysis (see Fig. 2). In particular, S-vacancies in MoS_2 ML were theoretically predicted to introduce localized gap states at 0.6–0.7 eV from the conduction band edge.³³ This suggests that the E_F position in the energy gap of MoS_2 ML is pinned by the high density of gap states introduced by S vacancies in the layers. The existence of pinning gap states in the MoS_2 ML is confirmed by a detailed ARPES study in the E_F binding energy region [Fig. 5(b)]. After MoS_2 ML deposition on HOPG, a clear increase in the photoemission signal at the E_F is observed, which reflects the distribution of localized gap states (indicated with G in Fig. 5(b)) introduced by S vacancies in the 2D lattice structure. Interestingly, a comparable density of gap states at E_F is also detected in the MoS_2 BL. This is in apparent contradiction with the results on the XPS analysis assigning a higher S-vacancies concentration (20% vs. 10%) to the MoS_2 BL. This observation can be rationalized by considering the different surface sensitivity of the two photoemission based



techniques. In the E_F region, the measured kinetic energy (~ 17 eV) by ARPES corresponds to an electron mean free path of ~ 3 Å.²¹ This value is much smaller with respect to the interlayer distance in MoS₂ BL (6.15 Å). Because of the high surface sensitivity of the ARPES measurements, the photoemission signal at the E_F of MoS₂ BL [Fig. 5(b)] mainly reflects the defect densities and related n-doping level of the outer layer at the interface with the vacuum, which is comparable to that of the MoS₂ ML case. The excess of defect detected by XPS measurement (electron mean free path ~ 5 nm (ref.)) in the BL sample may be related to S-vacancies introduced in the inner MoS₂ layer during the top layer growth.

The above information contributes to clarify the mechanism of the energy level alignment at the interface with the HOPG in both MoS₂ ML and BL system, as schematically illustrated in Fig. 5(c). In the energy level diagrams of Fig. 5(c) a work function value of 4.5 eV and 5.3 eV was respectively assumed for the HOPG⁴³ and freestanding undoped MoS₂ ML.⁴⁴

Once the ML is in contact with HOPG substrate (*i.e.* 1st layer deposition) interfacial charge transfers occurs to establish thermodynamic equilibrium, *i.e.* E_F alignment at the interface. At thermodynamic equilibrium, the position of the E_F in the energy gap on of MoS₂ layer (0.7 eV from the conduction band edge) is entirely determined by the pinning gap states density (G) introduced by S-vacancies. This may result in vacuum level (VL) misalignment (ΔVL) and may be related to a dipole formation at the MoS₂ ML/HOPG interface [Fig. 5(c)].

When an additional freestanding ML with comparable doping level is put in contact with ML/HOPG system (2nd layer deposition) further charge transfer can occur at the interface between the MLs to reach thermodynamic equilibrium. An additional dipole can also result, reflecting the amount of charge transfer required to reach the Fermi level alignment across the layer. However, due to the pinning condition induced by the density of gap state, no change in the core level binding energies is expected upon film thickness increase, in agreement with our XPS measurements. Once thermodynamic equilibrium is established the interlayer interaction results in the final BL valence band structure, determining the observed position of the E_F in the indirect energy gap.

It is worth to note in passing that the substrate work function and MoS₂ gap state density in each layer only affects the amount of charge transfer required for establishing the thermodynamic equilibrium across the various MoS₂ layers. A lower density of gap state (*i.e.* no pinning condition) in each MoS₂ layer may result, for example, in a gradual shift of the core level position as a function of thickness, as reported in previous XPS study on MoS₂ multilayers.⁴⁵ Moreover, in case of different doping levels among adjacent MoS₂ layers, an energy offset is expected at the layer/layer interface. As the valence band edge position in each layer is different, this can also affect the interlayer interaction and the valence band structure of the multilayer. The above model suggests a possible strategy for controlling the electronic properties of TMDC multilayers. Further experimental and theoretical studies on this issue are currently in being pursued.

4. Conclusions

We have presented a detailed investigation of the electronic properties of atomically thin MoS₂ layers, deposited on HOPG substrate by physical vapor deposition technique. A multi-domain structure was first evidenced in MoS₂ ML and BL by detailed STM analysis. XPS investigation suggests, for both samples, a relatively high concentration of S-vacancies. Despite the randomly oriented multi-domain structure and defect concentrations, a clear band structure was extracted by ARPES. Also the coexistence of in-plane rotational disorder and a measurable band dispersion was demonstrated suggesting a high density of electronic states along high symmetry directions of each single crystal domain. In particular, our ARPES data provide sufficient energy resolution to demonstrate the expected direct-to-indirect band gap transition from ML to BL even in these somewhat imperfect grown layers. These results show that a lab-based ARPES system, such as ours, could readily provide meaningful fundamental investigations of the electronic band structure of large-area grown multi-domain 2D layers of TMDCs. This is a significant development given that the current challenge to reproducibly grow large-area 2D TMDCs is contingent upon the ability to readily characterize their properties so as to provide the needed frequent feedback for tuning the growth processes. The availability of a lab-based ARPES system (as demonstrated here) would greatly mitigate the longer lead-times to obtain measurements from significantly more costly synchrotron ARPES facilities. This would provide the much needed analysis for determining the layer electronic properties at site in a timely manner. As an example, the complex interplay between the defect related density of electronic states and interlayer interaction in determining the final position of the E_F in the energy gap of TMDCs films was discussed and clarified for our PVD grown large-area samples. These results suggest a strategy for tuning the interlayer interaction and consequently the band structure of TMDCs multilayer structure, which in turn can affect related optoelectronic applications.

Conflicts of interest

There are no conflicts of interest to declare.

Acknowledgements

The authors acknowledge financial support from Valleytronics Projects (IMRE/15-2R0706), Singapore A*STAR 2D PHAROS project: 2D devices & materials for ubiquitous electronic, sensor and optoelectronic applications (Project No: SERC 152-70-00012), Dr Jisheng Pan (IMRE) for designing the ARPES system, PREVAC Company (Poland) for providing and installing the equipment and A. L. C. Pang for experimental support in Raman spectroscopy measurements.

References

- 1 K. S. Novoselov, D. Jiang, F. Schedin, T. J. Booth, V. V. Khotkevich, S. V. Morozov and A. K. Geim, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 10451–10453.



- 2 A. Splendiani, L. Sun, Y. Zhang, T. Li, J. Kim, C. Y. Chim, G. Galli and F. Wang, *Nano Lett.*, 2010, **10**, 1271–1275.
- 3 K. F. Mak, C. Lee, J. Hone, J. Shan and T. F. Heinz, *Phys. Rev. Lett.*, 2010, **105**, 136805.
- 4 A. Kormányos, G. Burkard, M. Gmitra, J. Fabian, V. Zólyomi, N. D. Drummond and V. Fal'ko, *2D Mater.*, 2015, **2**, 022001.
- 5 H. J. Conley, B. Wang, J. I. Ziegler, R. F. Haglund, S. T. Pantelides and K. I. Bolotin, *Nano Lett.*, 2013, **13**, 3626–3630.
- 6 W. S. Yun, S. Han, S. C. Hong, I. G. Kim and J. Lee, *Phys. Rev. B*, 2012, **85**, 033005.
- 7 Z. Y. Zhu, Y. C. Cheng and U. Schwingenschlögl, *Phys. Rev. B*, 2011, **84**, 153402.
- 8 T. P. Kaloni, L. Kou, T. Frauenheim and U. Schwingenschlögl, *Appl. Phys. Lett.*, 2014, **105**, 233112.
- 9 B. Radisavljevic, A. Radenovic, J. Brivio, V. Giacometti and A. Kis, *Nat. Nanotechnol.*, 2011, **6**, 147–150.
- 10 J. Tao, J. Chai, X. Lu, L. M. Wong, T. I. Wong, J. Pan, Q. Xiong, D. Chi and S. Wang, *Nanoscale*, 2015, **7**, 2497–2503.
- 11 T. Niu and A. Li, *Prog. Surf. Sci.*, 2015, **90**, 21–45.
- 12 W. Jin, P. C. Yeh, N. Zaki, D. Zhang, J. T. Sadowski, A. Al-Mahboob, A. M. Van Der Zande, D. a. Chenet, J. I. Dadap, I. P. Herman, P. Sutter, J. Hone and R. M. Osgood, *Phys. Rev. Lett.*, 2013, **111**, 106801.
- 13 Y. Zhang, T.-R. Chang, B. Zhou, Y.-T. Cui, H. Yan, Z. Liu, F. Schmitt, J. Lee, R. Moore, Y. Chen, H. Lin, H.-T. Jeng, S.-K. Mo, Z. Hussain, A. Bansil and Z.-X. Shen, *Nat. Nanotechnol.*, 2014, **9**, 111–115.
- 14 J. a. Miwa, M. Dendzik, S. S. Grønberg, M. Bianchi, J. V. Lauritsen, P. Hofmann and S. Ulstrup, *ACS Nano*, 2015, **9**, 6502–6510.
- 15 M. M. Ugeda, A. J. Bradley, S. F. Shi, F. H. Da Jornada, Y. Zhang, D. Y. Qiu, W. Ruan, S. K. Mo, Z. Hussain, Z. X. Shen, F. Wang, S. G. Louie and M. F. Crommie, *Nat. Mater.*, 2014, **13**, 1091–1095.
- 16 J. A. Miwa, S. Ulstrup, S. G. Sørensen, M. Dendzik, A. G. Čabo, M. Bianchi, J. V. Lauritsen and P. Hofmann, *Phys. Rev. Lett.*, 2015, **114**, 1–5.
- 17 N. Alidoust, G. Bian, S.-Y. Xu, R. Sankar, M. Neupane, C. Liu, I. Belopolski, D.-X. Qu, J. D. Denlinger, F.-C. Chou and M. Z. Hasan, *Nat. Commun.*, 2013, **5**, 1–8.
- 18 I. Tanabe, M. Gomez, W. C. Coley, D. Le, E. M. Echeverria, G. Stecklein, V. Kandyba, S. K. Balijepalli, V. Klee, A. E. Nguyen, E. Preciado, I.-H. Lu, S. Bobek, D. Barroso, D. Martinez-Ta, A. Barinov, T. S. Rahman, P. A. Dowben, P. A. Crowell and L. Bartels, *Appl. Phys. Lett.*, 2016, 252103.
- 19 J. Hong, Z. Hu, M. Probert, K. Li, D. Lv, X. Yang, L. Gu, N. Mao, Q. Feng, L. Xie, J. Zhang, D. Wu, Z. Zhang, C. Jin, W. Ji, X. Zhang, J. Yuan and Z. Zhang, *Nat. Commun.*, 2015, **6**, 6293.
- 20 Y. L. Huang, Y. Chen, W. Zhang, S. Y. Quek, C.-H. Chen, L.-J. Li, W.-T. Hsu, W.-H. Chang, Y. J. Zheng, W. Chen and A. T. S. Wee, *Nat. Commun.*, 2015, **6**, 6298.
- 21 M. P. Seah and W. A. Dench, *Surf. Interface Anal.*, 1979, **1**, 2–11.
- 22 J. F. Moulder, W. F. Stickle, P. E. Sobol and K. D. Bomben, *Handbook of X-ray Photoelectron Spectroscopy*, 1992, vol. 3.
- 23 F. Bussolotti, Z. Zhang, H. Kawai and K. E. J. Goh, *MRS Adv.*, 2017, **2**, 1527–1532.
- 24 G. Kresse and J. Hafner, *Phys. Rev. B*, 1993, **47**, 558–561.
- 25 J. Heyd, G. E. Scuseria and M. Ernzerhof, *J. Chem. Phys.*, 2003, **118**, 8207–8215.
- 26 A. A. Al-Hilli and B. L. Evans, *J. Cryst. Growth*, 1972, **15**, 93–101.
- 27 J. A. Venables, G. D. T. Spiller and M. Hanbucken, *Rep. Prog. Phys.*, 1984, **47**, 399–459.
- 28 H. Zhang, Y. Ma, Y. Wan, X. Rong, Z. Xie, W. Wang and L. Dai, *Sci. Rep.*, 2015, **5**, 8440.
- 29 G. Eda, H. Yamaguchi, D. Voiry, T. Fujita, M. Chen and M. Chhowalla, *Nano Lett.*, 2011, **11**, 5111–5116.
- 30 T. Daeneke, R. M. Clark, B. J. Carey, J. Z. Ou, B. Weber, M. S. Fuhrer, M. Bhaskaran and K. Kalantar-zadeh, *Nanoscale*, 2016, **8**, 15252–15261.
- 31 M. T. Greiner, L. Chai, M. G. Helander, W. M. Tang and Z. H. Lu, *Adv. Funct. Mater.*, 2012, **22**, 4557–4568.
- 32 L. P. Feng, J. Su and Z. T. Liu, *RSC Adv.*, 2015, **5**, 20538–20544.
- 33 W. Zhou, X. Zou, S. Najmaei, Z. Liu, Y. Shi, J. Kong, J. Lou, P. M. Ajayan, B. I. Yakobson and J. C. Idrobo, *Nano Lett.*, 2013, **13**, 2615–2622.
- 34 Y. Gan and H. Zhao, *Phys. Lett. A*, 2014, **378**, 2910–2914.
- 35 A. Damascelli, Z. Hussain and Z.-X. Shen, *Rev. Mod. Phys.*, 2003, **75**, 473–541.
- 36 H. Coy-Diaz, F. Bertran, C. Chen, J. Avila, J. Rault, P. Le Favre, M. C. Asensio and M. Batzill, *Phys. Status Solidi RRL*, 2015, **9**, 701–706.
- 37 S. Y. Zhou, G. H. Gweon, C. D. Spataru, J. Graf, D. H. Lee, S. G. Louie and A. Lanzara, *Phys. Rev. B*, 2005, **71**, 2–5.
- 38 A. A. Koós, P. Vancsó, G. Z. Magda, Z. Osváth, K. Kertész, G. Dobrik, C. Hwang, L. Tapasztó and L. P. Biró, *Carbon N. Y.*, 2016, **105**, 408–415.
- 39 L. Britnell, R. M. Ribeiro, A. Eckmann, R. Jalil, B. D. Belle, A. Mishchenko, Y.-J. Kim, R. V. Gorbachev, T. Georgiou, S. V. Morozov, A. N. Grigorenko, A. K. Geim, C. Casiraghi, A. H. C. Neto and K. S. Novoselov, *Science*, 2013, **340**, 1311–1314.
- 40 E. Cappelluti, R. Roldán, J. A. Silva-Guillén, P. Ordejón and F. Guinea, *Phys. Rev. B*, 2013, **88**, 075409.
- 41 T. Boeker, R. Severin, a. Mueller, C. Janowitz, R. Manzke, D. Voss, P. Krueger, a. Mazur and J. Pollmann, *Phys. Rev. Lett.*, 2001, **64**, 235305.
- 42 D. J. Trainer, A. V. Putilov, C. Di Giorgio, T. Saari, B. Wang, M. Wolak, R. U. Chandrasena, C. Lane, T.-R. Chang, H.-T. Jeng, H. Lin, F. Kronast, A. X. Gray, X. X. Xi, J. Nieminen, A. Bansil and M. Iavarone, *Sci. Rep.*, 2017, **7**, 40559.



- 43 H. Fukagawa, H. Yamane, T. Kataoka, S. Kera, M. Nakamura, K. Kudo and N. Ueno, *Phys. Rev. B*, 2006, **73**, 24–26.
- 44 H. Zhong, R. Quhe, Y. Wang, Z. Ni, M. Ye, Z. Song, Y. Pan, J. Yang, L. Yang, M. Lei, J. Shi and J. Lu, *Sci. Rep.*, 2016, **6**, 21786.
- 45 Y.-K. Lin, R.-S. Chen, T.-C. Chou, Y.-H. Lee, Y.-F. Chen, K.-H. Chen and L.-C. Chen, *ACS Appl. Mater. Interfaces*, 2016, **8**, 22637–22646.

