

Discovery of novel Thymol-TPP antibiotics that eradicate MRSA persisters

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Abstract:

The high prevalence of methicillin-resistant *Staphylococcus aureus* (MRSA) strains and the formation of non-growing, dormant "persisters" subsets help bacteria evade antibiotic treatment and enhance bacterial resistance, which poses a serious threat to human life and health. It is urgent to discover novel antibacterial therapies effective against MRSA persisters. Thymol is a common nutraceutical with weak antibacterial and antitumor activities. A series of Thymol triphenylphosphine (TPP) conjugates (**TPP-Thy3**) was designed and synthesized. These compounds showed significantly improved inhibitory activity against Gram-positive bacteria compared with Thymol. Among them, **Thy3d** displayed a low probability of resistance selection and showed excellent biocompatibility. Interestingly, **Thy3d** elicited a rapid killing effect of MRSA persisters (99.999%) at high concentration. Fluorescence experiments, electron microscopy, molecular dynamics simulation and bilayer experiment confirmed that **Thy3d** conjugates exerted potent antimicrobial activity by disrupting the integrity of the membrane of bacterial even the persister. Furthermore, **Thy3d** exhibited considerable efficacy in a mouse model of subcutaneous murine MRSA infection. In summary, **TPP-Thy3** conjugates are a series of novel antibacterial agents and could serve as a new therapeutic strategy for combating antibiotic resistance.

Keywords:

Antibacterial, Thymol, Triphenylphosphonium (TPP)-conjugates, Persister.

1. Introduction

Since the discovery of penicillin in 1928, antibiotics have become the most commonly used drugs for treating a variety of infections, and the widespread use of antibiotics has saved billions of patients' lives and extended human being's life for 15-20 years on average. However, the abuse and misuse of antibiotics have led to a drastic increase in bacterial resistance, posing a serious threat to public health[1]. Bacterial persisters, a small phenotype mutation subgroup, are often in a dormant, or inactive state, and can tolerate lethal concentrations of antibiotics and thus escape antibiotic treatment, leading to enhanced bacterial resistance[2]. Bacterial persisters play a critical role in perpetuating chronic and recurrent infections, which can not be treated with conventional antibiotics[3, 4]. Hence, the development of novel antimicrobial agents effective for treating resistant bacterial persisters is urgently needed.

However, few compounds have been reported with activity against persister cells[5]. Therefore, addressing drug resistance by eliminating persisters has become a research hotspot worldwide. Three strategies have been put forward to eliminate persisters[2]. One approach seeks to directly kill metabolically dormant persister cells[6, 7]. An alternative approach sensitizes the persister cells for conventional antibiotics through resuscitation, activating the antibiotic targets, or stimulating the antibiotic influx[8, 9]. The third approach employs molecules to disrupt or reduce the formation of persister cells. Triphenylphosphonium cation (TPP⁺), a lipophilic cation, can easily cross lipid bilayers due to its high lipophilicity and stable cationic charge, and the potential gradient drives the accumulation of TPP⁺-conjugates in the mitochondrial matrix[10]. There have been reports on the conjugation of TPP⁺ with drugs to improve their anticancer efficacy by increasing their distribution in the mitochondria[11, 12]. Since the bacterial cell wall is a phospholipid bilayer, TPP-conjugates could potentially cross the bacterial cell wall via similar interactions, accumulating in the bacterial cytoplasm and leading to an enhanced antibacterial effect[13, 14]. For example, the TPP-linked prodrugs ciprofloxacin-ester-PPh₃ and ciprofloxacin-amide-PPh₃ induced significant morphological changes in MRSA ST5 5016, including irregular deformation and membrane disruption, and some leakage of the cytoplasmic contents of bacteria to the extracellular environment[14]. Based on the membrane permeability of TPP⁺, we hypothesized that conjugation of TPP⁺ with natural products could further enhance its antibacterial activities, and have the potential to directly destroy persister cells.

Thymol is a naturally occurring phenolic monoterpene, which is one of the major constituents of the essential oils (10-64%) of thyme (*Thymus vulgaris* L., Lamiaceae), a medicinal plant with multiple therapeutic properties[15]. Thymol exhibits antimicrobial, antioxidant, anticarcinogenesis, anti-inflammatory, and antispasmodic activities, as well as a potential as a growth enhancer and immunomodulator[16]. However, due to the high concentrations required to reach therapeutic effects, the efficiency of Thymol in clinical applications was limited. Continuing our studies on novel antibiotic discovery[13, 17-23], we set out to conduct Structure-Activity Relationship (SAR) studies on Thymol in the hope of improving its antibacterial activity.

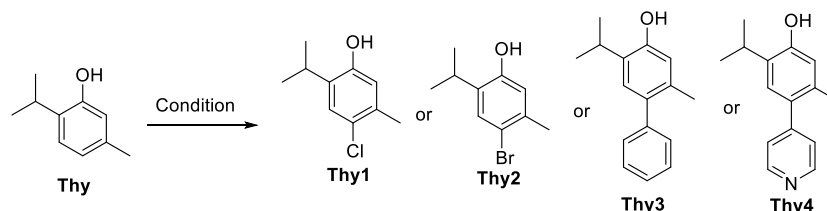
We synthesized a series of derivatives by introducing a substituent to the 4-position of Thymol. Antibacterial evaluation of these derivatives against the priority bacteria (*Staphylococcus aureus*, *Klebsiella pneumonia*, *Acinetobacter baumannii*, *Pseudomonas aeruginosa*, and *Enterobacter*

species) as defined by WHO identified derivatives with significantly improved antibacterial activities. We then selected the potent Thymol derivatives to form conjugates with TPP⁺, and further studied their SAR and antibacterial activities, leading to the discovery of Thymol-TPP conjugates effective against bacterial persisters. We also established their preliminary antibacterial mechanisms. The strategy reported in this study could help develop novel antibiotics to address the drug resistance challenges, from natural products that are often neglected in traditional bioassays due to their weak antibacterial activity.

2. Results and discussion

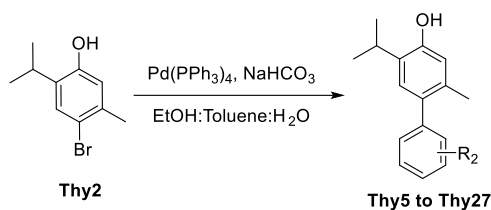
2.1 Synthesis of Thymol derivatives and TPP conjugates

The synthesis of designed Thymol derivatives **Thy1-4** is shown in Scheme 1. Commercially available Thymol reacted with SOCl₂ in CH₂Cl₂ to afford 4-chloro-2-isopropyl-5-methylphenol (**Thy1**), and TBABr₃ in MeOH/CH₂Cl₂ (v/v, 2:3) to afford 4-bromo-2-isopropyl-5-methylphenol (**Thy2**). Suzuki coupling of **Thy2** with boronic acids in EtOH/Toluene/H₂O (v/v/v, 2/2/1) produced **Thy3** and **Thy4**.



Scheme 1 Synthesis of **Thy1-Thy4**. Reaction conditions and reagents: for **Thy1**, SOCl₂, CH₂Cl₂; for **Thy2**, TBABr₃, CH₃OH/CH₂Cl₂ (v/v, 2/3); for **Thy3** and **Thy4**, Pd (PPh₃)₄, NaHCO₃, EtOH/Toluene/H₂O (v/v/v, 2/2/1). Thy means Thymol.

The synthesis of Thymol derivatives **Thy5-27** is shown in Scheme 2. The target compounds were obtained in good yields via Suzuki coupling of **Thy2** with various boronic acids.

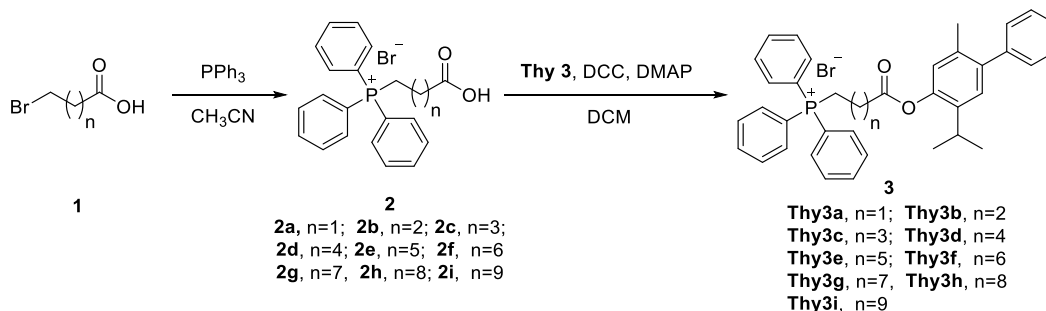


	R ₂		R ₂		R ₂
Thy5	2'-OCH ₃	Thy13	2'-F	Thy21	4'-COOH
Thy6	3'-OCH ₃	Thy14	3'-F	Thy22	2'-CN
Thy7	4'-OCH ₃	Thy15	4'-F	Thy23	3'-CN
Thy8	3'-OH	Thy16	2'-Cl	Thy24	4'-CN
Thy9	4'-OH	Thy17	3'-Cl	Thy25	2'-CF ₃
Thy10	2'-CH ₃	Thy18	4'-Cl	Thy26	3'-CF ₃
Thy11	3'-CH ₃	Thy19	2'-COOH	Thy27	4'-CF ₃
Thy12	4'-CH ₃	Thy20	3'-COOH		

Scheme 2 Synthesis of **Thy5-Thy27**

The synthesis of TPP-conjugates was started with bromoalkyl acid by reacting with PPh₃ in acetonitrile to obtain (carboxyalkyl) triphenyl-phosphonium bromides, which then reacted with the corresponding Thymol derivative **Thy3** to afford the TPP-conjugates via Steglich Esterification (Scheme 3).

The general synthetic route for these compounds is presented in Scheme 1-3. All spectral and analytical data of the newly synthesized compounds were in full agreement with the proposed structures (Extended Data Fig 1).



Scheme 3 Synthesis of TPP-conjugates **Thy3a-Thy3i**

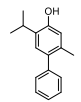
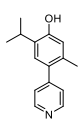
2.2 In vitro antibacterial activity of Thymol derivatives and TPP-conjugates

The ESKAPE bacteria are priority pathogens to fight against[24]. In our study, several Gram-positive bacteria (*S. aureus* ATCC 43300, MRSA USA 300, MRSA 3390) and Gram-negative bacteria (*E. coli* ATCC 25922, *E. coli* 53328, *P. aeruginosa* ATCC 27853, *P. aeruginosa* 535, *K. pneumonia* ATCC 700603, *K. pneumonia* 3026, *A. baumannii* ATCC 19606 and *A. baumannii* 1318) were selected as test bacteria in our studies. Thymol derivatives **Thy1-4**, **Thy5-27**, and **Thy3a-3i**

were first evaluated for their *in vitro* antibacterial activity against these bacteria through the double dilution method, using ciprofloxacin as the positive control (Table 1-3 and Extended Data Table 1-2).

For the first series of Thymol derivatives (**Thy1**, **Thy2**, and **Thy4**), most displayed moderate to weak *in vitro* antibacterial activity against all tested Gram-positive bacteria, with only about a 4-8 fold increase in antimicrobial activity compared to Thymol (Table 1). However, **Thy3** exhibited significantly improved antibacterial activity against *S. aureus* (MIC = 8-16 µg/mL), which was about 8-32 times more potent than natural Thymol, and thus was selected for further optimization. However, **Thy1-4** had no detectable antimicrobial activity against Gram-negative bacteria at concentrations up to 64 µg/mL (Extended Data Table 1).

Table 1 Antibacterial activity of Thymol derivatives (**Thy1-4**) against Gram-positive strains

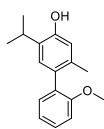
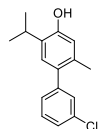
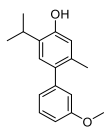
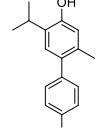
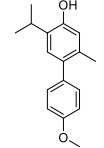
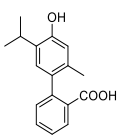
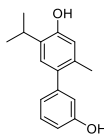
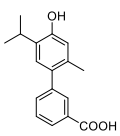
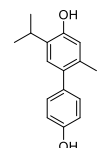
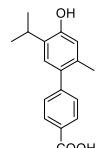
Comp.	Structure	MIC (µg/mL)		
		<i>S. a</i> ATCC 330041	MRSA USA 300	MRSA 3390
Thy		128	128	256
Thy1		64	64	64
Thy2		64	64	32
Thy3		16	8	16
Thy4		128	128	128
Cip	-	2	1	1

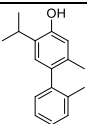
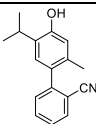
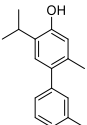
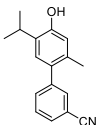
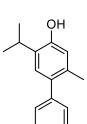
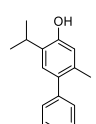
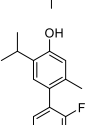
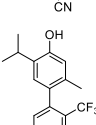
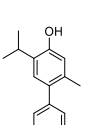
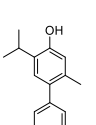
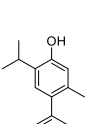
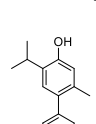
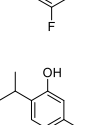
Thy=Thymol, MIC = Minimum inhibitory concentration. Cip = Ciprofloxacin. *S. a* = *S. aureus*.

Starting from **Thy3**, we further designed and synthesized the second series of Thymol derivatives (Scheme 2) and evaluated their antibacterial activities (Table 2 and Extended Data Table 2). Most of these derivatives maintained good antibacterial activity against Gram-positive *S. aureus* and MRSA, with MIC values less than 16 µg/mL except for **Thy5** and **Thy9**. Compared with **Thy3**, some compounds, such as **Thy25-27**, showed slightly improved antibacterial activity (Table 2). However, similar to **Thy3**, none of these compounds displayed antibacterial activity against Gram-

negative bacteria (Extended Data Table 2). SAR analysis suggested that electron-donating substituents on the phenyl group (**Thy5-12**) led to reduced activity in comparison with **Thy3**, while a methyl substituent at the C-2 or C-4 position of the phenyl ring (**Thy10**, **Thy12**) exhibited similar antibacterial activity (8 µg/mL). In contrast, derivatives with electron-withdrawing groups on the phenyl ring (**Thy13-27**) displayed a 7-30 fold improvement over Thymol, which is similar to or better than **Thy3**. Compounds with 3'-F (**Thy14**), 2'-Cl (**Thy17**), and 3'-COOH (**Thy20**) on the phenyl ring were similar to **Thy3** in terms of their antimicrobial activity, and derivatives with 3'-Cl (**Thy17**) and CF₃ (**Thy25-27**) on the phenyl ring showed a 2-fold improvement over **Thy3**. Introduction of Cl to the phenyl ring (**Thy16-18**), their antibacterial activity was in the order of meta > para > ortho. However, the introduction of CF₃ on the phenyl ring had a similar effect on their antibacterial activities, regardless of its position on the phenyl ring (**Thy25-27**).

Table 2 Antibacterial activity of Thymol derivatives (**Thy5-27**) against Gram-positive strains

Comp.	Structure	MIC (µg/mL)			Comp.	Structure	MIC (µg/mL)		
		<i>S. a</i>	MRSA	MRS			<i>S. a</i>	MRSA	MRS
		ATCC 330041	USA 300	A 3390			ATCC 330041	USA 300	A 3390
Thy5		>64	>64	>64	Thy17		4	8	8
Thy6		16	16	16	Thy18		8	4	8
Thy7		16	32	16	Thy19		16	8	8
Thy8		32	32	32	Thy20		8	8	16
Thy9		>64	>64	>64	Thy21		16	16	8

Thy10		8	8	16	Thy22		16	8	8
Thy11		16	8	8	Thy23		32	16	8
Thy12		8	16	8	Thy24		16	16	32
Thy13		8	16	16	Thy25		8	8	16
Thy14		16	16	16	Thy26		8	4	8
Thy15		16	16	16	Thy27		4	4	8
Thy16		8	8	8	Cip	-	0.5	1	2

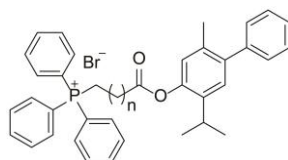
149 MIC = Minimum inhibitory concentration. MRSA = methicillin-resistant *S. aureus*. Cip = Ciprofloxacin. *S. a* = *S.*
150 *aureus*.

151 It has been reported that TPP⁺ conjugated with a cationic charge could interact with negatively
152 charged cell membranes more easily through electrostatic interaction, thus affecting bacterial cell
153 permeability. For example, TPP⁺-CL conjugates were membrane-targeting compounds, which can
154 disrupt bacterial cell membrane integrity, thus with enhanced antibacterial efficacies to combat the
155 CL resistance strains[13]. We hypothesized that the conjugation of TPP with Thymol derivatives
156 might enhance the permeability of these derivatives across bacterial membranes, leading to
157 improved antibacterial potency and broad antibacterial ranges. Thus, we designed and synthesized
158 a series of TPP conjugates linking **Thy3** to the lipophilic cation (carboxyalkyl)
159 triphenylphosphonium with varying linker chain lengths (Scheme 3). Antibacterial evaluations
160 (Table 3) identified TPP conjugates (**Thy3a-3i**) as the most potent compounds against MRSA with

MICs lower than 2 µg/mL, which are 4-16 folds more potent than **Thy3**, about 100 folds more potent than natural Thymol. The antibacterial activities of **Thy3a-3i** against MRSA were essentially the same (MIC = 0.5-2 µg/mL), regardless of the chain length. Furthermore, **Thy3d-3h** displayed good inhibitory activity against all tested Gram-negative bacterial strains except for *P. aeruginosa*. Unlike the report by Khailova., et al[25], the antibacterial activity of the Thymol derivatives did not correlate with the chain length. When the chain length is n = 4-8, it exerts strong and broad-spectrum antibacterial activities.

In our studies, the MIC value of Thymol against *S. aureus* was about 100-200 µg/mL (Table 1), and no antibacterial activity was detected against Gram-negative bacteria (Extended Data Table 1). After modification of Thymol, **Thy3-TPP** series compounds showed similar activity to the positive control ciprofloxacin, displaying excellent antibacterial activity against Gram-positive bacteria, with MIC at about 1 µg/mL (Table 3). Compared with Thymol, antibacterial activity was increased more than 100 times. Unlike Thymol, most **Thy3-TPP** compounds also showed good antibacterial effects against multidrug-resistant Gram-negative bacteria (Table 3). The above results demonstrated that, the antibacterial ability of compounds can be significantly improved after conjugation with the TPP⁺ group, providing a strategy for the design and modification of compounds in the future.

Table 3 Antibacterial activity of TPP-Thymol conjugates (**Thy3a-3i**)



		MIC (µg/mL)										
		Gram-positive bacteria				Gram-negative bacteria						
Comp.	n	<i>S. a</i>	MRS		<i>E. coli</i>	<i>E.</i>	<i>P. a</i>		<i>K. p</i>		<i>A. b</i>	
		ATCC	A	MRS	ATC	<i>coli</i>	ATC	<i>P. a</i>	ATC	<i>K. p</i>	ATC	<i>A. b</i>
		33004	USA	A	C	5332	C	535	C	302	C	131
		1	300	3390	25922	8	27853		70060	6	1960	8
Thy3a	1	1	1	0.5	4	>64	>64	>64	32	>64	32	16
Thy3b	2	2	2	1	16	16	64	>64	16	32	8	8
Thy3c	3	1	1	1	16	8	>64	>64	32	32	16	8
Thy3d	4	1	0.5	1	8	8	32	64	8	16	8	8
Thy3e	5	1	1	1	8	8	>64	>64	8	8	8	8
Thy3f	6	2	0.5	1	4	8	32	32	16	16	8	16

Thy3g	7	1	2	1	4	16	64	32	64	64	16	8
Thy3h	8	2	1	1	4	64	32	32	32	>64	32	16
Thy3i	9	2	2	1	>64	32	64	>64	>64	>64	>64	>64
Cip	-	0.5	1	2	0.06	0.12	1	1	0.12	0.06	0.5	1

MIC = Minimum inhibitory concentration. MRSA = methicillin-resistant *S. aureus*. Cip = Ciprofloxacin, *S. a* = *S.*

aureus. *P. a* = *P. aeruginosa*. *K. p* = *K. pneumonia*. *A. b* = *A. baumannii*

2.3 *In vitro* cytotoxicity study

TPP-conjugated compounds with the ability to destroy cell membranes usually have a small therapeutic window, most likely due to their non-specific cell membrane damage to bacteria and normal cells, which is a limiting factor in application potential as antibiotics[26]. We evaluated the cytotoxicity of **Thy3** and **Thy3a-Thy3i** against L02 cells (normal human liver-02) by MTT assay to evaluate their *in vitro* safety. As shown in Table 4 and Extended Data Fig 2, the cytotoxicity of these conjugates varied significantly. Conjugates **Thy3a**, **Thy3b**, **Thy3c**, **Thy3h**, and **Thy3i** showed higher toxicity than parent Thymol derivative **Thy3**, while the rest displayed reduced toxicity. **Thy3d** showed significantly lower toxicity than **Thy3**, with an $IC_{50} = 34.26 \mu\text{g/mL}$ (Table 4), representing the largest toxicity-activity window among the conjugates, and thus was selected to explore its antibacterial mechanism.

Table 4 IC_{50} of **Thy3**, and **Thy3a-Thy3i**

Compounds	Thy3	Thy3a	Thy3b	Thy3c	Thy3d	Thy3e	Thy3f	Thy3g	Thy3h	Thy3i
IC_{50} ($\mu\text{g/mL}$)	13.46	7.96	5.04	8.31	34.26	18.82	21.13	13.16	8.67	3.13

2.4 Time kill kinetics of **Thy3d**

To determine the conditions for further biological evaluations, a kill-time assay was carried out with the compound **Thy3d** (Fig 1A). At a concentration of $2 \times \text{MIC}$, a $\geq 2 \log_{10}$ reduction against MRSA was observed in 4 hours. However, the initial killing was not sustained, with bacterial re-growth after 24 h. Application of a higher concentration ($4 \times \text{MIC}$) resulted in a $\geq 2 \log_{10}$ reduction in 1 hour. Furthermore, the anti-bactericidal activity of **Thy3d** at $4 \times \text{MIC}$ was sustained over 24 hours. At the highest concentration tested ($8 \times \text{MIC}$), bacteria were eliminated after treatment for 1 hour, and the bactericidal activity was sustained for 24 hours. Ciprofloxacin inhibited bacterial growth at a lower concentration with an MIC ($2 \mu\text{g/mL}$) for a short time. Higher concentrations (4-

8 × MIC) of ciprofloxacin resulted in sustained bactericidal activity against MRSA for 24 hours, which was similar to that of **Thy3d**. Taken together, **Thy3d** exhibited a rapid bactericidal effect, killing MRSA in a concentration- and time-dependent manner.

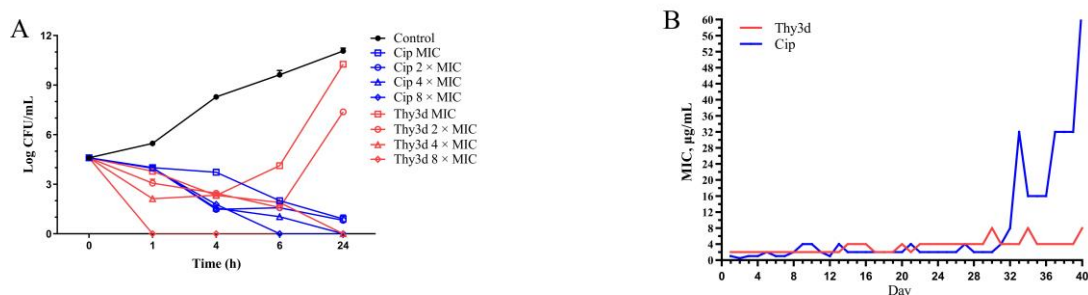


Fig 1 Time-kill kinetic, and bacterial resistance of **Thy3d**. (A) Time-kill assay of **Thy3d** against MRSA 3390. Data are mean ± SD for n = 3 biologically independent samples. (B) MRSA 3390 evolution under **Thy3d** and ciprofloxacin stress over 40 days of serial passages. Cip = Ciprofloxacin.

2.5 Resistance development study

The development of resistance is a great challenge that hampers antibiotic development and the eradication of nosocomial pathogens. We evaluated the selection of resistance through serial passaging in the presence of sub-inhibitory concentrations of **Thy3d** or ciprofloxacin up to 40 passages (Fig 1B). With the exposure to ciprofloxacin, it showed no significant change in the first 30 generations, but then exhibited obvious resistance, with a 100-fold increase in MIC after 40 passages. However, no significant change was observed for **Thy3d**, with only a 4-fold change after 40 passages. The above results indicate a low tendency for bacterial resistance with **Thy3d**. Since cell membrane synthesis involves complex biochemical processes, membrane-targeted antibacterial compounds generally have a low drug resistance[27]. We speculate that **Thy3d** delays the emergence of resistance possibly due to the membrane-targeting function of the TPP moiety.

2.6 Effect of **Thy3d** on the bacterial cell membrane

Thy3d contains a TPP group, which was reported to disrupt cell membranes[10, 28]. We thus explored whether **Thy3d** exerted its antibacterial effect by disrupting the bacterial cell membrane as we designed.

Consistent with our hypothesis, **Thy3d** caused DNA leakage, membrane permeabilization, and membrane depolarization in MRSA 3390 (Fig 2). Since nucleic acid leakage can be used as an index

of membrane permeability, we measured the absorption of **Thy3d** at 260 nm (Fig 2A)[29]. For all treatment groups, the absorption at 260 nm reached the peak after 4 hours and remained at a similar level after that. The absorption intensity at 260 nm increased with time and concentration compared with the control group. This result showed that **Thy3d** can disrupt cell membrane. Fluorescent probes SYTOXTM and DiSC₃(5) were used to detect membrane integrity and membrane potential changes of bacteria, respectively. The fluorescence intensity of SYTOXTM reached the peak within 5 min and remained at this level at the end of our test (Fig 2B). As shown in Fig 2C, the treated groups at MIC, 2 × MIC, 4 × MIC, and 8 × MIC showed induction of fluorescence compared to the control group, and all treated groups reached a steady state after 10 min (monitored by DiSC₃(5) assays uptake). The results indicate that the permeabilization and depolarization of the cell membrane of MRSA by the **Thy3d** was time- and dose-dependent.

SEM was used to further investigate the change in cellular morphology after membrane damage caused by **Thy3d**. A comparison of treated versus untreated MRSA 3390 reveals a significant difference in membrane morphology. In the absence of **Thy3d**, the bacteria appeared normal with intact and smooth surfaces. However, **Thy3d** destroyed the plasma membrane, and the tested bacteria showed rough, shrunken, and wrinkled surfaces (Fig 2D). These results support that **Thy3d** exerts its bactericidal effect by damaging the bacterial cell membrane.

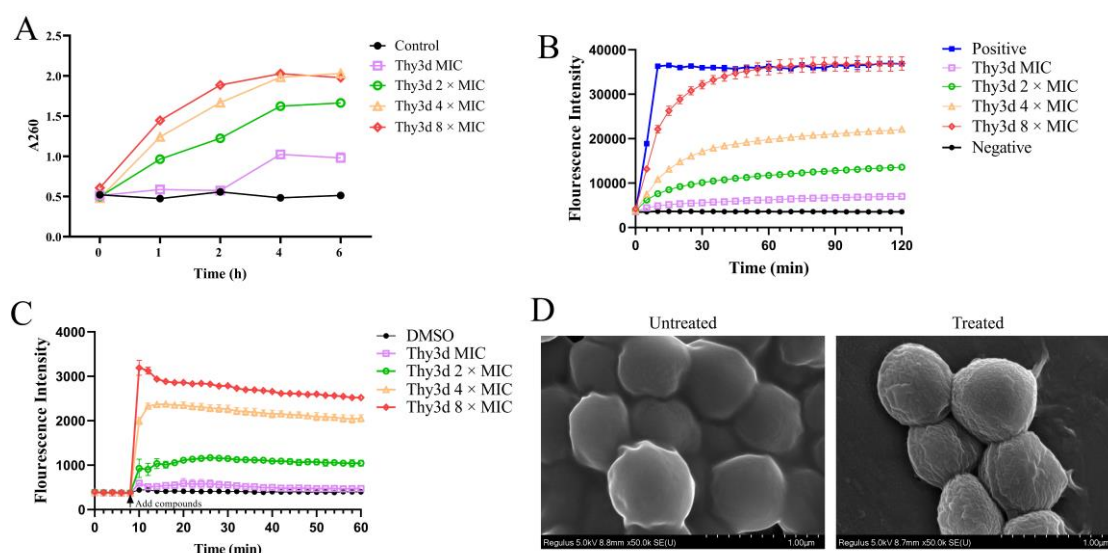


Fig 2 The mode of action of **Thy3d** on MRSA 3390. (A) Leakage of 260 nm absorbing material of MRSA 3390 treated with different concentrations of **Thy3d**. (B) Membrane permeabilization assay in MRSA 3390 monitoring uptake of SYTOXTM Green of **Thy3d**. (C) Membrane depolarization assay in MRSA 3390 monitoring uptake of

DiSC₃(5) of **Thy3d**. (A), (B), and (C), data are mean $\pm$ SD for n = 3 biologically independent samples. (D) Scanning electron micrographs showing bacterial cell membrane structures in 4 $\times$ MIC **Thy3d** treated (right) cells and DMSO control (left). Scale bars, 1 μ m.

2.7 Mechanism of **Thy3d** with biomembranes

All the above studies indicate that **Thy3d** is a membrane-active compound. However, its mechanism of interaction with the membrane is still unknown. To elucidate the molecular interactions between **Thy3d** and the bacterial cell membrane lipid bilayers, we employed molecular dynamics (MD) simulations to investigate the effects of **Thy3d** binding on the lipid bilayer organization[30], and electrical recording experiments to elucidate its mode of action in *S. aureus*[31].

As shown in Fig 3A, after simulation of the trajectory, **Thy3d** prefers to stay at the head group region, with the polar moiety interacting with the head groups and the hydrophobic moiety inserting into the lipid tail region of the membrane. The perturbation of the membrane-water interface destabilizes the membrane. With the extension of simulation time, the membrane interaction of **Thy3d** remained essentially unchanged. In addition, we also simulated the interaction of **Thy3d** with the mammalian membrane (Extended Data Fig 3), the mechanism of action is similar to that of the bacterial cell membrane. However, while **Thy3d** can penetrate either bacterial membrane or mammalian membrane, it penetrates deeper into the bacterial membrane (Fig 3B). Furthermore, we compared the free energy of **Thy3d** on two different cell membranes (Fig 3C). Free energy simulations suggest that **Thy3d** is more favorable to penetrate the bacterial membrane and experiences less energy barrier to penetrate bacterial membrane than mammalian membrane. This result is consistent with its strong antibacterial ability (Table 3) and weak cytotoxicity (Table 4) of **Thy3d**. The MD simulation results suggest that **Thy3d** may enhance the structural fluctuation within the water-lipid interface, and exert bacterial membrane damage. However, due to energy preference, mammalian cell membranes are less affected and therefore exhibit weak cytotoxicity.

To validate the MD simulation results, we carried out a series of experiments to explore the interactions between **Thy3d** and the lipid bilayer. 12 μ g/mL of **Thy3d** was added and stirred in the chamber, after the formation of a lipid bilayer in an orifice on a Teflon film[31]. The ionic current began to fluctuate significantly around 4 min (Figure 3D), indicating that due to the strong

interaction, **Thy3d** started to breach the membrane. During the next 4 min, the ionic current rapidly went up to a peak value (Fig 3D), suggesting that the continuous lipid extraction had passed the critical point ($t \approx 4.7$ min) and the pore formation in the lipid bilayer became possible. Applying an equal volume of DMSO, the ionic current remained unchanged at 0 pA (Extended Data Fig 4), suggesting a stable and unbroken lipid bilayer. Overall, these experimental observations are consistent with the simulations that **Thy3d** exerts its antibacterial ability mainly by infiltrating and penetrating cell membrane lipid bilayers.

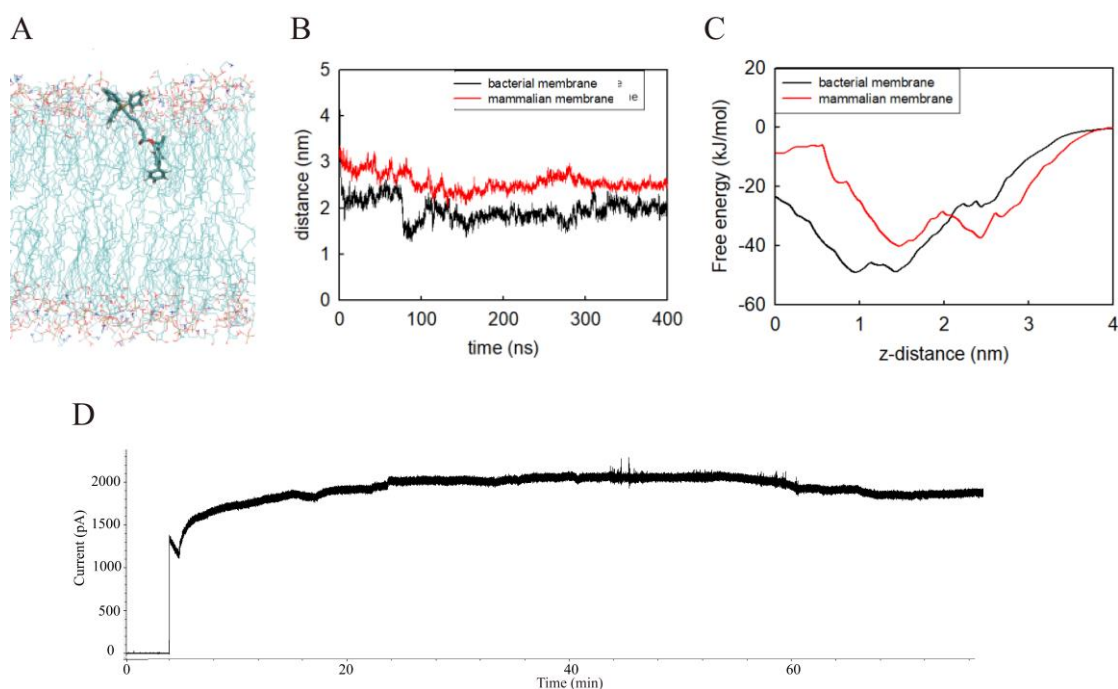


Fig 3 Studies on the interactions between **Thy3d** and the lipid bilayer. (A) Representative configurations of molecular dynamics simulations of **Thy3d** in bacterial membrane. (B) Distance of lipid in the bilayers in the bacterial or mammalian membrane in the MD simulations. (C) Free energy of **Thy3d** adsorption from the aqueous phase to the membrane phase constructed by umbrella sampling. The distance between the bilayer center and the center of mass of the compounds (z-distance) was used as the reaction coordinate. All simulations were repeated three times. (D) Experimental studies on interactions between **Thy3d** and the lipid bilayer. The experiment was performed at +100 mV with the lipid bilayer formed on a Teflon film. The electrolyte solution contains 1 M KCl, 10 mM Tris, and 1 mM EDTA (pH 7.6). $n = 2$ biologically independent experiments.

2.8 Eradicating MRSA persisters by **Thy3d**

Due to the dormant state of persisters, it can tolerate lethal concentrations of antibacterial drugs,

which help bacteria evade antibiotic treatment[32]. Previous studies suggest that small molecules capable of destroying bacterial cell membranes might be effective against bacterial persisters and in overcoming bacterial resistance[3, 33, 34]. **Thy3d** demonstrated potent inhibitory activity against MRSA, as well as targeting the cell membrane. We speculated that **Thy3d**, which destroys normal bacterial cell membranes, might also remove persisters. Thus its ability to eradicate MRSA persistence was further evaluated.

When treated with $4-8 \times \text{MIC}$ of **Thy3d** ($4-8 \mu\text{g/mL}$), the number of persisters decreased sharply over time (Fig 4A). **Thy3d** at its $4 \times \text{MIC}$ and $8 \times \text{MIC}$ elicited a rapid killing effect of 4 and 5 log units (99.99% and 99.999%). Furthermore, the rapid killing of MRSA persisters with $4 \times$ and $8 \times \text{MIC}$ of **Thy3d** was sustained for 7 hours, indicating that **Thy3d** is very effective in killing persister cells.

The killing effect of **Thy3d** on the persisters may be due to its rapid penetration of MRSA-persister membranes. When treated with $4-8 \times \text{MIC}$ **Thy3d** ($4-8 \mu\text{g/mL}$) against MRSA persister, the membrane permeability (Fig 4B) and membrane potential (Fig 4C) of the persisters changed significantly in a short time, reaching the peak within 30 min and 10 min respectively, and then maintaining the high fluorescence intensity. These results indicate that **Thy3d** can rapidly disrupt the cell membrane of the persister at high concentrations, causing DNA leakage (Fig 4D), which plays an important role in eliminating the persisters. MD experiments suggest that **Thy3d** can penetrate the membrane more efficiently owing to the lower energy barriers, which are more favorable for transferring energies (Fig 3B). These results are consistent with the observation that **Thy3d** is effective against persister cells. In the above exploration of the persistent bacteria, we found that the destruction effect of **Thy3d** on normal bacterial cell membranes also extended to the persistent bacteria, exerting the same membrane-breaking ability on persister membranes, effectively removing the persisters.

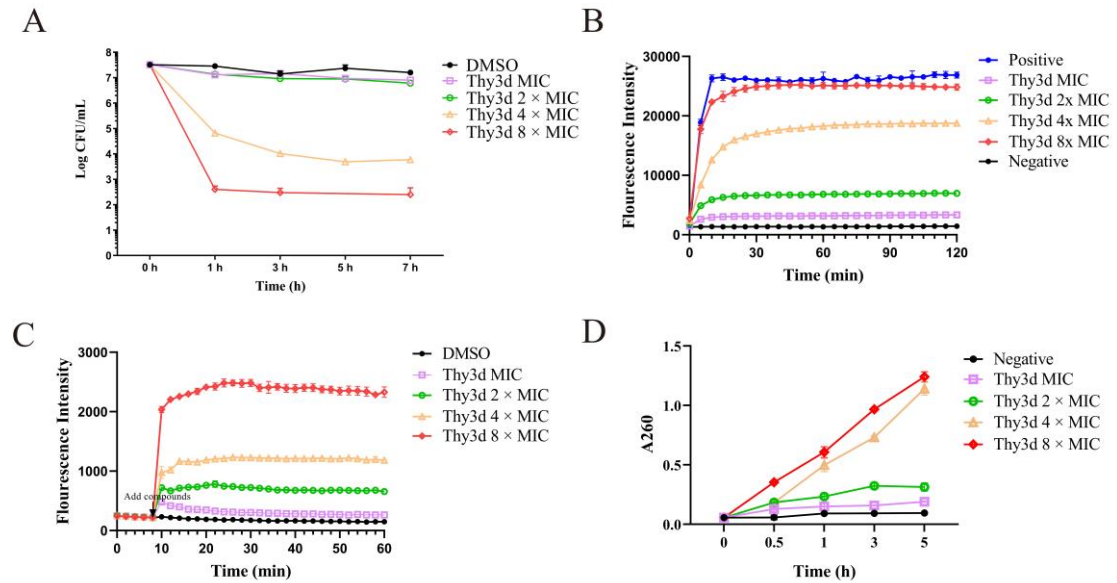


Fig 4 The killing effect of **Thy3d** on persister of MRSA. (A) Eradication of persister by **Thy3d**. (B) Membrane permeabilization assay for persisters, monitoring uptake of SYTOX™ Green of **Thy3d**. (C) Membrane depolarization assay for persisters, monitoring uptake of DiSC₃(5) of **Thy3d**. (D) Leakage of 260 nm absorbing material of persister cells treated with different concentrations of **Thy3d**. (A), (B), (C), and (D), data are mean ± SD for n = 3 biologically independent experiments. The strain is MRSA 3390.

2.9 In vivo efficacy of **Thy3d** in mouse subcutaneous infection model

We next established bioluminescent *S. aureus* Xen29 infection in a mouse topical infection model and evaluated the potential therapeutic application of **Thy3d** in the model.

In the saline group and **Thy3d** treated group, the fluorescence intensity of *S. aureus* Xen29 showed a dynamic pattern. There was a gradual increase in the early stage of infection and then a gradual decrease. Throughout the experiment, the fluorescence intensity of the **Thy3d** group and infection group began to appear different at 24 h, and **Thy3d** led to a significant reduction of fluorescence intensity ($p = 0.0155$ for **Thy3d** vs. vehicle) until 84 h (Fig 5A and 5B), and wound scabbing was noticeably improved. The number of colonies at infection was also reduced by about 10 times in the **Thy3d** group compared to the saline group (Fig 5C). Though **Thy3d** was not as effective as the positive control ($P = 0.0176$ for vancomycin vs. vehicle) (Fig 5). These results suggest that **Thy3d** has a certain influence on the development of subcutaneous infection. The weight loss exhibited by the mice during infection was significantly reduced when compared with saline controls and positive controls, suggesting a favorable safety property (Extended Data Fig 5).

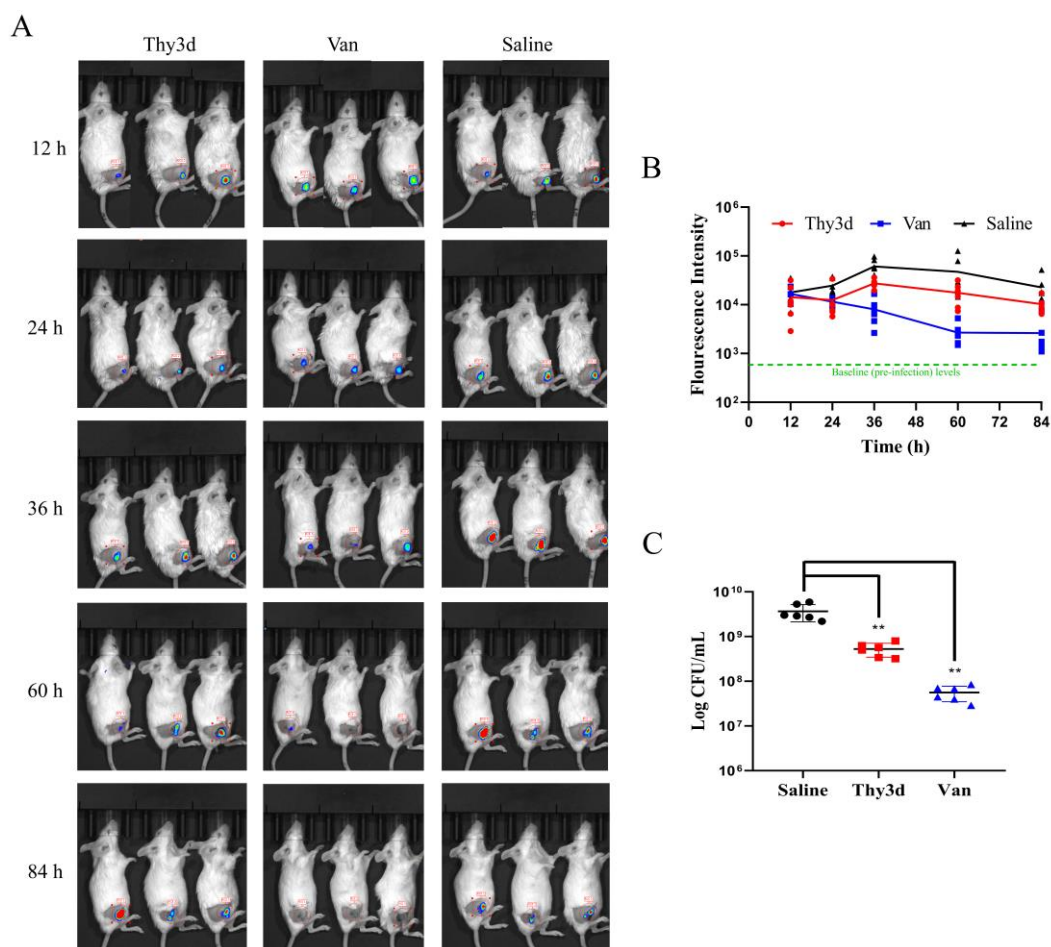


Fig 5 *In vivo* efficacy studies of **Thy3d** in topical skin infection model. (A) Animals were imaged using the Lumina II system (Perkin-Elmer) at 12, 24, 36, 60, and 84 h post-infection (only 3 mice shown per image) and (B) bioluminescence (photons per second) was determined. (C) At 84 h after the first treatment and infection, animals were sacrificed and samples of skin from each animal were homogenized and plated on nutrient agar to determine the CFUs per animal. Errors are mean $\pm$ SEM, $n = 6$ animals. Statistical differences between control and antibiotic treatment groups were analyzed by t-tests ($*p < 0.05$) compared with the indicative group.

3. Conclusions

A series of novel antibiotics was designed and synthesized based on widely applied natural Thymol, whose antibacterial activity was improved by nearly 100 times. These derivatives exhibited a significantly broader antibacterial spectrum, with potent activities against several ESKAPE pathogens.

The TPP-Thymol derivatives exert their antibacterial property by targeting bacterial membranes, displaying fast killing and low probability of resistance development. In particular, **Thy3d** effectively eradicates MRSA persister cells. Mechanistic studies suggested that **Thy3d** could

enhance the structural fluctuation within the water-lipid interface, infiltrate and penetrate lipid bilayers, and disrupt bacterial cell membranes, directly killing normal as well as persistent bacteria. Meanwhile, due to its free-energy preference, **Thy3d** has a higher affinity for bacterial cell membranes, thus **Thy3d** exhibited low toxicity for normal cells while maintaining potent activities for both normal and persister bacteria. While previous studies on TPP moiety mainly focused on anti-tumor activity[10, 35], our results demonstrate TPP conjugates could be designed to discover novel antibiotics against Gram-positive, Gram-negative as well as persister bacteria. Our studies support that natural products are still an important source for novel antibiotic discovery, and our work provides a new strategy to fight the threatening antibiotic resistance, particularly those caused by bacterial persisters which have no effective treatment to date.

4. Materials and methods

4.1 Synthesis

4.1.1 General remarks

All chemicals were purchased from a commercial source and used without further purification unless otherwise stated. The reactions were monitored by thin-layer chromatography. Column chromatography was performed on silica gel. ¹H NMR spectra were obtained on an Agilent 400 MR spectrometer, while ¹³C NMR spectra were obtained with proton decoupling on an Agilent 400 MR DD2 (100 MHz) or 600 MR DD2 spectrometer and were reported in ppm with residual solvent for internal standard (δ 77.16 (CDCl₃)). The chemical shifts were reported in parts per million (ppm), the coupling constants (*J*) were expressed in hertz (Hz) and signals were described as singlet (s), doublet (d), triplet (t), as well as multiplet (m). The NMR data was analyzed by MestReNova software. High-resolution mass spectra were obtained on a Bruker Solarix 7.0 T spectrometer. The purity of all test compounds was determined to be >95% by HPLC. All other chemicals and solvents were commercially available and were used as received.

4.1.2 4-Chloro-2-isopropyl-5-methylphenol (Thy1)

To a solution of Thy (1 eq.) in DCM, was added a solution of SO₂Cl₂ (1.2 eq.) in DCM at 0 °C, and the mixture was stirred at room temperature for 2 h. After the reaction completion, the solvent was concentrated under reduced pressure, the crude residue was purified by silica gel

chromatography to afford **Thy1**. ¹H NMR (600 MHz, CDCl₃) δ 7.15 (s, 1H), 6.62 (s, 1H), 4.82 (bs, 1H), 3.17 – 3.12 (m, 1H), 2.28 (s, 3H), 1.24 (d, *J* = 6.6 Hz, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 151.22, 134.11, 133.85, 126.97, 126.00, 117.78, 26.99, 22.61, 19.68. HRMS (ESI) *m/z*: calcd. for C₁₀H₁₂ClO⁺, [M-H]⁺, 183.0582. Found 183.0581.

4.1.3 4-Bromo-2-isopropyl-5-methylphenol (**Thy2**)

To a solution of **Thy** (1 eq.) in MeOH: DCM (2:3 vol.), was added tetrabutylammonium tribromide (1.0 eq.) at RT, and the mixture was stirred for 3 h. After the reaction completion, the solvent was concentrated under reduced pressure, the crude residue was purified by silica gel chromatography to afford **Thy2**. ¹H NMR (600 MHz, CDCl₃) δ 7.29 (s, 1H), 6.64 (s, 1H), 4.72 (s, 1H), 3.15 – 3.10 (m, 1H), 2.30 (s, 3H), 1.23 (d, *J* = 6.9 Hz, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 151.97, 136.01, 134.22, 130.19, 117.78, 115.74, 26.99, 22.63, 22.53. HRMS (ESI) *m/z*: calcd. for C₁₀H₁₂BrO⁺, [M-H]⁺, 227.0077. Found 227.0075.

4.1.4 General procedure for the preparation of **Thy3-Thy27**

To a solution of **Thy2** (1 eq.) and the respective boronic acid (2.5 eq.) in water:1,4-dioxane (4:1), was added sodium bicarbonate (10 eq.), degassed with nitrogen for 10 min, then added Pd(PPh₃)₄ (0.1 eq.), and the mixture was stirred at 80 °C overnight. After the reaction completion, the reaction mixture was diluted with water, and extracted with EtOAc twice, the organic layer was washed with brine, dried with anhydrous Na₂SO₄, and concentrated under reduced pressure to get a crude compound. The crude product was purified by silica gel chromatography.

5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-ol (**Thy3**): ¹H NMR (600 MHz, CDCl₃) δ 9.20 (s, 1H), 7.36 (t, *J* = 7.5 Hz, 2H), 7.25 (t, *J* = 5.2 Hz, 3H), 6.88 (s, 1H), 6.67 (s, 1H), 3.25 – 3.14 (m, 1H), 2.10 (s, 3H), 1.14 (d, *J* = 6.9 Hz, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 151.94, 142.10, 134.86, 133.92, 131.81, 129.59, 128.21, 128.14, 126.55, 117.07, 26.97, 22.83, 20.15. HRMS (ESI) *m/z*: calcd. for C₁₆H₁₇O⁺, [M-H]⁺, 225.1285. Found 225.1281.

2-Isopropyl-5-methyl-4-(pyridin-4-yl) phenol (**Thy4**): ¹H NMR (400 MHz, DMSO- *d*₆) δ 9.46 (s, 1H), 8.56 (d, *J* = 6.0 Hz, 1H), 7.32 (d, *J* = 6.1 Hz, 1H), 6.97 (s, 1H), 6.72 (s, 1H), 3.21-3.10 (m, 1H), 2.17 (s, 3H), 1.16 (d, *J* = 6.9 Hz, 6H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 154.57, 149.33, 149.23, 132.73, 132.25, 129.27, 127.19, 124.27, 116.98, 116.93, 26.16, 22.47, 22.47, 19.79, 19.76. HRMS

410 (ESI) m/z : calcd. for $C_{15}H_{16}NO^-$, $[M-H]^-$, 226.1237. Found 226.1234.
 411 5-Isopropyl-2'-methoxy-2-methyl-[1,1'-biphenyl]-4-ol (**Thy5**): 1H NMR (400 MHz, $CDCl_3$) δ 7.34
 412 (m, 1H), 7.17 (dd, $J = 7.3, 1.9$ Hz, 1H), 7.05 – 6.95 (m, 3H), 6.66 (s, 1H), 4.82 (s, 1H), 3.79 (s, 3H),
 413 3.25 – 3.14 (m, 1H), 2.08 (s, 3H), 1.27 (d, $J = 6.8$ Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 156.92,
 414 151.99, 135.38, 131.63, 131.42, 131.22, 130.94, 128.42, 120.52, 116.55, 116.51, 110.72, 55.53,
 415 55.48, 26.94, 22.83, 19.63, 19.62. HRMS (ESI) m/z : calcd. for $C_{17}H_{19}O_2^-$, $[M-H]^-$, 255.1391. Found
 416 255.1385.
 417 5-Isopropyl-3'-methoxy-2-methyl-[1,1'-biphenyl]-4-ol (**Thy6**): 1H NMR (400 MHz, $CDCl_3$) δ 7.25
 418 – 7.22 (m, 2H), 7.05 (s, 1H), 6.97 – 6.93 (m, 2H), 6.66 (s, 1H), 4.83 (s, 1H), 3.86 (s, 3H), 3.25 –
 419 3.15 (m, 1H), 2.20 (s, 3H), 1.27 (d, $J = 6.8$ Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 158.34, 151.77,
 420 134.54, 134.39, 134.00, 131.80, 130.58, 128.22, 117.04, 117.01, 113.55, 55.46, 26.94, 22.84, 20.22.
 421 HRMS (ESI) m/z : calcd. for $C_{17}H_{19}O_2^-$, $[M-H]^-$, 255.1391. Found 255.1387.
 422 5-Isopropyl-4'-methoxy-2-methyl-[1,1'-biphenyl]-4-ol (**Thy7**): 1H NMR (600 MHz, $CDCl_3$) δ 7.23
 423 (d, $J = 8.2$ Hz, 2H), 7.04 (s, 1H), 6.94 (d, $J = 8.1$ Hz, 2H), 6.66 (s, 1H), 4.77 (s, 1H), 3.85 (s, 3H),
 424 3.22 – 3.15 (m, 1H), 2.20 (s, 3H), 1.26 (d, $J = 6.7$ Hz, 6H). ^{13}C NMR (151 MHz, $CDCl_3$) δ 158.32,
 425 151.81, 134.63, 134.38, 133.96, 131.88, 130.58, 128.21, 117.06, 113.59, 55.44, 26.95, 22.84, 20.18.
 426 HRMS (ESI) m/z : calcd. for $C_{17}H_{19}O_2^-$, $[M-H]^-$, 255.1391. Found 255.1387.
 427 5'-Isopropyl-2'-methyl-[1,1'-biphenyl]-3,4'-diol (**Thy8**): 1H NMR (400 MHz, $CDCl_3$) δ 7.24 (t, $J =$
 428 8.4 Hz, 2H), 7.03 (s, 1H), 6.86 (d, $J = 7.5$ Hz, 1H), 6.79 - 6.74 (m, 2H), 6.64 (s, 1H), 3.20 – 3.12
 429 (m, 1H), 2.19 (s, 3H), 1.24 (d, $J = 6.8$ Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 155.35, 152.03,
 430 143.84, 134.37, 133.84, 131.88, 129.30, 128.00, 122.20, 117.10, 116.60, 113.50, 29.84, 26.92, 22.83,
 431 20.10. HRMS (ESI) m/z : calcd. for $C_{16}H_{17}O_2^-$, $[M-H]^-$, 241.1234. Found 241.1231.
 432 5-Isopropyl-2-methyl-[1,1'-biphenyl]-4,4'-diol (**Thy9**): 1H NMR (400 MHz, $DMSO-d_6$) δ 9.31 (s,
 433 1H), 9.08 (s, 1H), 7.06 – 7.01 (m, 2H), 6.82 (s, 1H), 6.77 – 6.72 (m, 2H), 6.62 (s, 1H), 3.18 – 3.08
 434 (m, 1H), 2.07 (s, 3H), 1.12 (d, $J = 6.9$ Hz, 6H). ^{13}C NMR (101 MHz, $DMSO-d_6$) δ 155.78, 153.04,
 435 132.45, 132.35, 132.15, 131.53, 130.11, 127.19, 127.15, 116.55, 116.49, 114.80, 26.10, 22.62, 20.02,
 436 19.98. HRMS (ESI) m/z : calcd. for $C_{16}H_{17}O_2^-$, $[M-H]^-$, 241.1234. Found 241.1232.
 437 5-Isopropyl-2,2'-dimethyl-[1,1'-biphenyl]-4-ol (**Thy10**): 1H NMR (400 MHz, $CDCl_3$) δ 7.34 – 7.24
 438 (m, 3H), 7.21 – 7.17 (m, 1H), 7.00 (s, 1H), 6.72 (s, 1H), 5.06 (s, 1H), 3.33 – 3.22 (m, 1H), 2.14 (s,
 439 3H), 2.04 (s, 3H), 1.31 (d, $J = 6.9$ Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 151.67, 141.77, 136.45,

440 134.33, 134.23, 131.65, 130.01, 129.89, 127.56, 127.08, 125.57, 116.60, 116.57, 77.16, 26.86, 22.92,
 441 22.85, 20.08, 19.49. HRMS (ESI) m/z : calcd. for $C_{17}H_{19}O^-$, $[M-H]^-$, 239.1441. Found 239.1438.
 442 5-Isopropyl-2,3'-dimethyl-[1,1'-biphenyl]-4-ol (**Thy11**): 1H NMR (400 MHz, $CDCl_3$) δ 7.39 – 7.34
 443 (m, 1H), 7.24 - 7.18 (m, 3H), 7.14 (s, 1H), 6.72 (s, 1H), 5.12 (s, 1H), 3.34 -3.24 (m, 1H), 2.48 (s,
 444 3H), 2.28 (s, 3H), 1.35 (d, J = 6.9 Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 151.85, 142.08, 137.69,
 445 134.93, 133.83, 131.87, 130.31, 128.12, 128.00, 127.26, 126.65, 117.07, 117.05, 26.92, 22.84, 21.61,
 446 20.13. HRMS (ESI) m/z : calcd. for $C_{17}H_{19}O^-$, $[M-H]^-$, 239.1441. Found 239.1438.
 447 5-Isopropyl-2,4'-dimethyl-[1,1'-biphenyl]-4-ol (**Thy12**): 1H NMR (400 MHz, $CDCl_3$) δ 7.27 (s, 4H),
 448 7.12 (s, 1H), 6.70 (s, 1H), 4.98 (bs, 1H), 3.31 – 3.21 (m, 1H), 2.45 (s, 3H), 2.26 (s, 3H), 1.32 (d, J
 449 = 6.9 Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 151.80, 139.16, 136.13, 134.74, 133.89, 131.87,
 450 129.44, 128.85, 128.21, 117.08, 117.06, 26.92, 22.84, 21.27, 20.16. HRMS (ESI) m/z : calcd. for
 451 $C_{17}H_{19}O^-$, $[M-H]^-$, 239.1441. Found 239.1438.
 452 2'-Fluoro-5-isopropyl-2-methyl-[1,1'-biphenyl]-4-ol (**Thy13**): 1H NMR (400 MHz, $CDCl_3$) δ 7.39
 453 – 7.28 (m, 2H), 7.26 – 7.13 (m, 2H), 7.10 (s, 1H), 6.73 (s, 1H), 5.30 (bs, 1H), 3.33 – 3.22 (m, 1H),
 454 2.18 (s, 3H), 1.32 (d, J = 7.0 Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 159.96 (d, J = 245.2 Hz),
 455 152.54, 135.21, 132.06 (d, J = 3.8 Hz), 131.92, 129.41 (d, J = 16.8 Hz), 128.80 (d, J = 8.0 Hz),
 456 128.41 (d, J = 4.5 Hz), 124.01 (d, J = 3.6 Hz), 116.84, 116.82, 115.55 (d, J = 22.8 Hz), 26.90, 22.78,
 457 19.60, 19.56. HRMS (ESI) m/z : calcd. for $C_{16}H_{16}FO^-$, $[M-H]^-$, 243.1191. Found 243.1187.
 458 3'-Fluoro-5-isopropyl-2-methyl-[1,1'-biphenyl]-4-ol (**Thy14**): 1H NMR (400 MHz, $CDCl_3$) δ 7.38
 459 – 7.30 (m, 1H), 7.08 (dd, J = 7.8, 2.8 Hz, 1H), 7.04 – 6.98 (m, 3H), 6.66 (s, 1H), 4.82 (bs, 1H), 3.25
 460 – 3.14 (m, 1H), 2.19 (s, 3H), 1.26 (d, J = 6.9 Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 162.64 (d, J
 461 = 245.5 Hz), 152.30, 133.85, 133.60 (d, J = 1.9 Hz), 132.02, 129.53 (d, J = 8.5 Hz), 128.04, 125.35
 462 (d, J = 2.8 Hz), 117.19, 117.17, 116.51 (d, J = 21.0 Hz), 113.39 (d, J = 21.0 Hz), 26.98, 22.79, 20.07.
 463 HRMS (ESI) m/z : calcd. for $C_{16}H_{16}FO^-$, $[M-H]^-$, 243.1191. Found 243.1189.
 464 4'-Fluoro-5-isopropyl-2-methyl-[1,1'-biphenyl]-4-ol (**Thy15**): 1H NMR (400 MHz, $CDCl_3$) δ 7.33
 465 – 7.26 (m, 2H), 7.16 - 7.10 (m, 2H), 7.08 (s, 1H), 6.70 (s, 1H), 5.07 (bs, 1H), 3.31 - 3.21 (m, 1H),
 466 2.21 (s, 3H), 1.31 (d, J = 7.0 Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 161.83 (d, J = 245.0 Hz),
 467 152.04, 138.00 (d, J = 3.4 Hz), 133.92, 133.81, 132.02, 131.03 (d, J = 8.0 Hz), 128.14, 117.15,
 468 117.12, 114.96 (d, J = 21.1 Hz), 26.94, 22.80, 20.08. HRMS (ESI) m/z : calcd. for $C_{16}H_{16}FO^-$, $[M-$
 469 $H]^-$, 243.1191. Found 243.1187.

470 2'-Chloro-5-isopropyl-2-methyl-[1,1'-biphenyl]-4-ol (**Thy16**): ^1H NMR (400 MHz, CDCl_3) δ 7.47
471 – 7.42 (m, 1H), 7.31 – 7.20 (m, 3H), 6.95 (s, 1H), 6.66 (s, 1H), 4.73 (s, 1H), 3.23 – 3.13 (m, 1H),
472 2.04 (s, 3H), 1.26 – 1.23 (m, 6H). ^{13}C NMR (101 MHz, CDCl_3) δ 152.32, 140.76, 134.79, 134.00,
473 132.22, 131.76, 131.63, 129.46, 128.43, 127.88, 126.61, 116.61, 26.94, 22.90, 22.69, 19.48. HRMS
474 (ESI) m/z : calcd. for $\text{C}_{16}\text{H}_{16}\text{ClO}^-$, $[\text{M}-\text{H}]^-$, 259.0895. Found 259.0889.

475 3'-Chloro-5-isopropyl-2-methyl-[1,1'-biphenyl]-4-ol (**Thy17**): ^1H NMR (400 MHz, CDCl_3) δ 7.36
476 – 7.28 (m, 3H), 7.20 (dt, $J = 6.8, 1.8$ Hz, 1H), 7.04 (s, 1H), 6.68 (s, 1H), 4.95 (bs, 1H), 3.25 – 3.16
477 (m, 1H), 2.21 (s, 3H), 1.28 (d, $J = 6.9$ Hz, 6H). ^{13}C NMR (101 MHz, CDCl_3) δ 152.31, 143.92,
478 133.92, 133.85, 133.40, 132.06, 129.60, 129.36, 128.01, 127.82, 126.65, 117.17, 117.14, 26.95,
479 22.79, 20.07. HRMS (ESI) m/z : calcd. for $\text{C}_{16}\text{H}_{16}\text{ClO}^-$, $[\text{M}-\text{H}]^-$, 259.0895. Found 259.0892.

480 4'-Chloro-5-isopropyl-2-methyl-[1,1'-biphenyl]-4-ol (**Thy18**): ^1H NMR (400 MHz, CDCl_3) δ 8.10
481 – 8.05 (m 2H), 7.59 – 7.48 (m, 2H), 7.06 (s, 1H), 6.69 (s, 1H), 3.26 – 3.15 (m, 1H), 2.20 (s, 3H),
482 1.27 (d, $J = 7.0$ Hz, 6H). ^{13}C NMR (101 MHz, CDCl_3) δ 171.92, 152.36, 142.52, 135.00, 133.92,
483 133.54, 132.12, 131.24, 129.22, 128.40, 128.12, 117.24, 27.02, 22.80, 20.11. HRMS (ESI) m/z :
484 calcd. for $\text{C}_{16}\text{H}_{16}\text{ClO}^-$, $[\text{M}-\text{H}]^-$, 259.0895. Found 259.0892.

485 4'-Hydroxy-5'-isopropyl-2'-methyl-[1,1'-biphenyl]-2-carboxylic acid (**Thy19**): ^1H NMR (400 MHz,
486 $\text{DMSO}-d_6$) δ 12.35 (bs, 1H), 9.09 (s, 1H), 7.71 (d, $J = 7.7$ Hz, 1H), 7.47 (t, $J = 7.6$ Hz, 1H), 7.37 (t,
487 $J = 7.6$ Hz, 1H), 7.16 (d, $J = 7.6$ Hz, 1H), 6.72 (s, 1H), 6.59 (s, 1H), 3.18 – 3.08 (m, 1H), 1.91 (s,
488 3H), 1.09 (d, $J = 6.8$ Hz, 6H). ^{13}C NMR (101 MHz, $\text{DMSO}-d_6$) δ 169.34, 153.31, 141.55, 132.95,
489 132.59, 131.80, 131.29, 130.80, 130.49, 128.89, 126.56, 116.02, 26.11, 22.59, 19.56. HRMS (ESI)
490 m/z : calcd. for $\text{C}_{17}\text{H}_{17}\text{O}_3^-$, $[\text{M}-\text{H}]^-$, 269.1183. Found 269.1180.

491 4'-Hydroxy-5'-isopropyl-2'-methyl-[1,1'-biphenyl]-3-carboxylic acid (**Thy20**): ^1H NMR (600 MHz,
492 CDCl_3) δ 8.06 (d, $J = 4.8$ Hz, 2H), 7.55 (d, $J = 7.6$ Hz, 1H), 7.52 – 7.48 (m, 1H), 7.05 (s, 1H), 6.69
493 (s, 1H), 3.24 – 3.17 (m, 1H), 2.20 (s, 3H), 1.27 (d, $J = 6.6$ Hz, 6H). ^{13}C NMR (151 MHz, CDCl_3) δ
494 172.27, 152.41, 142.57, 135.02, 133.90, 133.55, 132.22, 131.26, 129.29, 128.40, 128.12, 117.29,
495 27.02, 22.80, 20.08. HRMS (ESI) m/z : calcd. for $\text{C}_{17}\text{H}_{17}\text{O}_3^-$, $[\text{M}-\text{H}]^-$, 269.1183. Found 269.1178.

496 4'-Hydroxy-5'-isopropyl-2'-methyl-[1,1'-biphenyl]-4-carboxylic acid (**Thy21**): ^1H NMR (400 MHz,
497 CDCl_3) δ 8.15 (d, $J = 8.1$ Hz, 2H), 7.43 (d, $J = 8.2$ Hz, 2H), 7.07 (s, 1H), 6.69 (s, 1H), 3.24 – 3.17
498 (m, 1H), 1.27 (d, $J = 6.9$ Hz, 6H). ^{13}C NMR (151 MHz, CDCl_3) δ 172.19, 152.63, 147.99, 133.84,

499 133.59, 132.31, 130.16, 129.77, 128.03, 127.41, 117.39, 77.16, 26.99, 22.79, 20.11. HRMS (ESI)
 500 m/z : calcd. for $C_{17}H_{17}O_3^-$, $[M-H]^-$, 269.1183. Found 269.1179.

501 4'-Hydroxy-5'-isopropyl-2'-methyl-[1,1'-biphenyl]-2-carbonitrile (**Thy22**): 1H NMR (600 MHz,
 502 $CDCl_3$) δ 7.73 (d, $J = 7.7$ Hz, 1H), 7.61 (t, $J = 7.7$ Hz, 1H), 7.42 (t, $J = 7.7$ Hz, 1H), 7.38 (d, $J = 7.8$
 503 Hz, 1H), 7.02 (s, 1H), 6.68 (s, 1H), 4.98 (s, 1H), 3.23 - 3.16 (m, 1H), 2.12 (s, 3H), 1.26 (d, $J = 6.8$
 504 Hz, 6H). ^{13}C NMR (151 MHz, $CDCl_3$) δ 153.35, 146.36, 134.19, 132.96, 132.49, 132.19, 131.11,
 505 130.52, 128.03, 127.25, 118.69, 117.17, 113.20, 26.90, 22.71, 19.58. HRMS (ESI) m/z : calcd. for
 506 $C_{17}H_{16}NO^-$, $[M-H]^-$, 250.1237. Found 250.1234.

507 4'-Hydroxy-5'-isopropyl-2'-methyl-[1,1'-biphenyl]-3-carbonitrile (**Thy23**): 1H NMR (400 MHz,
 508 $CDCl_3$) δ 7.62 - 7.59 (m, 2H), 7.57 - 7.47 (m, 2H), 6.99 (s, 1H), 6.69 (s, 1H), 5.02 (bs, 1H), 3.26 -
 509 3.16 (m, 1H), 2.18 (s, 3H), 1.26 (d, $J = 6.8$ Hz, 6H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 152.81, 143.38,
 510 134.13, 133.76, 133.07, 132.44, 132.33, 130.19, 129.01, 128.02, 119.15, 117.35, 112.27, 77.16,
 511 26.96, 22.75, 20.01. HRMS (ESI) m/z : calcd. for $C_{17}H_{16}NO^-$, $[M-H]^-$, 250.1237. Found 250.1235.

512 4'-Hydroxy-5'-isopropyl-2'-methyl-[1,1'-biphenyl]-4-carbonitrile (**Thy24**): 1H NMR (400 MHz,
 513 $CDCl_3$) δ 7.69 (d, $J = 8.2$ Hz, 2H), 7.42 (d, $J = 8.1$ Hz, 2H), 7.01 (s, 1H), 6.70 (s, 1H), 5.09 (bs, 1H),
 514 3.27 - 3.17 (m, 1H), 2.20 (s, 3H), 1.26 (d, $J = 6.8$ Hz, 6H). ^{13}C NMR (151 MHz, $CDCl_3$) δ 153.05,
 515 147.19, 133.64, 132.67, 132.55, 132.00, 130.33, 127.91, 119.24, 117.44, 110.03, 77.16, 26.93, 22.73,
 516 20.01. HRMS (ESI) m/z : calcd. for $C_{17}H_{16}NO^-$, $[M-H]^-$, 250.1237. Found 250.1234.

517 5-Isopropyl-2-methyl-2'-(trifluoromethyl)-[1,1'-biphenyl]-4-ol (**Thy25**): 1H NMR (600 MHz,
 518 $CDCl_3$) δ 7.74 (d, $J = 7.9$ Hz, 1H), 7.54 (t, $J = 7.6$ Hz, 1H), 7.44 (t, $J = 7.8$ Hz, 1H), 7.25 (s, 1H),
 519 6.94 (s, 1H), 6.64 (s, 1H), 4.78 (bs, 1H), 3.22 - 3.15 (m, 1H), 1.95 (s, 3H), 1.22 (d, $J = 6.8$ Hz, 6H).
 520 ^{13}C NMR (151 MHz, $CDCl_3$) δ 154.77, 143.61, 137.03, 134.90, 134.15, 133.83, 133.44, 131.83 (q,
 521 $J = 29.2$ Hz), 130.49, 129.72, 128.60 (q, $J = 5.2$ Hz), 126.73 (q, $J = 273.9$ Hz), 118.86, 29.25, 25.40,
 522 25.16, 22.22. ^{13}C NMR (151 MHz, $cdcl_3$) δ 152.24, 141.08, 134.50, 132.37, 131.62, 131.31, 130.91,
 523 129.31 (q, $J = 29.2$ Hz), 127.98, 127.19, 126.08 (q, $J = 5.2$ Hz), 124.21 (q, $J = 273.9$ Hz), 116.33,
 524 26.73, 22.88, 22.65, 19.70. HRMS (ESI) m/z : calcd. for $C_{17}H_{16}F_3 O^-$, $[M-H]^-$, 293.1159. Found
 525 293.1151.

526 5-Isopropyl-2-methyl-3'-(trifluoromethyl)-[1,1'-biphenyl]-4-ol (**Thy26**): 1H NMR (400 MHz,
 527 $CDCl_3$) δ 7.58 (dd, $J = 6.6, 1.8$ Hz, 2H), 7.54 - 7.48 (m, 2H), 7.03 (s, 1H), 6.69 (s, 1H), 4.86 (bs,
 528 1H), 3.26 - 3.16 (m, 1H), 2.19 (s, 3H), 1.27 (d, $J = 7.0$ Hz, 6H). ^{13}C NMR (151 MHz, $CDCl_3$) δ

155.05, 145.41, 136.45, 135.92, 135.47, 134.84, 133.10 (q, $J = 32.0$ Hz), 131.16, 130.65, 128.80 (q, $J = 3.5$ Hz), 126.95 (q, $J = 272.3$ Hz), 125.89 (q, $J = 3.5$ Hz), 119.86, 29.57, 25.36, 22.52. ^{13}C NMR (151 MHz, cdCl_3) δ 152.51, 142.87, 133.91, 133.39, 132.93, 132.30, 130.57 (q, $J = 32.0$ Hz), 128.62, 128.11, 126.28 (q, $J = 3.9$ Hz), 124.42 (q, $J = 272.3$ Hz), 123.36 (q, $J = 3.8$ Hz), 117.32, 27.04, 22.77, 20.00. HRMS (ESI) m/z : calcd. for $\text{C}_{17}\text{H}_{16}\text{F}_3 \text{O}^-$, $[\text{M}-\text{H}]^-$, 293.1159. Found 293.1154.

5-Isopropyl-2-methyl-4'-(trifluoromethyl)-[1,1'-biphenyl]-4-ol (**Thy27**): ^1H NMR (400 MHz, CDCl_3) δ 7.66 (d, $J = 6.0$ Hz, 2H), 7.43 (d, $J = 8.3$ Hz, 2H), 7.04 (d, $J = 3.3$ Hz, 1H), 6.69 (d, $J = 2.2$ Hz, 1H), 4.85 (bs, 1H), 3.26 – 3.15 (m, 1H), 2.20 (s, 3H), 1.28 – 1.25 (m, 6H). ^{13}C NMR (151 MHz, CDCl_3) δ 152.55, 145.84, 133.87, 133.43, 132.25, 129.90, 128.76 (q, $J = 32.2$ Hz), 128.08, 125.11 (q, $J = 3.8$ Hz), 124.54 (q, $J = 271.6$ Hz), 117.32, 27.00, 22.77, 20.03. HRMS (ESI) m/z : calcd. for $\text{C}_{17}\text{H}_{16}\text{F}_3 \text{O}^-$, $[\text{M}-\text{H}]^-$, 293.1159. Found 293.1154.

4.1.5 General procedure for the preparation of intermediates **2a-2i**

The solution of the respective bromoalkylacid (1 eq.) and triphenylphosphine (1 eq.) in acetonitrile (5 mL) was refluxed for 72 h. After the reaction completion, the solvent was removed and the crude product was purified by silica gel chromatography to afford intermediate **2a-2i**.

(2-Carboxyethyl)triphenylphosphonium bromide (**2a**): ^1H NMR (400 MHz, CDCl_3) δ 7.81 – 7.60 (m, 15H), 3.76 – 3.65 (m, 2H), 2.96 – 2.85 (m, 2H). ^{13}C NMR (101 MHz) δ 171.42, 171.28, 135.41, 135.39, 133.73, 133.63, 130.76, 130.64, 117.90, 117.04, 28.02, 19.15, 18.61.

(3-Carboxypropyl)triphenylphosphonium bromide (**2b**): ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ 7.92 – 7.67 (m, 15H), 3.74 – 3.68 (m, 2H), 2.52 (t, $J = 6.9$ Hz, 2H), 1.73 – 1.68 (m, 2H). ^{13}C NMR (101 MHz, $\text{DMSO}-d_6$) δ 173.32, 134.93, 133.62, 133.52, 130.31, 130.19, 118.78, 117.93, 33.72, 33.54, 20.12, 19.61, 17.78.

(4-Carboxybutyl)triphenylphosphonium bromide (**2c**): ^1H NMR (400 MHz, CDCl_3) δ 7.86 – 7.57 (m, 15H), 3.64 – 3.57 (m, 2H), 2.58 (t, $J = 6.8$ Hz, 2H), 1.94 – 1.87 (m, 2H), 1.74 – 1.65 (m, 2H). ^{13}C NMR (101 MHz, CDCl_3) δ 175.06, 135.28, 135.25, 133.77, 133.67, 130.76, 130.64, 118.52, 117.66, 33.87, 25.73, 25.56, 22.71, 22.20, 21.67.

(5-Carboxypentyl)triphenylphosphonium bromide (**2d**): ^1H NMR (400 MHz, CDCl_3) δ 7.96 – 7.49 (m, 15H), 3.55 – 3.48 (m, 2H), 2.26 (t, $J = 6.5$ Hz, 2H), 1.61 – 1.54 (m, 6H). ^{13}C NMR (101 MHz,

CDCl₃) δ 175.91, 135.19, 135.16, 133.66, 133.56, 130.68, 130.55, 118.52, 117.66, 34.17, 29.64, 29.48, 24.05, 22.76, 22.25, 22.01, 21.97.

(6-Carboxyhexyl)triphenylphosphonium bromide (**2e**): ¹H NMR (400 MHz, CDCl₃) δ 7.89 – 7.59 (m, 15H), 3.63 – 3.51 (m, 2H), 2.36 (t, J = 7.1 Hz, 2H), 1.69 – 1.57 (m, 4H), 1.54 – 1.50 (m, 2H), 1.36 – 1.29 (m, 2H). ¹³C NMR (101 MHz, CDCl₃) δ 176.44, 135.24, 133.78, 133.68, 130.75, 130.62, 118.70, 117.84, 34.47, 29.60, 29.45, 27.91, 24.27, 22.85, 22.35, 22.18.

(7-Carboxyheptyl)triphenylphosphonium bromide (**2f**): ¹H NMR (400 MHz, CDCl₃) δ 7.71 (m, 15H), 3.56 – 3.46 (m, 2H), 2.28 (t, J = 7.3 Hz, 2H), 1.52 (m, 6H), 1.34 – 1.11 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 176.70, 135.19, 135.17, 133.66, 133.56, 130.68, 130.56, 118.61, 117.75, 34.40, 30.02, 29.86, 28.25, 28.15, 24.52, 22.84, 22.34.

(8-Carboxyoctyl)triphenylphosphonium bromide (**2g**): ¹H NMR (400 MHz, CDCl₃) δ 7.91 – 7.45 (m, 15H), 3.55 – 3.48 (m, 2H), 2.24 (t, J = 7.4 Hz, 2H), 1.67 – 1.39 (m, 6H), 1.27 – 1.09 (m, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 176.92, 135.18, 135.16, 133.64, 133.54, 130.66, 130.54, 118.61, 117.76, 34.44, 30.18, 30.02, 28.51, 24.60, 22.89, 22.47, 22.42.

(9-Carboxynonyl)triphenylphosphonium bromide (**2h**): ¹H NMR (400 MHz, CDCl₃) δ 7.92 – 7.59 (m, 15H), 3.67 – 3.61 (m, 2H), 2.36 (t, J = 7.2 Hz, 2H), 1.66 – 1.43 (m, 6H), 1.27 – 1.17 (m, 8H). ¹³C NMR (101 MHz, CDCl₃) δ 177.12, 135.15, 133.59, 133.49, 130.63, 130.51, 118.54, 117.68, 34.38, 30.25, 30.09, 28.63, 24.61, 22.81, 22.39.

(10-Carboxydecyl)triphenylphosphonium bromide (**2i**): ¹H NMR (400 MHz, CDCl₃) δ 7.89 – 7.52 (m, 15H), 3.58 – 3.49 (m, 2H), 2.28 (t, J = 7.4 Hz, 2H), 1.62 – 1.45 (m, 6H), 1.24 – 1.06 (m, 10H). ¹³C NMR (101 MHz, CDCl₃) δ 177.33, 135.17, 135.14, 133.65, 133.55, 130.65, 130.53, 118.65, 117.80, 34.44, 30.39, 30.24, 28.99, 28.86, 28.77, 24.73, 22.93, 22.55, 22.50, 22.43.

4.1.6 General procedure for the preparation of **Thy3a-Thy3i**

To a solution of intermediates **2a-2i** (1 eq.) in DCM, added DCC and DMAP, and **Thy3**, then the mixture was stirred overnight at RT. After the reaction completion, the solvent was removed and the crude product was purified by silica gel chromatography to afford **Thy3a-Thy3i**.

(3-((5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-yl)oxy)-3-oxopropyl)triphenylphosphonium bromide (**Thy3a**): ¹H NMR (600 MHz, DMSO-*d*₆) δ 7.92 (t, J = 7.3 Hz, 3H), 7.88 – 7.85 (m, 6H), 7.81 – 7.77 (m, 6H), 7.43 (t, J = 7.5 Hz, 2H), 7.38 – 7.30 (m, 3H), 7.11 (s, 1H), 6.86 (s, 1H), 4.00 – 3.95

(m, 2H), 3.09 – 3.04 (m, 2H), 2.91 – 2.86 (m, 1H), 2.15 (s, 3H), 1.09 (d, $J = 6.9$ Hz, 6H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 169.63, 169.51, 146.60, 140.61, 139.52, 137.16, 135.12, 135.10, 133.84, 133.77, 133.38, 130.36, 130.28, 129.01, 128.26, 127.70, 127.02, 123.64, 118.29, 117.71, 26.75, 26.26, 22.93, 19.66, 16.98, 16.62. HRMS (ESI) m/z : calcd for $\text{C}_{37}\text{H}_{36}\text{O}_2\text{P}$, $[\text{M}-\text{Br}]^+$, 543.2447. Found 543.2465.

(4-((5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-yl)oxy)-4-oxobutyl)triphenylphosphonium bromide (**Thy3b**): ^1H NMR (400 MHz, DMSO- d_6) δ 7.91 – 7.74 (m, 15H), 7.41 (t, $J = 7.3$ Hz, 2H), 7.35 – 7.30 (m, 3H), 7.09 (s, 1H), 6.98 (s, 1H), 3.71 – 3.62 (m, 2H), 2.88 (t, $J = 7.2$ Hz, 2H), 2.14 (s, 3H), 1.92 – 1.83 (m, 2H), 1.07 (d, $J = 6.9$ Hz, 6H). ^{13}C NMR (101 MHz, DMSO- d_6) δ 170.95, 146.69, 140.71, 139.34, 137.03, 135.09, 133.68, 133.58, 133.36, 130.43, 130.31, 129.05, 128.30, 127.68, 127.04, 124.01, 118.71, 117.85, 33.68, 33.49, 26.61, 22.82, 20.07, 19.72, 19.56, 18.71, 17.79. HRMS (ESI) m/z : calcd for $\text{C}_{38}\text{H}_{38}\text{O}_2\text{P}$, $[\text{M}-\text{Br}]^+$, 557.2604. Found 557.2623.

(5-((5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-yl)oxy)-5-oxopentyl)triphenylphosphonium bromide (**Thy3c**): ^1H NMR (600 MHz, DMSO- d_6) δ 7.89 (t, $J = 7.4$ Hz, 3H), 7.84 – 7.73 (m, 12H), 7.43 (t, $J = 7.5$ Hz, 2H), 7.36 – 7.32 (m, 3H), 7.10 (s, 1H), 6.85 (s, 1H), 3.69 – 3.64 (m, 2H), 2.88 – 2.83 (m, 1H), 2.72 (t, $J = 7.4$ Hz, 2H), 2.15 (s, 3H), 1.88 – 1.83 (m, 2H), 1.70 – 1.63 (m, 2H), 1.07 (d, $J = 6.9$ Hz, 6H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 171.73, 146.75, 140.71, 139.28, 137.14, 134.95, 133.62, 133.55, 133.37, 130.32, 130.24, 129.02, 128.26, 127.66, 126.99, 123.85, 118.77, 118.20, 32.49, 26.58, 25.26, 25.14, 22.81, 21.25, 21.23, 20.20, 19.86, 19.67. HRMS (ESI) m/z : calcd for $\text{C}_{39}\text{H}_{40}\text{O}_2\text{P}$, $[\text{M}-\text{Br}]^+$, 571.2760. Found 571.2771.

(6-((5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-yl)oxy)-6-oxohexyl)triphenylphosphonium bromide (**Thy3d**): ^1H NMR (600 MHz, DMSO- d_6) δ 7.89 (t, $J = 7.3$ Hz, 3H), 7.83 – 7.73 (m, 12H), 7.43 (t, $J = 7.5$ Hz, 2H), 7.36 – 7.32 (m, 3H), 7.11 (s, 1H), 6.90 (s, 1H), 3.62 – 3.57 (m, 2H), 2.91 – 2.87 (m, 1H), 2.59 (t, $J = 7.3$ Hz, 2H), 2.15 (s, 3H), 1.71 – 1.66 (m, 4H), 1.62 – 1.54 (m, 4H), 1.11 (d, $J = 6.9$ Hz, 6H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 171.88, 146.82, 140.74, 139.24, 137.17, 134.91, 133.65, 133.59, 133.37, 130.29, 130.21, 129.03, 128.27, 127.65, 126.99, 123.92, 118.85, 118.28, 33.09, 29.32, 29.20, 26.62, 23.64, 22.83, 21.59, 21.56, 20.38, 20.04, 19.68. HRMS (ESI) m/z : calcd for $\text{C}_{40}\text{H}_{42}\text{O}_2\text{P}$, $[\text{M}-\text{Br}]^+$, 585.2917. Found 585.2930.

(7-((5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-yl)oxy)-7-oxoheptyl)triphenylphosphonium bromide (**Thy3e**): ^1H NMR (600 MHz, DMSO- d_6) δ 7.88 (t, $J = 7.3$ Hz, 3H), 7.82 – 7.73 (m, 12H), 7.43 (t,

616 $J = 7.5$ Hz, 2H), 7.36 – 7.32 (m, 3H), 7.11 (s, 1H), 6.91 (s, 1H), 3.61 – 3.56 (m, 2H), 2.92 – 2.85 (m,
 617 1H), 2.58 (t, $J = 7.4$ Hz, 2H), 2.15 (s, 3H), 1.63 – 1.50 (m, 6H), 1.36 – 1.41 (m, 2H), 1.10 (d, $J =$
 618 6.9 Hz, 6H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 171.96, 146.83, 140.74, 139.22, 137.12, 134.89,
 619 133.62, 133.55, 133.36, 130.28, 130.20, 129.02, 128.25, 127.62, 126.97, 123.95, 118.85, 118.28,
 620 33.33, 29.53, 29.42, 27.58, 26.61, 24.81, 24.13, 22.79, 21.68, 21.65, 20.37, 20.04, 19.65. HRMS
 621 (ESI) m/z : calcd for $\text{C}_{41}\text{H}_{44}\text{O}_2\text{P}$, $[\text{M}-\text{Br}]^+$, 599.3073. Found 599.3092.
 622 (8-((5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-yl)oxy)-8-oxooctyl)triphenylphosphonium bromide
 623 (**Thy3f**): ^1H NMR (600 MHz, DMSO- d_6) δ 7.88 (t, $J = 7.2$ Hz, 3H), 7.82 – 7.72 (m, 12H), 7.43 (t,
 624 $J = 7.5$ Hz, 2H), 7.39 – 7.30 (m, 3H), 7.11 (s, 1H), 6.91 (s, 1H), 3.59 – 3.53 (m, 2H), 2.92 – 2.88 (m,
 625 1H), 2.59 (t, $J = 7.3$ Hz, 2H), 2.15 (s, 3H), 1.65 – 1.60 (m, 2H), 1.56 – 1.50 (m, 2H), 1.49 – 1.44 (m,
 626 2H), 1.35 – 1.28 (m, 4H), 1.11 (d, $J = 6.9$ Hz, 6H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 172.06,
 627 146.85, 140.75, 139.23, 137.16, 134.89, 133.61, 133.55, 133.38, 130.29, 130.21, 129.02, 128.27,
 628 127.64, 126.98, 123.94, 118.88, 118.31, 33.36, 29.76, 29.65, 28.14, 27.81, 26.61, 24.31, 22.82,
 629 21.78, 21.75, 20.35, 20.02, 19.66. HRMS (ESI) m/z : calcd for $\text{C}_{42}\text{H}_{46}\text{O}_2\text{P}$, $[\text{M}-\text{Br}]^+$, 613.3230. Found
 630 613.3246.
 631 (9-((5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-yl)oxy)-9-oxononyl)triphenylphosphonium bromide
 632 (**Thy3g**): ^1H NMR (600 MHz, DMSO- d_6) δ 7.88 (t, $J = 7.3$ Hz, 3H), 7.81 – 7.73 (m, 12H), 7.43 (t,
 633 $J = 7.5$ Hz, 2H), 7.38 – 7.30 (m, 3H), 7.11 (s, 1H), 6.92 (s, 1H), 3.59 – 3.52 (m, 2H), 2.93 – 2.89 (m,
 634 1H), 2.58 (t, $J = 7.3$ Hz, 2H), 2.15 (s, 3H), 1.65 – 1.60 (m, 2H), 1.55 – 1.49 (m, 2H), 1.48 – 1.43 (m,
 635 2H), 1.35 – 1.20 (m, 6H), 1.11 (d, $J = 7.0$ Hz, 6H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 172.02, 146.84,
 636 140.74, 139.20, 137.13, 134.85, 133.60, 133.54, 133.36, 130.26, 130.18, 129.00, 128.24, 127.61,
 637 126.95, 123.93, 118.87, 118.30, 33.43, 29.82, 29.71, 28.36, 28.29, 27.99, 26.60, 24.37, 22.80, 21.77,
 638 21.74, 20.35, 20.02, 19.64. HRMS (ESI) m/z : calcd for $\text{C}_{43}\text{H}_{48}\text{O}_2\text{P}$, $[\text{M}-\text{Br}]^+$, 627.3386. Found
 639 627.3398.
 640 (10-((5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-yl)oxy)-10-oxodecyl)triphenylphosphonium bromide
 641 (**Thy3h**): ^1H NMR (400 MHz, CDCl_3) δ 7.76 – 7.57 (m, 15H), 7.32 (t, $J = 7.3$ Hz, 2H), 7.27 – 7.19
 642 (m, 3H), 7.06 (s, 1H), 6.80 (s, 1H), 3.52 – 3.46 (m, 2H), 2.94 – 2.87 (m, 1H), 2.50 (t, $J = 7.5$ Hz,
 643 2H), 2.12 (s, 3H), 1.62 – 1.70 (m, 2H), 1.56 – 1.50 (m, 4H), 1.32 – 1.28 (m, 2H), 1.27 – 1.17 (m,
 644 6H), 1.10 (d, $J = 7.0$ Hz, 6H). ^{13}C NMR (101 MHz, CDCl_3) δ 172.50, 146.86, 141.32, 139.59, 137.11,
 645 134.95, 134.93, 133.65, 133.37, 133.27, 130.43, 130.31, 129.04, 127.88, 126.63, 123.49, 118.42,

117.56, 34.14, 33.58, 30.32, 30.16, 28.91, 28.83, 26.94, 24.99, 24.76, 22.87, 22.42, 22.38, 21.93, 19.88. HRMS (ESI) m/z : calcd for $C_{44}H_{50}O_2P$, $[M-Br]^+$, 641.35. Found 641.36.

(11-((5-Isopropyl-2-methyl-[1,1'-biphenyl]-4-yl)oxy)-11-oxoundecyl)triphenylphosphonium bromide (**Thy3i**): 1H NMR (600 MHz, DMSO- d_6) δ 7.87 (t, $J = 7.2$ Hz, 3H), 7.82 – 7.71 (m, 12H), 7.43 (t, $J = 7.5$ Hz, 2H), 7.36 – 7.32 (m, 3H), 7.11 (s, 1H), 6.92 (s, 1H), 3.56 – 3.52 (m, 2H), 2.94 – 2.89 (m, 1H), 2.59 (t, $J = 7.3$ Hz, 2H), 2.15 (s, 3H), 1.66 – 1.61 (m, 2H), 1.54 – 1.48 (m, 2H), 1.46 – 1.41 (m, 2H), 1.36 – 1.31 (m, 2H), 1.28 – 1.25 (m, 4H), 1.20 – 1.18 (m, 4H), 1.12 (d, $J = 6.8$ Hz, 6H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 172.06, 146.85, 140.75, 139.20, 137.14, 134.87, 134.85, 133.60, 133.53, 133.37, 130.26, 130.18, 129.01, 128.25, 127.62, 126.96, 123.93, 118.86, 118.30, 33.44, 29.87, 29.77, 28.74, 28.64, 28.60, 28.40, 28.11, 26.59, 24.43, 22.81, 21.77, 21.74, 20.34, 20.01, 19.64. HRMS (ESI) m/z : calcd for $C_{45}H_{52}O_2P$, $[M-Br]^+$, 655.3699. Found 655.3709.

4.2 Antimicrobial activity

The minimum inhibitory concentration (MIC) of the compounds was tested by the Clinical and Laboratory Standards Institute (CLSI) broth dilution procedure with minor modifications. Two Gram-positive bacteria (*S. aureus* and MRSA) and four Gram-negative bacteria (*E. coli*, *A. baumannii*, *P. aeruginosa*, and *K. pneumoniae*) were selected as the tested bacteria. Each bacterial suspension was adjusted to a concentration of 1×10^6 CFU/mL. All compounds were thoroughly dried before weighing. Initially, the compounds were dissolved in dimethyl sulfoxide (DMSO) to prepare the stock solutions. The tested compounds and ciprofloxacin were then prepared in liquid Luria–Bertan media. The concentration of Thymol was 512, 256, 128, 64, 32, 16, 8, 4, 2, 1 and 0.5 μ g/mL. Their MICs of the preferred compounds **Thy1-27**, **Thy3a-3i**, and ciprofloxacin were also tested at concentrations of 64, 32, 16, 8, 4, 2, 1, 0.5, 0.25, 0.125, and 0.0625 μ g/mL. After incubation at 37 °C for 18 h, the inhibition of bacterial growth in each well was determined by measuring absorbance at 595 nm with a microplate reader. The MIC of the compounds was defined as the lowest concentration that completely inhibited bacterial growth.

4.3 Cell toxicity

To determine the cytotoxicity of the compounds, we used the MTT dye reduction assay against the L02 cell. In brief, 1×10^5 cells in a 96-well plate were and cultured overnight in DMEM and at

37 °C with 5% CO₂. The cells were washed twice with phosphate-buffered saline (PBS) and fresh media was added. The compounds were serially diluted by halving concentrations in media and the cells were incubated for 72 h. Then, MTT solution (5 mg/mL) was added to the cells for 4 h. After incubation, the precipitated MTT formazan crystal was dissolved by DMSO. Absorbance at 595 nm was measured using a microplate reader.

4.4 Time kill kinetics

The *in vitro* time-kill kinetics of **Thy3d** and ciprofloxacin were performed against MRSA 3390. The bacteria were incubated on LB at 37 °C for 8 h, and then the inoculum was adjusted to approximately 10⁶ CFU/mL. The bacterial suspension was incubated with samples at different concentrations (1, 2, 4, and 8 × MIC) at 37 °C. The aliquots were then removed from the mixtures at different time points (0, 1, 4, 6, and 24 h), and were serially diluted 10-fold in PBS (pH = 7.2) and plated with TSA medium. The plates were incubated at 37 °C for 24 h. The bactericidal efficacy of the compounds was evaluated by counting the bacterial colonies grown on the plates.

4.5 Drug resistance study

MRSA 3390 was employed to assess the development of resistance by the serial passaging method, as previously described[36]. The initial MICs for the compounds against MRSA 3390 were obtained according to the above method. Bacteria harvested from 96-well plates of 1/2 MIC levels, diluted in MHB to a concentration of 1 × 10⁶ CFU/mL and then subjected to MIC determination. The experiment was repeated every day for 40 days.

4.6 MRSA persister cells eradicating assay

As has been previously demonstrated, stationary-phase cells of *S. aureus* can be used to model persister cells[37, 38]. Briefly, a single clone of MRSA 3390 was inoculated into a TSB medium and cultivated until the stationary period. Then the culture was treated with ciprofloxacin (final concentration 40 µg/mL, 20 × MIC) for 6 h. The bacteria cells were collected by centrifugation (8000 rpm, 5 min, 4 °C) and washed with sterile PBS. Re-suspend the bacteria in sterile PBS to a density of OD₆₀₀ = 0.5. The bacteria re-suspension was treated with different concentrations of **Thy3d** (1, 2, 4, and 8 × MIC). Subsequently, 10 µL of the bacterial solution was taken at set time points (0, 1, 3, 5, and 7 h) and applied to the LB agar plate after series dilution. The number of

colonies was counted after 24 hours of culture at 37 °C. Bacteria re-suspension containing 1% DMSO was used as a negative control. Make three duplicate wells for each concentration of the drug, two plates in parallel for each dilution.

4.8 DNA leakage

The membrane permeability was performed by estimating the amounts of nucleic acids released as previously described with brief modification[39]. A single colony of MRSA 3390 was cultured in 20 mL TSB overnight at 37 °C. The culture was centrifuged at 8000 rpm for 5 min, then the supernatant was discarded and the bacterial pellet was washed with PBS. The bacteria were then re-suspended in PBS. The bacterial suspension was treated with different concentrations (1, 2, 4, and 8 × MIC) of **Thy3d**. The treatment of bacteria re-suspension containing 1% DMSO was used as a negative control. 1 mL sample was taken from cultures at different points (0, 1, 2, 4, and 6 h) and immediately sieved with 0.22 µm syringe filters to remove the bacterial cells. The released amount of DNA in the filtrate was measured at 260 nm with a micro-spectrophotometer.

The persisters were obtained according to method 4.6, and the DNA leakage of the persister was tested as described above.

4.9 SYTOX Green assay

MRSA 3390 was incubated at 37 °C overnight. The culture was centrifuged at 8000 rpm for 5 min, then the supernatant was discarded and the bacterial pellet was washed with PBS to prepare bacterial suspension (2×10^7 CFU/mL). SYTOX™ Green was added to the bacterial suspension with a final concentration of 5 µM and kept in the dark for 20 min[40]. After the fluorescence signal had stabilized, the different concentrations (1, 2, 4, and 8 × MIC) of compound **Thy3d** (final DMSO concentration of 0.5%) were added to the mixture, and the fluorescence intensity changes were monitored and recorded every 5 min, for a total of 120 min. The treatment of bacteria re-suspension containing 0.5% DMSO was used as a negative control, and the treatment of 1% triton X-100 was used as a positive control. The effects of the **Thy3d** on the membrane permeability of normal bacteria were detected.

The persisters were obtained according to method 4.6, and the membrane permeability of the persister was tested as described above.

4.10 Membrane depolarization assay

MRSA 3390 was incubated in LB at 37 °C overnight. The culture was centrifuged at 8000 rpm for 5 min, then the supernatant was discarded and the bacterial pellet was washed 3 times with assay buffer (5 mM HEPES, 20 mM glucose, 100 mM KCl, pH 7.4), and finally resuspended in assay buffer, adjusting the OD₆₀₀ to 0.1. DiSC₃(5) (3,3'-dipropylthiacarbocyanine) dye was added to the cells at a final concentration of 3 µM[41]. The above bacteria suspension with DiSC₃(5) was added to each well of a 96-well black microplate and incubated at 37 °C until a stable fluorescence was observed. **Thy3d** was added into the wells at different concentrations (1, 2, 4, and 8 × MIC). Plates were immediately read on an Envision at $\lambda_{\text{ex}} = 622 \text{ nm}$ / $\lambda_{\text{em}} = 670 \text{ nm}$ (reading every 2 min) until the maximal intensity was achieved. The effects of the **Thy3d** on the membrane depolarization of normal bacteria were detected.

The persisters were obtained according to method 4.6, and the membrane depolarization of the persister was tested as described above.

4.11 SEM assay

A standard procedure was followed for SEM analysis[42]. MRSA 3390 was incubated to the logarithmic growth phase, then suspended in PBS buffer (pH 7.2) and adjusted to approximately 5×10^7 CFU/mL. The bacterial suspension was mixed with 4 × MIC of **Thy3d** and incubated at 37 °C for 1 h. The PBS-treated group was used as a negative control. All samples were centrifuged at 5000 g for 5 min, washed, and resuspended in PBS. Bacterial cells were fixed with 2.5% glutaraldehyde and then incubated at 37 °C on a glass slide for 20 min. Samples were dehydrated by passing through gradient ethanol (10%, 30%, 50%, 70%, 90%, and 100%) for 5 min at each concentration. Slides were then dried with an automatic critical point drying instrument (Leica EM CPD300), and a small amount of gold was sputtered onto the samples using a sputter coater system. Morphology images were collected using a scanning electron microscope (Regulus 8100, Hitachi, Japan).

4.12 Molecular dynamics simulations

Atomistic molecular dynamics (MD) simulations were carried out to study the mode of interactions of **Thy3d** with a model bacterial membrane constructed out of 96 zwitterionic lipids POPE and 32 anionic lipids POPG and with the mammalian membrane was modelled using 96

zwitterionic lipids POPC and 32 cholesterol[43]. These models can capture the main structural features of bacterial membrane and has been widely used in previous studies[44-47].

To understand the membrane affinity of **Thy3d**, the free energy of adsorption of **Thy3d** on bacterial and mammalian membranes was carried out using umbrella sampling[48]. The distance between the center of mass of the **Thy3d** and the bilayer center was chosen as the reaction coordinate. In each umbrella sampling, a total of 14 window simulations spanning the distance from 0 nm to 4 nm were carried out. Each window simulation was run for 400 ns, the free energy profile was constructed using the weighted histogram analysis method (WHAM)[49].

CHARMM36m force field was used to model both the **Thy3d** and the membranes, and water molecules were modeling using CHARMM modified TIP3P model[50]. The topology of the peptide and lipids was generated using CHARMM-GUI[51]. In each simulation, the LJ interactions were calculated using a cutoff of 1.2 nm, with a force switch between 1.0 nm to 1.2 nm. The short-range electrostatic interactions were calculated with a cutoff of 1.2 nm while the long-range electrostatic interactions were computed using PME[52]. During the MD simulations, the covalent bonds involving hydrogen atoms were constrained using the LINCS algorithm, enabling a time step of 2 fs to be used. The simulations were run in an NPT ensemble, with temperature maintained at 300 K using the Nose-Hoover method and pressure maintained using the Parrinello-Rahman method with semi-isotropic pressure coupling. All simulations were carried out using GROMACS 2021.

4.13 Bilayer experiment and electrical recording.

The bilayer experiments were performed according to the method described by Zhou with minor modifications[31]. Briefly, this experiment was performed in a planar bilayer chamber, which was separated into two compartments by a Teflon film, and the lipid bilayer was formed at a 156- μ m-diameter aperture of the Teflon film using DPhPC according to the Montal–Mueller method[53]. All the experiments were carried out in the chamber with 2 mL of buffer solutions added in each compartment containing 1 M KCl, 10 mM Tris, and 1 mM EDTA (pH 7.6). The higher ion concentration than the one (0.15 M) used in the simulation can enhance the ionic current through the pore[54] achieving improved signal-to-noise ratios and meanwhile can screen charges of membrane/**Thy3d** at pH 7.6. **Thy3d** were added to the grounded cis compartment. An Axopatch 200B amplifier (Molecular Devices, Sunnyvale, CA) was used to record the ionic currents that were

low-pass filtered with a built-in four-pole Bessel filter at 2 kHz and sampled at 50 kHz by a computer equipped with a Digidata 1440A converter (Molecular Devices).

4.14 Mouse skin infection model

Six-week-old adult female Balb/C mice were randomly divided into three groups with six mice in each group, and rendered neutropenic by injecting two doses of cyclophosphamide intraperitoneally 4 days (150 mg/kg) and 1 day (100 mg/kg) before experimental infection. *S. aureus* Xen29 bioluminescent bacteria were cultured under standard conditions[55]. On Day -1, the fur on the right leg of each of the mice was removed by shaving and application of chemical depilatories. On Day 0, approximately 1×10^9 CFU of bacteria in 100 μ L were inoculated by subcutaneous injection in the right thigh of each mouse. A saline solution containing 5% DMSO was used to dissolve the drug. The success of modeling was observed by bioluminescent *in vivo* imaging 12 h after infection, and uninfected mice were excluded. **Thy3d** (8 mg/kg, 100 μ L), vancomycin (8 mg/kg, 100 μ L), or salina (100 μ L) was injected into the mice by subcutaneous injection 16, 24, 36, 48, 60, and 72 h post-infection. The skin wound of the mice was monitored daily, and the weight of the mice was recorded every day.

Bioluminescent *in vivo* imaging was performed at 12, 24, 36, 60, and 84 h post-infection. On Day 4, 84 hours after the infection, animals were sacrificed. Samples of skin from each animal were homogenized and plated on nutrient agar to determine the CFUs per animal. Animals were imaged using a Lumina II system (Perkin-Elmer), and bioluminescence (photons per second) was determined.

4.15 Statistical analysis

The statistical tests and several biological replicates and/or experiments were stated in the legend of the figure, with significance set at $p < 0.05$. All statistical data were analyzed by GraphPad Prism 8.0 statistical software.

Declaration of competing interest

Authors declare no competing interests.

Data availability

Data will be made available on request.

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