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Research paper

Multifunctional energy storage and photoluminescence of Er-modified KNN-based transparent ferroelectric ceramics

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ABSTRACT

Against the backdrop of increasing miniaturization and integration of electronic components, the demand for materials with multifunctionality has increased significantly. Among these, transparent fluorescent ferroelectric ceramics exhibiting ferroelectricity, optical transparency, and photoluminescence (PL) have garnered significant attention. However, an interdependent relationship exists in a ferroelectric material among polarization, transparency, and photoluminescence, which presents a challenge for optimizing the coupling of optoelectronic properties. In this work, the doping concentration of Er³⁺ in 0.825(K_{0.5}Na_{0.5})NbO₃-0.175Sr(Sc_{0.5}Nb_{0.5})O₃: x%Er (x = 0–0.15) system was modulated by first-principle calculations through compositional design and performance-influencing-factor-analysis strategies. The experimental results showed that grain size of the ceramic was reduced to 28 μm at x = 0.05, concentration of vacancy defects in the lattice was low, and band gap value was increased to 3.105 eV. The multifunctional ceramic, while maintaining an excellent recoverable energy storage density ($W_{\text{rec}} = 2.03 \text{ J/cm}^3$) and energy storage efficiency ($\eta = 75.67\%$), demonstrated a 56% (1100 nm) good near-infrared transmittance and upconversion photoluminescence properties at 527, 549 nm, and 667 nm exhibiting weak green, strong green, and weak red light, respectively. This study provides a theoretical foundation and new approach for realizing the multifunctionality of photoelectric couple by introducing rare earth elements as luminescent centers into ferroelectric ceramics.

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1. Introduction

As electronic components move towards miniaturization and integration, single-function materials can no longer meet the increasingly complex and varied needs of electronic devices [1–6]. The development of integrated multifunctional materials has become a hot research topic in the field of electronic components [7–13]. Ferroelectric ceramics are regarded as an ideal substrate for realizing multifunctional materials due to their ferroelectric

properties and high-temperature stability [14–17]. The introduction of Sm³⁺ into BNT-BT ceramics enables synergistic enhancement of electrical and luminescent properties, providing new possibilities for the development of optoelectronic integration and coupling devices [18]. PNN-PZT ceramics exhibit piezoelectric effects and varistor properties, enabling kinetic responses to specific critical application scenarios [19]. The diverse physical environments triggered by different doping sites of Ce in BT-based ceramics, the feasibility of multifunctional BT-based ferroelectric ceramics with excellent electrothermal and electrostrictive effects is verified [20,21]. Rare earth elements are often used as active ions in luminescent materials due to their unique 4f inner layer structure [22–25]. By introducing rare earth elements into ferroelectric ceramics as light-emitting centers, the versatility of optoelectronic

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coupling can be realized, which provides strong support for the development of new functional materials [26–29]. To integrate the optical properties into the ferroelectric system, low phonon energy host materials with nonradiative relaxation properties were selected, which can effectively improve the fluorescence lifetime and upconversion photoluminescence (UCPL) efficiency of RE intermediate energy levels [30,31]. KNN-based ferroelectric ceramics exhibit excellent dielectric, ferroelectric, and optical properties and are considered as one of the most promising lead-free multifunctional ferroelectric ceramics [32–36]. Their low phonon energy levels also make them suitable PL substrates. Therefore, KNN-based ceramics have been noticed and studied in the field of multifunctional transparent ferroelectric materials [37–39].

Sm^{3+} -doped $(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3$ transparent ceramics have exhibited reversible photochromic (PC) behavior and the related modulation of transmittance and luminous intensity [40]. Ho^{3+} -doped $(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3\text{--SrTiO}_3$ transparent ceramics exhibit high transmittance in the near-infrared (NIR) region (~70%) and possess a stable emissive color at low temperature [41]. Among rare earth ions, erbium (Er) is one of the most commonly used upconversion ions. Its energy level structure includes a $^4\text{I}_{15/2} \rightarrow ^4\text{I}_{11/2}$ transition, a process highly responsive to low-energy photon excitation [42]. Er-doped $(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3\text{--}(\text{Ba}, \text{Sr})\text{TiO}_3$ ceramics demonstrate a high degree of reversible modulation in transparency, photoluminescence intensity, and conductivity. These properties endow the ceramic material with an exceptional ability to store and modulate optoelectronic information [43]. KNNLB-Er ceramics demonstrate 50% transmittance in the visible range with notable modulation contrast, where ΔAbs reaches 16.2%, and ΔR is 34% [44]. However, there is a scarcity of report on ferroelectric materials that possess photoluminescence (PL) performance while maintaining both significant energy storage properties and high transmittance, both experimentally and mechanistically. Therefore, there is an urgently needed to develop a transparent ferroelectric ceramic material that enables the effective coexistence of transmittance, luminescence, and electrical properties.

In this investigation, by adopting a composition-performance influencing factor strategy, different concentrations of Er^{3+} were doped into the KNN-SSN system through density functional theory (DFT) calculations, and the changes in electronic structure and related properties were assessed semi-quantitatively. The 0.825KNN-0.175SSN: $x\%\text{Er}$ multifunctional ceramics exhibit not only excellent energy storage characteristics and high transmittance but also significant photoluminescence effects. As illustrated in Fig. 1, optimized doping of Er^{3+} has reduced the grain size of the material, significantly enhancing its energy storage performance. The combined effects of lattice distortion, changes in vacancy concentration, and an increase in bandgap energy have collectively contributed to the remarkable transparency and upconversion photoluminescence (UCPL) behavior as observed. Our systematic analysis of the crystal structure, ferroelectric properties, and optical characteristics of the ceramics, the underlying mechanisms offers both a theoretical foundation and practical guidance for developing high-performance multifunctional optoelectronic devices.

2. Experimental section

Ceramic samples were prepared by stoichiometric ratio 0.825 $(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3\text{--}0.175\text{Sr}(\text{Sc}_{0.5}\text{Nb}_{0.5})\text{O}_3$: $x\%\text{Er}$ ($x = 0, 0.05, 0.10, 0.15$) from starting chemical reagents K_2CO_3 (99.0%), Na_2CO_3 (99.8%), Nb_2O_5 (99.5%), SrCO_3 (99.0%), Sc_2O_3 (99.9%), Er_2O_3 (99.9%). After the raw materials were weighed, the ball mill for the raw mixture was performed with anhydrous ethanol for 12 h, and then the mixed slurry was dried at 80 °C until the anhydrous ethanol

was completely volatilized. The dried mixture was calcined in a crucible at 950 °C for 5 h, and then the calcined powder was ball-milled again for 12 h. After drying, the powder was mixed with 5% (mass fraction) PVA solution and ground. Then, disk-like samples with a diameter of 12 mm and a thickness of 0.3–0.5 mm were obtained by isostatic pressing at 250 MPa. Finally, the samples were sintered at 1200–1300 °C for 2 h, and then polished to a thickness of 0.3 mm for transmittance test, and polished to a thickness of 0.15 mm for ferroelectric performance measurement.

The phase structure of the samples was characterized by X-ray diffraction (XRD, D8 Advance A25, Bruker, Germany). The surface microstructure and element localization of the samples were detected by field emission scanning electron microscopy (FE-SEM, Nova Nano SEM450, FEI, Czech). The distribution of the corresponding particle size was estimated by FESEM micrograph measurement. The transmittance of the sample was measured using an ultraviolet–visible spectrophotometer (UV-3600, Shimadzu Co, Japan). Upconversion photoluminescence spectra were measured under 980 nm laser excitation (FLS1000, Edinburgh). The dielectric response spectra at different frequencies and temperature ranges were measured using a precision impedance analyzer (LCR meter, TH2382, China). The polarization electric field (P – E) hysteresis loop was measured at 100 Hz using a ferroelectric test system (PolyK Technologies, USA). The AC impedance spectra were measured by an electrochemical workstation (CHI700e, CH Instruments Inc, China) from 460 °C to 640 °C. X-ray photoelectron spectroscopy (XPS, Axis Ultra DLD, United Kingdom).

First-principles calculations were performed using Materials Studio software based on density functional theory (DFT) and the supercell method in conjunction with the virtual crystal approximation (VCA). To address electron–electron interactions, the Perdew–Burke–Ernzerhof (PBE) functional within the generalized gradient approximation (GGA) was utilized [45].

The supercell calculation of KNN-SSN: $x\%\text{Er}$ systems with different components was carried out, and the atomic-scale theoretical model of KNN-based systems was constructed. Specifically, a $2 \times 3 \times 3$ orthogonal KNN supercell was established by the particular quasi-random structure (SQS) method, containing 150 atoms [46]. The calculated formation energy information at different positions in the Er-substituted KNN lattice (see Fig. 1) shows that both E_f ($\text{Er} \rightarrow \text{Nb}$) and E_f ($\text{Er} \rightarrow \text{O}$) are positive, while E_f ($\text{Er} \rightarrow \text{K}$) and E_f ($\text{Er} \rightarrow \text{Na}$) are negative, and E_f ($\text{Er} \rightarrow \text{Na}$) is slightly lower than E_f ($\text{Er} \rightarrow \text{K}$). From the thermodynamic point of view, it is shown that Er mainly replaces Na at the A-site. This confirms that Er is thermodynamically favorable for the A-site doping of the KNN-SSN system, and Er doping can be achieved by introducing Er_2O_3 into the KNN-SSN system.

By varying the A-site K, Na, and Sr atoms and the B-site Nb, Sc, and O atoms, different levels of Er^{3+} doping were introduced into the KNN-SSN system using the VCA method. The wave functions were expanded using a plane wave basis set with a cut-off energy of 620 eV, while Brillouin zone integration was performed with a $7 \times 7 \times 7$ Monkhorst–Pack k-point grid. The energy convergence criterion was set to less than 1×10^{-6} eV/atom, and the force convergence criterion was set to less than 0.01 eV/Å. The Dmol3 module was employed for crystal orbital overlap population (COOP) analysis to effectively elucidate the orbital overlap effects.

3. Results and discussion

The X-ray diffraction (XRD) patterns of 0.825KNN-0.175SSN: $x\%\text{Er}$ ($x = 0, 0.05, 0.10, 0.15$) ceramics with different contents at room temperature as well as localized magnification in the range of 45°–46° are shown in Fig. 2a. The results show that all samples have a pure chalcogenide single-phase structure without impurity

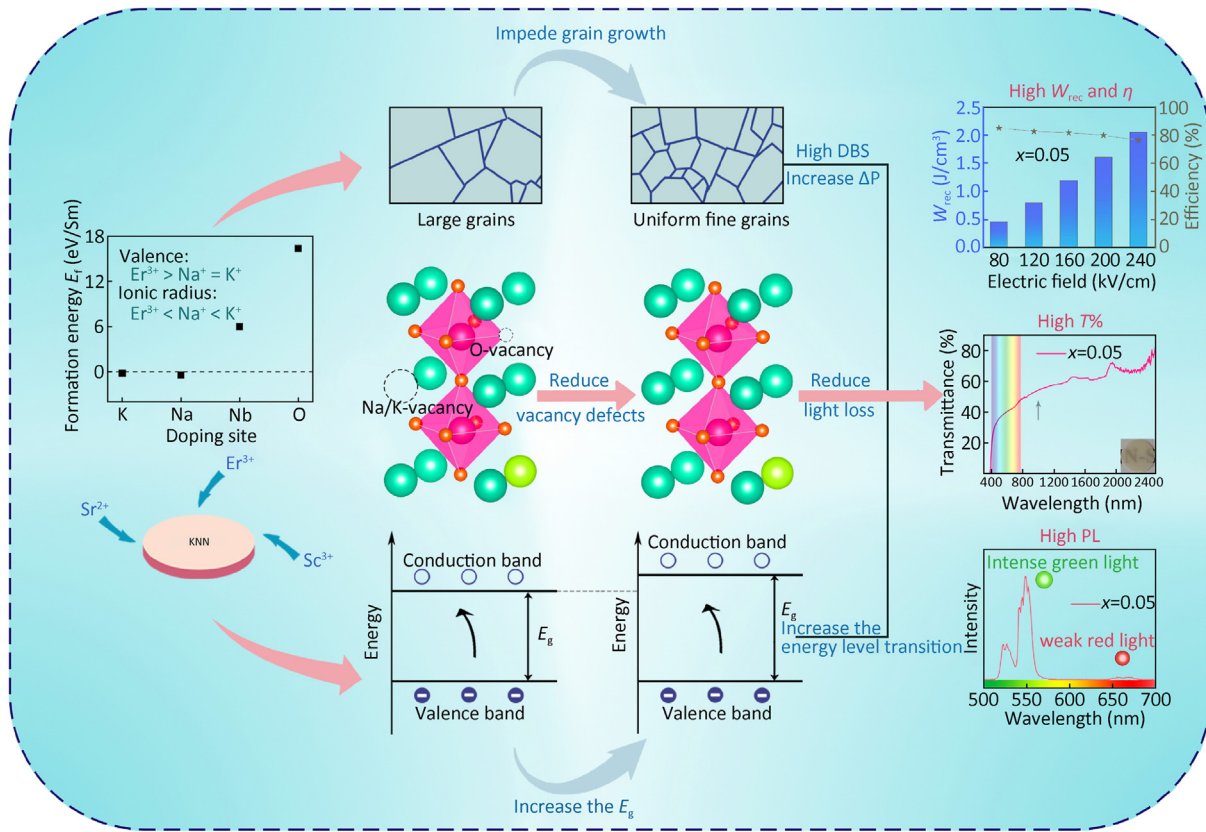


Fig. 1. Schematic diagram of the strategy for tuning energy storage performance, transmittance, and photoluminescence behavior of 0.825KNN-0.175SSN ceramics by doping with Er^{3+} .

secondary phases, indicating that Er^{3+} , Sr^{2+} , and Sc^{3+} have diffused into the KNN lattice to form a new solid solution. The doping element Er^{3+} ions, which have a radius (1.23 Å, CN = 12) close to that of the host K^+ (1.64 Å, CN = 12), Na^+ (1.39 Å, CN = 12), and Sr^{2+} ions (1.44 Å, CN = 12), will preferentially occupy the A-sites of the ABO_3 chalcogenide structure, which leads to lattice contraction. The localized magnified image shows that the diffraction peak position shifts to a higher angle with increasing Er content, further confirming this substitution phenomenon. A single peak of (200) is present in all samples in the 45° – 46° angle range for a pseudo-cubic phase structure with high symmetry. Ceramics with pseudo-cubic crystal structures usually have high optical transparency due to the small light scattering at the grain boundaries of isotropic lattice structures [47,48].

To accurately quantify the changes in the phase structure and lattice parameters of the 0.825KNN-0.175SSN: $x\%\text{Er}$ ceramics, a tetragonal-phase ($P4mm$) based Rietveld refinement was performed using GSAS software [49,50]. Additional details of the relevant Rietveld refinement structural parameters and tetragonality (c/a) are shown in Table S1 and Fig. 2b. Reliability coefficients (R_{wp} and R_p) are below 10%, which ensures the confidence of the final refined fitting results. The lattice parameters a , b , and c vary significantly with increasing Er content, leading to different degrees of lattice distortion. The c/a ratio is commonly used to characterize the degree of lattice distortion in tetragonal phases, and the more the value deviates from 1, the lower the structural symmetry [51]. Fig. 2c–d illustrate the Rietveld refinement results for the $x = 0$ and $x = 0.05$ components, respectively. The results for the remaining components are shown in Fig. S1. Compared to the 0.825KNN-0.175SSN ($c/a = 0.99997$), the c/a ratio increases to

1.00001 at $x = 0.05$, indicating enhanced symmetry of the structure after doping. Fig. 2e demonstrates the change in atomic displacement in the oxygen octahedron before and after doping, where the B-site Nb ion moves along the [001] direction, resulting in an increase in the Nb– O_{a2} bond length and a decrease in the Nb– O_{a1} bond length. This change results in an increase in the O–Nb–O bond angles in the octahedra and an improvement in the lattice symmetry.

Microstructural scanning of the thermally corroded surface of the ceramic samples after polishing was carried out using field emission scanning electron microscopy (FE-SEM) to investigate the effect of Er doping on the grain size of 0.825KNN-0.175SSN ceramic, and the relevant results are shown in Fig. 3a–d. The results showed that all ceramic samples exhibited a dense microstructure without obvious pores. With the increase of Er content, the average grain size of the ceramic samples showed a tendency to decrease and then increase, as shown in Fig. 3e. Especially at $x = 0.05$, the ceramic grains were fine and uniform, with a high density of grain boundaries, which helped to obtain high DBS and excellent transparency [52,53]. This indicates that a moderate amount of Er can effectively inhibit the grain growth of 0.825KNN-0.175SSN ceramics. This may be related to the doping of Er^{3+} as a high valence donor, when Er^{3+} replace the A-site in the KNN lattice, additional cationic vacancies (e.g., V'_{K} or V'_{Na}) are generated to maintain the space charge balance due to the difference in valence states [54]. This leads to a decrease in the lattice diffusion coefficient and a weakening of mass transfer, thus inhibiting grain growth [38,55]. In addition, reducing the grain size helps to enhance the coupling effect between grains and grain boundaries, which in turn hinders the domain flip, and is conducive to lowering the residual polarization strength and improving the

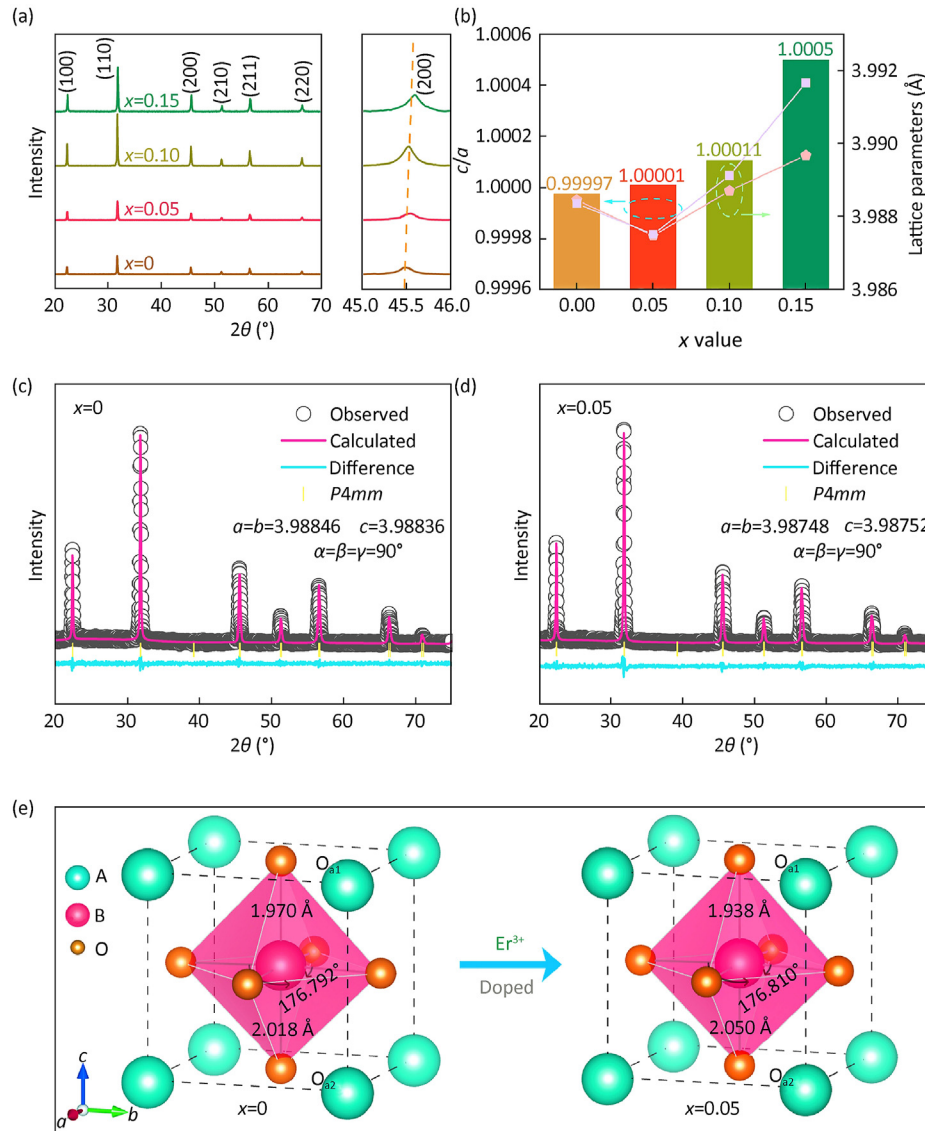


Fig. 2. 0.825KNN-0.175SSN: x %Er ceramics in the 2θ range of (a) XRD spectra at 20° – 70° and 45° – 46° ; (b) Values of the lattice parameters a , b , c and the c/a ratio; Rietveld refinement results for (c) $x = 0$ and (d) $x = 0.05$ samples; (e) Schematic representation of the protocell structure of doped and undoped ceramics.

energy storage performance [56,57]. EDS spectroscopic analyses confirmed the presence of the elements K, Na, Sr, Sc, Nb, O, and Er in the samples, as shown in Fig. 3f.

The measure of transparency is the percentage of the remaining light energy I to the total energy T after the light with an energy of I_0 passes through the material [57–60]. Formula (1) is as follows:

$$T = \frac{I}{I_0} = (1 - R_2) \exp[-(\gamma + \alpha) \cdot t] \quad (1)$$

where T is the transmittance, R_2 is the reflection loss of the sample on two surfaces, γ is the total scattering coefficient, α is the absorption coefficient, and t is the thickness [61]. For polycrystalline ceramics, light scattering caused by a large number of grain boundaries and pores inside is the main factor affecting transparency [62]. The transmission spectra and photos of 0.825KNN-0.175SSN: x %Er ceramics are shown in Fig. 4a. Through the ceramic sheet, the letters on the paper can still be distinguished. It can be seen from Fig. 4a that the transmittance of the $x = 0.05$ ceramic sample in the visible light (780 nm) and near-infrared region

(1100 nm) is 48% and 56%, respectively. This may be due to the decrease in light scattering and the increase of light transmittance caused by the high symmetry of the pseudo-cubic phase lattice ($c/a = 1.00001$) and the small grain size. According to the Tauc equation, the optical band gap value (E_g) can be estimated from the absorption spectrum:

$$(\alpha h\nu)^2 = A(h\nu - E_g) \quad (2)$$

where ν is the photon frequency, h is Planck's constant ($h = 4.13571015$ eV), and A is a constant and α is the $-\lg(T)$ value [61]. Based on the obtained values of $(\alpha h\nu)^2$ versus $h\nu$, the band gap plots of different components are plotted as shown in Fig. 4b. The intercept of the curve with the x -axis is the band gap value. The sample with $x = 0.05$ has the largest E_g value (3.04 eV). Higher E_g value helps to reduce the energy consumption of electron leaps and thus increases the transmittance following the law of transmittance.

The upconversion spectra and luminescence mechanism of 0.825KNN-0.175SSN: x %Er ceramics are presented in Fig. 4c–d.

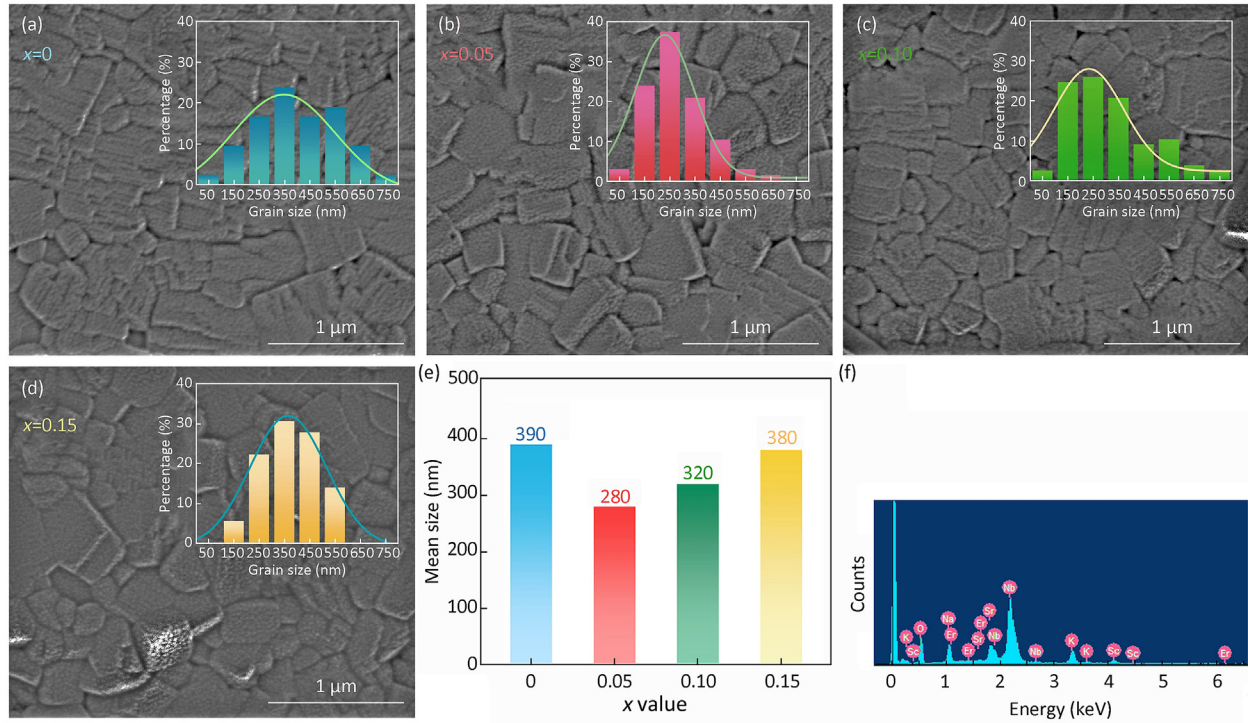


Fig. 3. SEM micrographs of the polished and thermally corroded surfaces of 0.825KNN-0.175SSN: x%Er ceramics with insets showing histograms of the grain size distributions: (a) $x = 0$; (b) $x = 0.05$; (c) $x = 0.10$; (d) $x = 0.15$; (e) Average grain sizes of the component ceramics; (f) Elemental EDS spectra of the $x = 0.05$ ceramic samples.

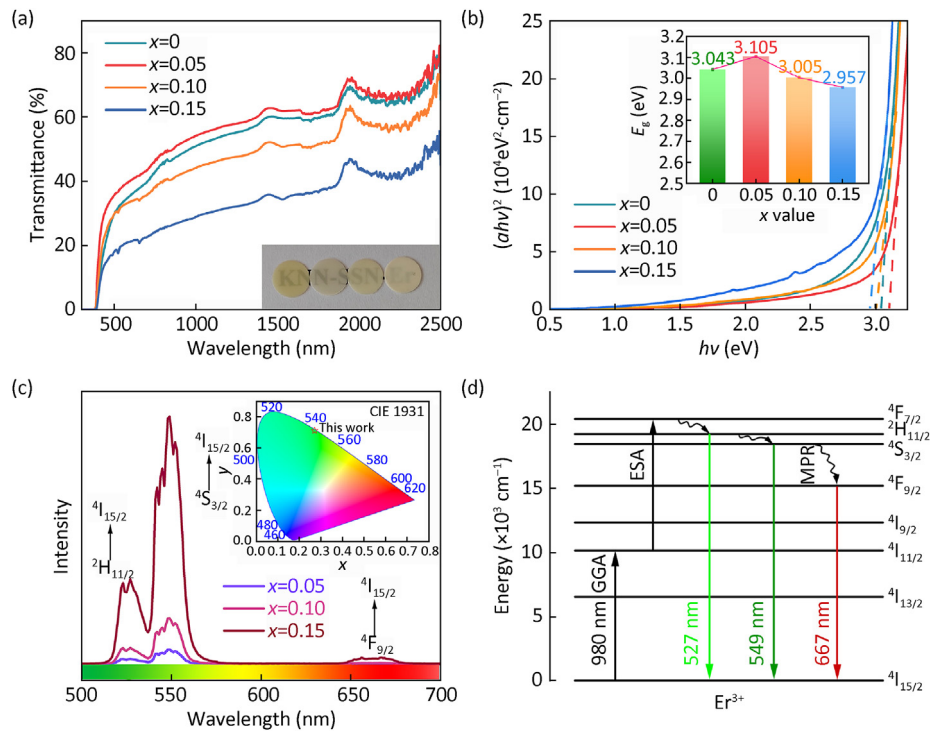


Fig. 4. (a) Optical transmission spectra and photographs of 0.825KNN-0.175SSN: x%Er ceramic samples; (b) Plot of $(ahv)^2$ vs. $h\nu$, with the inset showing the bandgap columnar distribution of each component; (c) UCPL spectra of 0.825KNN-0.175SSN: x%Er ceramics (inset showing the CIE 1931 x-y color-matching function); (d) Schematic diagram of UCPL luminescence energy level migration.

Using a 980 nm light source to excite the samples, the Er^{3+} in the ground state $4I_{15/2}$ energy level absorb the photons (ground state absorption, GSA) and migrate to the excited state $4I_{11/2}$ energy level

[43]. Some of these Er^{3+} can populate the $4F_{7/2}$ energy level by upconversion of two-step excitation (excited state absorption, ESA). Subsequently multiphonon relaxation occurs, and the Er^{3+}

sequentially relax to the $^2\text{H}_{11/2}$, $^4\text{S}_{3/2}$, and $^4\text{F}_{9/2}$ energy levels, generating emitted light by radiative relaxation to the ground state (Fig. 3d). Due to the small energy difference of the energy levels, a large number of Er^{3+} in the $^4\text{F}_{7/2}$ energy level relax to the $^2\text{H}_{11/2}$ energy level and then quickly relax further to the $^4\text{S}_{3/2}$ energy level. This leads to the formation of weak green light and strong green light (527 nm and 549 nm) from the radiative transitions of the $^2\text{H}_{11/2}$ and $^4\text{S}_{3/2}$ energy levels to the ground state, respectively. The weak red light corresponds to the transmigration of the $^4\text{F}_{9/2}$ energy level to the ground state. Due to the large energy difference, only a small number of Er^{3+} at the $^4\text{S}_{3/2}$ energy level will non-radiatively relax to the $^4\text{F}_{9/2}$ energy level, resulting in a weak red emission at 667 nm (Fig. 4c). Fig. S3a illustrates that when the Er atom is positioned at the A site, there is a pronounced shift in the position of the neighboring O atom towards the Er atom. This shift leads to the distortion of the NbO_6 octahedron and creates a locally non-uniform chemical bonding environment. In contrast, Fig. S3b shows that the chemical bonding environment in the B-site plane, which is distant from the Er atom, remains highly uniform. This observation indicates that the impact of Er substitution at the A site is relatively localized. Notably, this locally non-uniform chemical bonding environment is known to enhance PL behavior [39].

The functional mechanism of the carrier in 0.825KNN-0.175SSN: $x\%\text{Er}$ ceramic can be explained in reference to impedance spectroscopy. The Zview software is used to fit the impedance data from low frequency to high frequency to understand the resistance contribution of grain boundaries and grains [63]. Based on the equivalent circuit model (Fig. S2a) consisting of a combination of R_{gb} , C_{gb} (grain boundary resistance and capacitance) and R_g , C_g (grain resistance and capacitance) elements, Z' and Z'' (Nyquist or Cole-Cole plots) (Fig. 5a and d) and C' spectra (Fig. 5b and e) were plotted at several temperatures, as well as at 510 °C and Z'' and M'' overlap plots at 510 °C (Fig. 5c and f) to represent the complex impedance behavior of the ceramics. Fig. 5a and d shows that the resistance of grains and grain boundaries decreases with increasing temperature, and a negative temperature coefficient resistance effect (NTCR effect) exists [64]. When $x = 0, 0.05$, a single, regular Debye-like semicircle appears in the Nyquist diagram, indicating that the grain boundary resistance R_{gb} is significantly higher than the grain resistance R_g , and the grain boundary contribution is more closely related to the conductivity behavior. The atomic arrangement near the grain boundary is disordered and electron scattering increases [65]. On the contrary, when $x = 0.10, 0.15$ (Fig. S2b), two semicircular arcs are observed from the figure, where the high-frequency region is related to the grain effect and the low-frequency region is related to the grain boundary effect [66]. This implies that the enhanced resistive contribution of the grains may lead to an increase in the grain size. The distribution of low and high capacitance plateaus in the C' spectrum is related to the grain effect and grain boundary effect [67]. It has been observed that the C' spectrum of 0.825KNN-0.175SSN: 0.05%Er (Fig. 5e) shows a low capacitance plateau (~ 0.11 nF/cm) in the high-frequency range and a high capacitance plateau (~ 2.45 nF/cm) in the low-frequency range. In addition, the normalized spectra of Z'' and M'' (Fig. 5c and f) showed that after Er^{3+} modification, the peak positions of $Z''/Z''_{\max}$ and $M''/M''_{\max}$ had an overlapping tendency, indicating an increase in electrical homogeneity, which improves the resistivity of the 0.825KNN-0.175SSN: 0.05%Er ceramic [68]. The corresponding impedance results are in agreement with the conclusion of fine and uniform grain size distribution exhibited by SEM.

The variation of the thermal conductivity of the grains is shown in Fig. 6a, and the linear variation of $\ln(\sigma T)$ versus $1000/T$ proves that there is no phase transition in the temperature range studied. The gradual increase of the conductivity value with temperature

proves that the conductivity in this material corresponds to a thermally activated transport process. The linear region of this can be fitted with the Arrhenius equation below:

$$\sigma = \sigma_0 \exp\left(\frac{-E_a}{k_B T}\right) \quad (3)$$

where E_a is the activation energy for grain conductivity, σ is a pre-exponential factor including charge carrier mobility and density of states, k_B is Boltzmann's constant, and T is the Kelvin temperature [66]. When the value of the activation energy (E_a) is less than 1 eV, it is an oxyanionic conduction. On the contrary, an activation energy greater than 1 eV corresponds to electron conduction [69]. Fig. 6a–b shows the Arrhenius fitted line plots and the corresponding E_a values summarized for each component, respectively. As a whole, the E_a value ranges from 0.84 eV to 1.55 eV, which is less than half of the corresponding ceramic E_g (Fig. 4b). Among them, the highest E_a value of 1.55 eV is found at $x = 0.05$. This indicates that this sample has a lower concentration of vacancy defects, which is favorable for transparency. Meanwhile, the $x = 0.10$ and 0.15 samples have smaller E_a values, resulting in lower carrier transport energy barriers. The E_a value of the $x = 0.15$ sample is less than 1 eV, indicating that the conduction process may be mainly caused by the mobility of the oxygen vacancies, with a higher concentration of vacancy defects and lower transparency of the sample. This can be attributed to free electron and vacancy defect generation due to excess Er doping. It can be inferred that the concentration of oxygen vacancies in the system decreases when a small amount of Er^{3+} ions substitute for the A-site, as illustrated by Formula (4):



However, when the Er^{3+} ion is overdoped, the non-equivalent donor substitution will lead to more free electrons to be generated to maintain the electrically neutral state. As shown below:



An additional analysis employing X-ray photoelectron spectroscopy (XPS) was conducted to elucidate the variations in oxygen vacancy defect concentrations within the ceramic samples [70,71]. The obtained spectra, as illustrated in Fig. S4, reveal distinct peaks for the 0.825KNN-0.175SSN and 0.825KNN-0.175SSN: 0.05%Er ceramics, with the O 1s XPS spectra detailed in Fig. 5g–h. The observed three peaks at 529.7, 531.4 eV, and 532.2 eV are respectively attributed to lattice oxygen (O1), oxygen vacancies (O2), and surface-adsorbed oxygen (O3). A quantitative analysis of the relative peak areas of O2, as depicted in Fig. 5i, indicates a reduction in oxygen vacancy concentration from 32.83% to 28.83% in the 0.825KNN-0.175SSN: 0.05%Er ceramic compared to the 0.825KNN-0.175SSN counterpart, consistent with the observed changes in transmittance as shown in Fig. 4a.

The A-site substitution of Er^{3+} may lead to the creation of K/Na vacancies in the 0.825KNN-0.175SSN lattice to balance the excess charge induced by the doping of ions with different valence states. This process may lead to the formation of defective dipoles, whose reorientable properties in the presence of an external electric field play an important role in improving the dielectric properties of 0.825KNN-0.175SSN ceramic [72]. Fig. 7a, b demonstrates the variation of dielectric constant (ϵ_r) and dielectric loss ($\tan\delta$) with temperature for 0.825KNN-0.175SSN and 0.825KNN-0.175SSN: 0.05%Er ceramic at different frequencies. Compared with 0.825KNN-0.175SSN, 0.825KNN-0.175SSN: 0.05%Er ceramic had a temperature of maximum permittivity (T_m) shifted toward lower

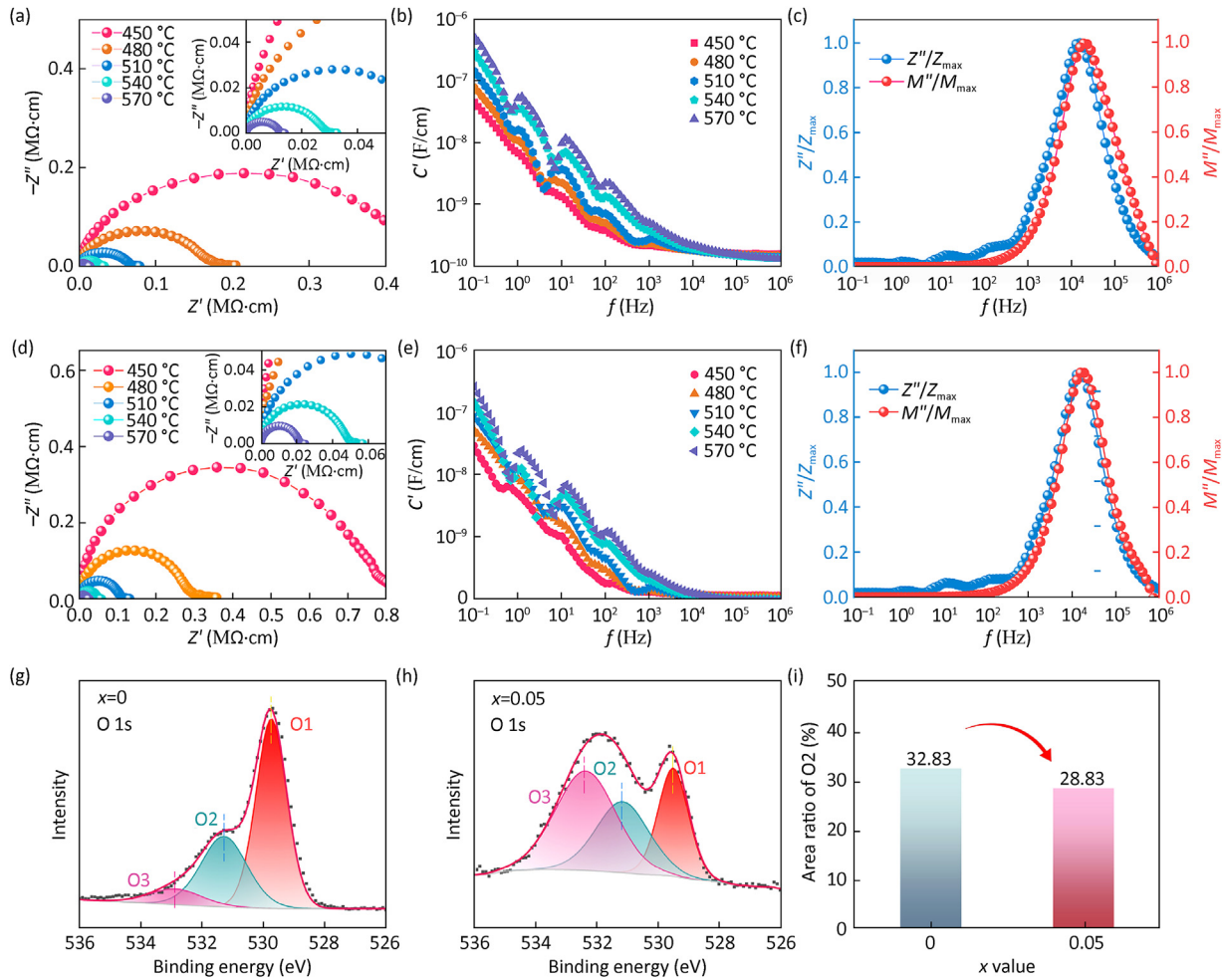


Fig. 5. Composite impedance spectra of (a) 0.825KNN-0.175SSN and (d) 0.825KNN-0.175SSN: 0.05%Er; C' spectra of (b) 0.825KNN-0.175SSN and (e) 0.825KNN-0.175SSN: 0.05%Er; Frequency correlation plots of $Z''/Z'_{\max}$ and $M''/M'_{\max}$ for (c) 0.825KNN-0.175SSN and (f) 0.825KNN-0.175SSN: 0.05%Er; XPS spectra of O 1s: (g) 0.825KNN-0.175SSN, (h) 0.825KNN-0.175SSN: 0.05%Er, (i) O2 photoelectron peak area change.

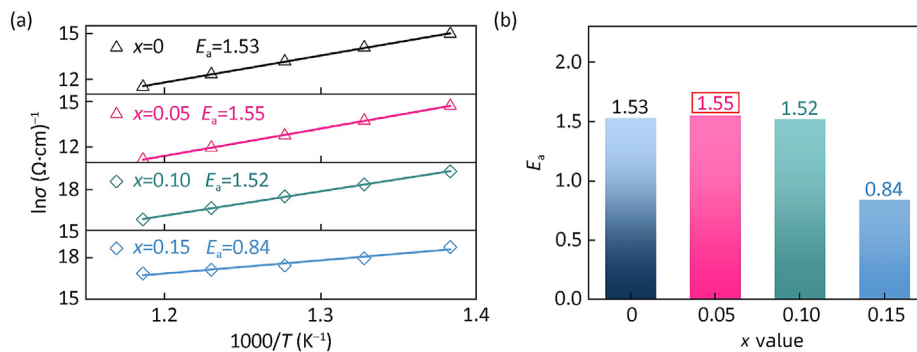


Fig. 6. (a) Arrhenius-type plots of the conductivity with the temperature of 0.825KNN-0.175SSN: x%Er ceramic and (b) the simulated E_a values change diagram.

temperature, and obvious diffusive phase transition and frequency dispersion phenomena appeared, which indicated that the 0.825KNN-0.175SSN: 0.05%Er ceramic exhibited typical relaxation ferroelectric features. This phenomenon may be due to the localized distortion of the 0.825KNN-0.175SSN lattice induced by the doping of Er^{3+} with different radii and valence states, which in turn increases the disorder of the lattice. This distortion disrupts the long-range ferroelectric ordering and favors the generation of polar

nanoregions (PNRs). Due to the presence of PNRs and the internal disorder of the ceramic samples, some hysteresis relaxation occurs in the polarization response of the ceramics upon application of an external electric field. The dielectric constant of the material was fitted using the modified Curie-Excise law to further analyze the relaxation behavior of the material. The formula is calculated as follows:

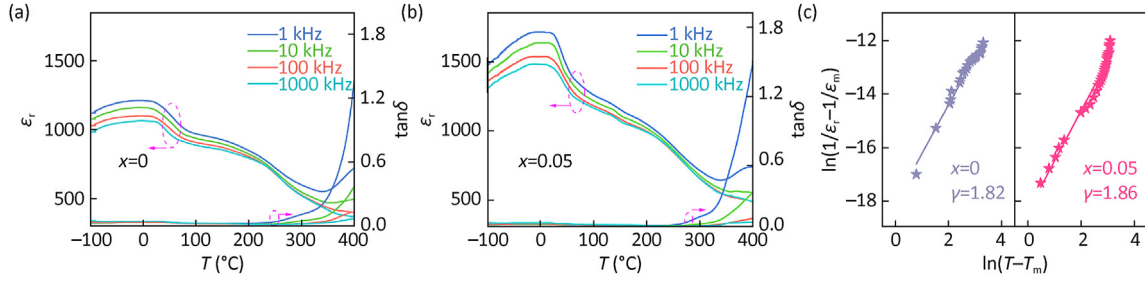


Fig. 7. (a) Temperature-dependent dielectric response of 0.825KNN-0.175SSN and (b) 0.825KNN-0.175SSN: 0.05%Er; (c) $\ln(1/\varepsilon - 1/\varepsilon_m)$ as a function of $\ln(T - T_m)$ and the corresponding γ value for 0.825KNN-0.175SSN and 0.825KNN-0.175SSN: 0.05%Er samples, at 1000 kHz.

$$\gamma = \ln(T - T_m) \ln(1/\varepsilon - 1/\varepsilon_m) \quad (6)$$

where γ is the diffusion coefficient, T_m is the temperature corresponding to the maximum dielectric constant, and ε_m is the maximum dielectric constant. The change in the value of γ from 1 to 2 marks the transition from a normal ferroelectric to an ideal relaxing ferroelectric [66]. Fig. 7c shows the $\ln(1/\varepsilon - 1/\varepsilon_m)$ at 1000 kHz versus $\ln(T - T_m)$, the diffusivity γ reaches 1.86 when $x = 0.05$, which indicates that the 0.825KNN-0.175SSN: 0.05%Er sample exhibits more significant relaxation characteristics.

The P - E rings of 0.825KNN-0.175SSN: $x\%$ Er ceramic samples and the corresponding evolution of maximum polarization ($P_{\max}$) and residual polarization strength (P_r) measured at room temperature at 100 kV/cm are shown in Fig. 8a and c. The P - E rings of 0.825KNN-0.175SSN: $x\%$ Er ceramic are shown in Fig. 8a. The tilted elongated P - E rings indicate that the ceramics exhibit relaxation ferroelectric behavior, which is promising for energy storage capacitor applications. Both $P_{\max}$ and P_r first decrease and then increase with increasing Er content, which is attributed to the fact that a moderate amount of Er_2O_3 disrupts the long-range ferroelectric ordering of the 0.825KNN-0.175SSN ceramics, which leads to the creation of polarized nanoregions (PNRs) and enhances the

relaxation properties of the ceramics. The energy storage density and energy storage efficiency can be expressed by the following equations [73]:

$$W_{\text{rec}} = \int_{P_r}^{P_{\max}} E dP \quad (7)$$

where W_{rec} , W_{loss} , $P_{\max}$, P_r , and E represent the recoverable energy density, energy loss, maximum polarization strength, residual polarization strength, and electric field of the sample, respectively. By calculating the P - E curves, the energy storage characteristics of 0.825KNN-0.175SSN: $x\%$ Er ceramics under the action of 100 kV/cm electric field are shown in Fig. 8b, and $P_{\max}$ reaches a maximum value of $29.55 \mu\text{C}/\text{cm}^2$ at $x = 0.05$ and 86% efficiency. The P - E ring of 0.825KNN-0.175SSN ceramics becomes thinner after moderate Er doping, which indicates that Er doping can improve its energy storage performance. To further explore the energy storage properties of 0.825KNN-0.175SSN: 0.05%Er ceramics, the unipolar P - E rings were measured under different applied electric fields, as shown in Fig. 8d. The P - E ring consistently maintained a low P_r value over the applied electric field range (Fig. 8f). As a result, the η remained at a relatively high level as the $P_{\max}$ value increased linearly. Finally, an energy density of $2.03 \text{ J}/\text{cm}^3$ was obtained at

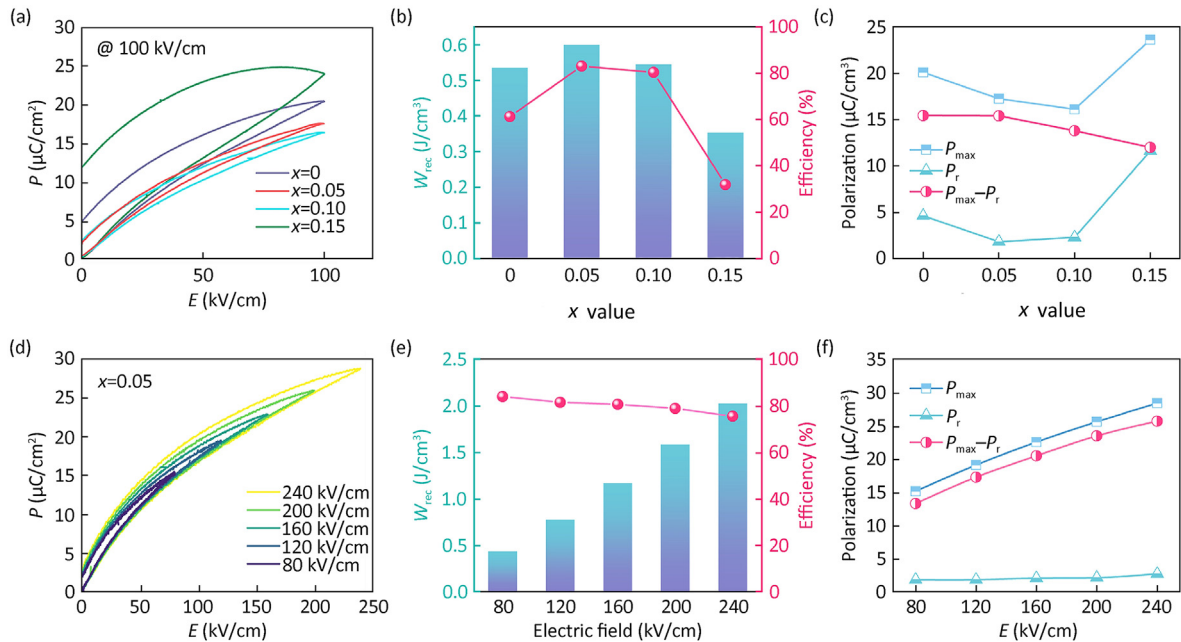


Fig. 8. (a) Unipolar P - E hysteresis loop, (b) W_{rec} and η values and (c) $P_{\max}$, P_r , and $P_{\max} - P_r$ values for 0.825KNN-0.175SSN: $x\%$ Er ceramic at 100 kV/cm; (d) Unipolar P - E hysteresis loops; (e) W_{rec} and η values; (f) $P_{\max}$, P_r , and $P_{\max} - P_r$ values for 0.825KNN-0.175SSN: 0.05%Er ceramics at different electric fields.

Table 1

Comparison of 0.825KNN-0.175SSN: 0.05%Er ceramic with ceramics of different systems, including KNN-based and BNT-based ceramics, in terms of recoverable energy storage density, energy storage efficiency, and light transmission rate.

Material	W_{rec} (J/cm ³)	η (%)	T (%)	Ref.
NaNbO ₃ -0.04CaZrO ₃	0.55	63.00		[67]
KNN-0.015 EB	1.85	69.00	<50.0	[74]
BNT-0.5NN	5.14	79.65	32.0	[75]
0.9KNN-0.1ST: 0.1Er	0.86	<60.00	39.1	[76]
KNN-0.08BaMn	0.80		>60.0	[77]
0.9BNT-0.1BZT: 0.6%Er ³⁺	2.95	51.30	28.0	[78]
0.825KNN-0.175SSN:0.05%Er	2.03	75.67	48.0	This work

240 kV/cm while η reached 75.67% (Fig. 8e). Table 1 compares the performance of 0.825KNN-0.175SSN: 0.05%Er ceramics with previously reported ceramics of different systems including KNN-based and BNT-based ceramics [67,74–78], in terms of recoverable energy storage density, energy storage efficiency, and light transmission. The results show that the 0.825KNN-0.175SSN: 0.05% Er ceramics are competitive in terms of electrical and optical performance.

For further clarification for the underlying mechanism of the crystal structure transformation of Er₂O₃ incorporated in the 0.825KNN-0.175SSN system, crystal orbitals overlapping totals

(COOP) simulations have been performed. The integrated COOP value (ICOOP) provides a way to quantify the bonding strength [79]. COOP is the bonding information extracted from the density states obtained by weighting the overlap totals and is used to show the strength and nature of the positive bonding (COOP > 0) or anti-bonding (COOP < 0) interactions, where a COOP value of zero indicates the absence of bonding interactions. The COOP curves of the Nb–O interactions before and after doping are presented in Fig. 9a–b, showing that there are significant Nb–O bonding interactions within the valence band, with two significant bonding peaks between –20 eV and 0 eV attributed to hybridization interactions between the 4d orbitals of the Nb atoms in the B-site and the O-2p orbitals. The O_p and O_a refer to the planar oxygen atoms and the apical of the oxygen atom. Compared to 0.825KNN-0.175SSN, the 0.825KNN-0.175SSN: 0.05%Er structure has a slightly lower bond energy for Nb–O_p and an increased bond energy for Nb–O_a, which may lead to distortions of the O_p planes and their neighboring NbO₆ octahedra. This is in agreement with the result of improved crystal structure symmetry with displacement of B-site ions along the [001] direction exhibited in Fig. 2e.

The stability of ferroelectricity in perovskite materials has been reported to be closely related to the degree of hybridization between the outermost d orbitals of the B-site atoms and the p orbitals of the oxygen atoms [28,80,81]. In KNN materials,

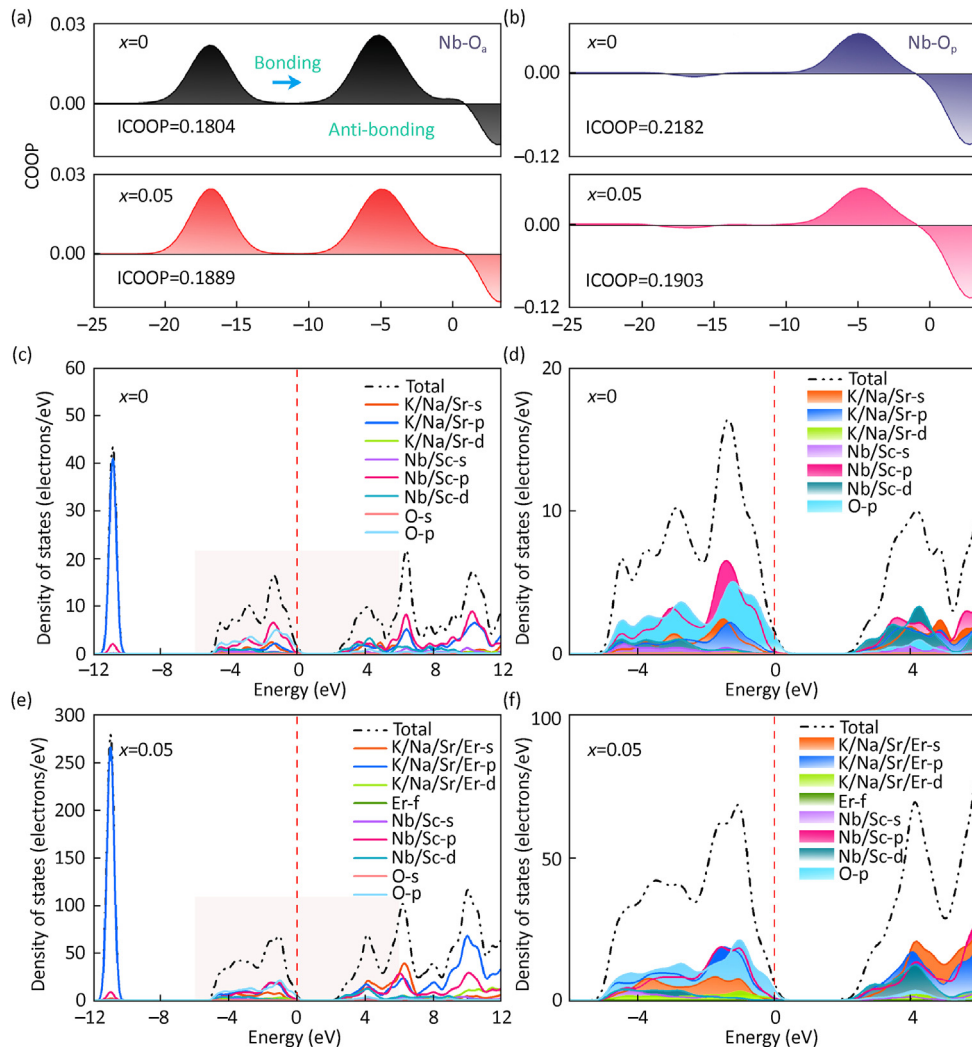


Fig. 9. Interaction COOP of (a) Nb–O_a and (b) Nb–O_p for the 0.825KNN-0.175SSN and 0.825KNN-0.175SSN: x%Er structures; (c–f) Total and partial density of states plots.

ferroelectricity mainly originates from the strong hybridization between the Nb-4d orbitals and the O-2p orbitals [60,76]. The hybridization effect between the 4d orbitals of the B-site Nb atoms and the 2p orbitals of the O atoms significantly attenuates the intrinsic short-range repulsive force, thus enhancing the stability of the ferroelectric structure. The quantification of this degree of hybridization can effectively assess the strength of ferroelectricity [76]. To deeply explore the mechanism of the change in ferroelectricity of the KNN-SSN system by Er_2O_3 doping, electronic density of states calculations were carried out. Fig. 9c–f demonstrates the total density of states (TDOS) and partial density of states (PDOS) analyses for the 0.825KNN-0.175SSN structure and the 0.825KNN-0.175SSN: 0.05%Er structure. As shown in Fig. 9c and e, the PDOS values of K/Na/Sr ions at the A-site in the KNN-SSN system are lower than 4 eV, indicating weak hybridization with the O-2p orbital. The valence band (−6 to 0 eV range) consists mainly of the synergistic contribution of the O-2p orbitals with the other ion orbitals, while the conduction band (2–6 eV range) is mainly dominated by the Nb-4d state, with a small contribution from O-2p. This suggests that the transfer of electrons from the valence band to the conduction band involves hybridization between the Nb-4d state and the O-2p state. In the doped KNN-SSN: Er system, the DOS distributions near the top of the valence band (VBM) and the bottom of the conduction band (CBM) are extended to higher energy ranges, and the difference in the PDOS values between the A-site ions and the O-2p state decreases, and the hybridization effect is enhanced. Meanwhile, the hybridization between the B-site Nb-4d orbitals and the O-2p orbitals is weakened, and the long-range ordered ferroelectric states are disrupted, which may lead to the coupled relaxation behavior that occurs in the 0.825KNN-0.175SSN: 0.05%Er ceramic. This is consistent with the low P_r phenomenon in the unipolar P – E loops of Fig. 8d.

4. Conclusions

Supported by first-principles calculations, the feasibility of simultaneously enhancing energy storage and transparent luminescence properties is investigated by doping various concentrations of Er^{3+} into ferroelectric KNN ceramic system. The ceramic samples of $(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3$ -0.175Sr($\text{Sc}_{0.5}\text{Nb}_{0.5}\text{O}_3$): $x\%$ Er were synthesized using the conventional solid-state reaction method. The results demonstrate that local lattice distortion and effective grain refinement enable the 0.825–0.175SSN: 0.05%Er ceramic to exhibit excellent energy storage properties, with a W_{rec} of 2.03 J/cm³ and an η -value of 75.67% under an electric field of 240 kV/cm for $x = 0.05$. In the meanwhile, the low concentration of vacancy defects and a high band gap ($E_g = 3.105$ eV) result in a transmittance of 53% for light of 1100 nm, and strong green upconversion luminescence. The high resistance (0.103 MΩ at 510 °C) and the electrical homogeneity of the samples strongly support their superior energy storage performance. The density-of-states map of Nb–O bonding, along with the analysis of ICOOP value variations, elucidates the mechanism underlying the relaxor ferroelectricity of the system. The presence of localized, inhomogeneous bonding environments on the A-site planes suggests that the A-site substitution of Er has a localized effect, which enhances the photoluminescence performance and lower remnant polarization with energy storage efficiency of the system. This study provides a robust theoretical and experimental foundation for developing rare-earth ion-doped lead-free ferroelectric ceramic with novel and multiple optoelectronic functions.

CRediT authorship contribution statement

Zhichao Gong: Writing – original draft, Formal analysis, Data curation. **Haojie Yue:** Methodology. **Kailing Fang:** Writing – review & editing, Software. **Kun Guo:** Writing – review & editing, Supervision, Investigation. **Kang Li:** Investigation, Data curation. **Chong Guo:** Software, Resources. **Huacheng Zhang:** Data curation. **Ziliang Deng:** Formal analysis. **Zhiyong Liu:** Writing – review & editing, Resources. **Bing Xie:** Writing – review & editing, Investigation. **Pu Mao:** Writing – review & editing, Methodology. **Jinshan Lu:** Validation, Resources. **Shifeng Guo:** Resources, Investigation, Conceptualization. **Kui Yao:** Writing – review & editing. **Francis Eng Hock Tay:** Writing – review & editing, Software.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.jmat.2024.100993>.

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